DOI: 10.1002/chem.200801478

Total Synthesis of Discodermolide: Optimization of the Effective Synthetic Route

Elsa de Lemos,^[a] François-Hugues Porée,^[a] Arnaud Bourin,^[a] Julien Barbion,^[a] Evangelos Agouridas,^[a] Marie-Isabelle Lannou,^[a] Alain Commerçon,^[b] Jean-François Betzer,^{*[a]} Ange Pancrazi,^{*[a]} and Janick Ardisson^{*[a]}

Abstract: An efficient and modulable total synthesis of discodermolide (DDM), a unique marine anticancer polyketide is described including related alternative synthetic approaches. Particularly notable is the repeated application of a crotyltitanation reaction to yield homoallylic (Z)-O-ene-carbamate alcohols with excellent selectivity. Advantage was taken of this reaction not only for the stereocontrolled building of the *syn-anti* methyl-hydroxy–

Introduction

In 1990, Gunasekera and co-workers reported the isolation of discodermolide (DDM) (1),^[1] a unique polyketide marine metabolite obtained in low yields from the Caribbean deepwater sponge *Discodermia dissoluta*. Biological studies revealed DDM to be a potent microtubule-stabilizing agent

[a] Dr. E. de Lemos, Dr. F.-H. Porée, Dr. A. Bourin, Dr. J. Barbion, Dr. E. Agouridas, Dr. M.-I. Lannou, Dr. J.-F. Betzer, Dr. A. Pancrazi, Prof. J. Ardisson Université Paris Descartes Faculté de Pharmacie, CNRS UMR 8638 4 avenue de l'Observatoire 75270 Paris Cedex (France) Fax: (+33)143291403 E-mail: jean.francois.betzer@icsn.cnrs-gif.fr ange.pancrazi@parisdescartes.fr janick.ardisson@parisdescartes.fr [b] Dr. A. Commerçon Chemical&Analytical Sciences Natural Product Chemistry, Sanofi-Aventis Centre de Recherche de Vitry-Alfortville 13 Quai Jules Guesde

94403 Vitry-sur-Seine Cedex (France)

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/chem.200801478.

methyl triads of DDM, but also for the direct construction of the terminal (Z)-diene. Of particular interest is also the installation of the C13=C14 (Z)-double bond through a highly selective dyo-tropic rearrangement. The preparation of the middle C8-C14 fragment in two

Keywords:allylationcross-couplingdiscodermolidenickel • total synthesis

sequential stages and its coupling to the C1–C7 moiety was a real challenge and required careful optimization. Several synthetic routes were explored to allow high and reliable yields. Due to the flexibility and robust character of this approach, it might enable a systematic structural variation of DDM and, therefore, the elaboration and exploration of novel discodermolide structural analogues.

that, like Taxol (Paclitaxel), arrests cells at the G2/M boundary of the cell cycle.^[2] The cytotoxicity values reported for discodermolide against breast, prostate, colon, lung, and ovarian cancer cell lines are generally in the low nm range. Comparative studies showed that DDM was 1000-fold more potent than Taxol in promoting the same microtubule polymerization/bundling. This product is also more water-soluble than taxoid compounds, and acts synergistically in combination with paclitaxel.^[3] Extensive investigation of the site and mode of binding of DDM to β-tubulin as well as the determination of its bioactive conformation led to the proposal of a pharmacophore model.^[2d] Due to the potential therapeutic applications and the extreme scarcity of this compound, considerable synthetic efforts directed towards DDM were provided, culminating in several total syntheses,^[4] including the development of a preparative-scale approach.^[5] Phase I clinical trials were initiated from Novartis Pharma AG; however, despite encouraging results, trials were suspended due to adverse toxicity at high doses.^[6] Owing to the remarkable profile of DDM and its potential as a lead structure, conception and synthesis of analogues have to be developed.

In this full account, we report synthetic details of our total synthesis of DDM (1), including related alternative synthetic approaches that enabled us to develop the final effective synthetic route.^[7] The modulable character of this ap-



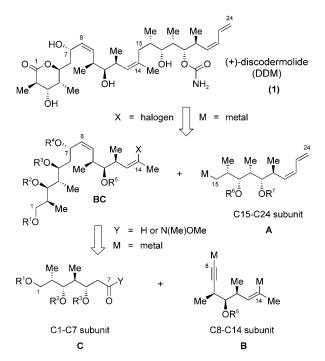


© 2008 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

proach might allow structural modifications of DDM and, therefore, the elaboration and investigation of novel discodermolide structural analogues.

Results and Discussion

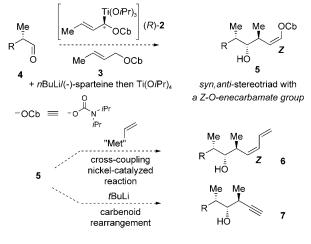
Retrosynthetic and strategic considerations: Our synthetic strategy relies upon the disassembly of DDM into three fragments of similar size and stereochemical complexity (**A** C15–C24, **B** C8–C14, **C** C1–C7; Scheme 1). The linkage of these subunits appears to us to be optimally convergent, the two internal (*Z*)-alkenes C8=C9 and C13=C14 being pivotal in the choice of disconnection strategies either in acetylide addition in the aldehyde/reduction sequence or Pd-catalyzed sp²–sp³ coupling.



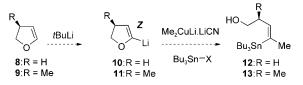
Scheme 1. Retrosynthetic analysis of discodermolide.

A central feature of our approach was the repeated matched addition of the chiral secondary (*R*)-crotyltitanium reagent **2** to α -(*S*)-methyl aldehyde **4**. As exemplified in Scheme 2, this crotyltitanation reaction initially developed by Hoppe,^[8] and widely used in total syntheses by our group,^[9] yielded homoallylic adducts **5** encompassing a *syn*-*anti* methyl-hydroxy-methyl triad linked to a (*Z*)-*O*-enecarbamate group, with excellent diastereoselectivity. The enantioenriched (*R*)- α -(*N*,*N*-diisopropylcarbamoyloxy)crotyltitanium **2** could easily be prepared in situ from deprotonation of crotyl diisopropylcarbamate **3** with *n*BuLi/(–)-sparteine and tetra(isopropoxy)titanium.^[10]

Our plan was precisely to take advantage of the reactivity of this terminal (Z)-O-enecarbamate function in two ways. First, a nickel-catalyzed cross-coupling reaction would con-



Metallate rearrangement



Scheme 2. Crotyltitanation of aldehyde 4. Insaturation setup from (Z)-acyclic or cyclic enol ether.

figure the terminal C21–C24 (*Z*)-diene **6** of fragment \mathbf{A} ,^[11] whereas a carbenoid rearrangement would install the terminal C8–C9 alkyne **7** of subunit **B** in one step.^[12]

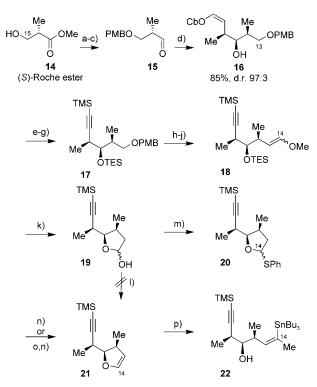
Another highlight of our strategy was the stereocontrolled building of the C13=C14 trisubstituted (Z)-double bond. With our C14-C15 disconnection, elected by several research groups, the elaboration of the pivotal segment **B**, which encompassed a (Z)-vinyl halide at the C14-position, was needed for the sp²-sp³ cross-coupling reaction between **A** and **B**. A (Z)-vinyl iodide can be obtained in one step, by Zhao aldehyde olefination,^[13] but the yields were modest and the Z/E stereoselectivity remained disappointing.^[14] To circumvent this reaction, we considered focusing on a dyotropic rearrangement of the 5-lithiodihydrofurans 10 and 11, obtained by metalation of cyclic enol ethers 8 and 9 leading to (Z)-alkenes 12 and 13. This attractive reaction, first examined by Fujisawa^[15] and developed by Kocienski^[16] and our group,^[17] would allow the construction of various (Z) or (E)-trisubstituted double bonds, in expected good yields and with total control of their geometry.

This retrosynthetic plan should ultimately allow for modifications of every substructure embedded into the frame of the natural compounds.

Preparation of the building blocks: The recent determination of the DDM bioactive "U"-shaped conformation strongly supports the crucial role of the middle part of the DDM molecule.^[2b] The preparation of this C8–C14 fragment region **B** was a real challenge; the strategy involved two main stages, a crotyltitanation for the setup of the C10–C12 *anti–syn* stereotriad and a dyotropic rearrangement to build the C13=C14 trisubstituted (*Z*)-double bond. Two alterna-

tive routes were examined, which differed in the ordering of the two key steps.

The first sequence started with the commercially available (S)-(+)-Roche ester **14** (Scheme 3). Protection of **14** as a *p*-methoxybenzyl ether^[18] was followed by reduction of the



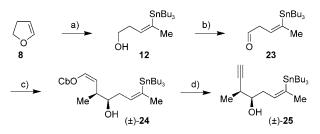
Scheme 3. a) CCl₃C(N)OPMB, PPTS (5 mol %), CH₂Cl₂/cyclohexane 1:2, RT, 40 h, 88%; b) LAH, THF, 0°C→RT, 3 h, 95%; c) TEMPO (2 mol%), NaOCl, KBr, NaHCO₃, CH₂Cl₂/H₂O; d) (R)-2, cyclohexane/ pentane 1:7, -78°C, 2 h, 85% (2 steps, c-d), d.r. 97:3; e) TESOTf, 2,6-lutidine, CH_2Cl_2 , 0°C \rightarrow RT, 2 h, 97%; f) *t*BuLi, Et₂O, -30°C \rightarrow -20°C, 45 min, 84 %; g) *n*BuLi, THF, $-78 \,^{\circ}\text{C} \rightarrow 0 \,^{\circ}\text{C}$, 1 h, then TMSCl, $-78 \,^{\circ}\text{C} \rightarrow$ RT, overnight, 92 %; h) DDQ, CH₂Cl₂/H₂O 95:5, 0°C → RT, 30 min, 86 %; i) IBX, DMSO, RT, 3 h; j) (methoxymethyl)triphenylphosphonium chloride, LiHMDS, -10°C, 1 h, 85%; k) acetone/H2O 9:1, HCl, 65°C, 1 h, 61% (3 steps, h-k); l) PTSA, PPTS, CuSO₄, NH₄NO₃ or Burgess reagent; m) PhSH, BF₃·OEt₂, 4 Å MS, Et₂O, 1 h, 80%; n) DBU, neat, 245°C, 30%; o) mCPBA, NaHCO₃, PhH, 0°C, 1 h, then, reflux, 1 h, 19%; p) tBuLi, Me₂CuLi·LiCN, THF/Et₂O/1:2, 0°C \rightarrow RT, 3 h, then nBu₃SnCl, -30 °C \rightarrow RT, 5 h, 20%. mCPBA = 3-chloroperbenzoic acid; DBU=1,8diazabicyclo[5.4.0]undec-7-ene; DDQ = 2,3-dichloro-5,6-dicyano-1,4-benzoquinone; d.r.=diasteromeric ratio; HMDS=hexamethyldisilazane; IBX=2-iodoxybenzoic acid; MS=molecular sieves; PMB=p-methoxybenzyl; PPTS = pyridinium p-toluenesulfonate; PTSA = p-toluenesulfonic acid; TEMPO=tetramethylpiperdinyloxy free radical; TES=triethylsilyl; TMS = trimethylsilyl.

methyl ester to the primary alcohol and subsequent oxidation to aldehyde 15.^[19] The crotyltitanation reaction of chiral aldehyde 15, with enantioenriched reagent (*R*)-2 led to the desired diastereomer 16 in good yield and with an excellent diastereomeric ratio (97:3) (Scheme 3). Obtention of a *O*enecarbamate moiety by the crotyltitanation process allowed for the generation of alkyne 17 by a Fritsch-Buttenberg–Wiechell rearrangement.^[12] Enol ether **18** was readily available from **17**, after PMB ether cleavage with DDQ, IBX oxidation of the resulting alcohol, and subsequent aldehyde homologation with the commercially available (methoxymethyl)triphenylphosphonium ylide. Aqueous acidic hydrolysis of the resulting enol ether **18**, accompanied by removal of the TES group afforded lactol **19**.

The next step should allow us to build the C13=C14 trisubstituted functionalized (Z)-double bond by means of a dyotropic rearrangement. However, the envisaged transformation of lactol 19 into the 2,3-dihydrofuran (DHF) derivative 21, turned out to be more difficult than expected. Specifically, all attempts to effect a direct elimination (PTSA,^[20] PPTS,^[21] CuSO₄,^[22] NH₄NO₃,^[23] or Burgess reagent)^[24] from compound 19 failed to give the expected DHF 21. However, lactol 19 could be smoothly converted to hemithioketal 20. Upon DBU pyrolysis of 20 with concomitant distillation of the elimination product,^[25] the desired dihydrofuran **21** was obtained in 30% yield. Attempts to improve this outcome by conversion of phenyl sulfide 20 into the corresponding sulfoxide derivative^[26] and subsequent elimination delivered DHF 21 in only 19% yield. Application to 21 of dyotropic rearrangement conditions (1,2-cuprate transfer from cyano-Gilman dimethylcuprate followed by tri-n-butyltin trapping) led to the expected C8-C14 segment 22 in 20% yield. Unfortunately, this 15 step sequence afforded the C8-C14 subunit 22 in a very poor yield and thus could not provide sufficient quantities to carry on the synthesis.

The formation of the DHF ring with an alkyne part was a critical step in the synthesis. The strategy was reconsidered and an alternative tactic was suggested based on an earlier formation of the DHF ring. By using the same approach, we inverted the order of the crotyltitanation–dyotropic rearrangement sequence. This required the introduction of a C13=C14-functionalized double bond bearing a metallic or halide function at a very early stage of synthesis. According to the literature, vinyl iodide hydrogenolysis occurred in the subsequent Lindlar reduction step,^[41] and, therefore, we choose to introduce a vinyl tin function as the precursor of the vinyl halide core at the C14 position.

Stability of the vinyl stannane function during C8-C14 subunit elaboration was a challenging problem. With the aim to determine the chemical compatibility of this moiety, we decided to prepare a C12-desmethyl model (\pm) -25 (Scheme 4). While the formation of alkene 12 in one step by metallate rearrangement with a cyano-Gilman dimethylcuprate from commercially available 2,3-dihydrofuran 8 proceeded smoothly,^[27] significant problems were encountered during conversion of homoallylic alcohol 12 into aldehyde 23. Extensive screening was performed. Classical oxidation methods, such as Swern,^[28] Doering-Parikh,^[29] or the use of PCC^[30] led to decomposition with loss of the tin function. On the other hand, milder reagents, such as tetrapropylammonium perruthenate (TPAP)/N-methylmorpholine N-oxide (NMO)^[31] or IBX^[32] resulted in recovery of the starting material. The Saigo-Mukaiyama protocol,^[33] employing 1,1'-(azodicarbonyl)dipiperidine (ADD) as a hydride acceptor

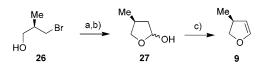


Scheme 4. a) *t*BuLi, Me₂CuLi-LiCN, THF/Et₂O/1:2, 0°C \rightarrow RT, 3 h, then *n*Bu₃SnCl, -30°C \rightarrow RT, 5 h, 61%; b) TEMPO (10 mol%), BAIB, CH₂Cl₂, RT, 2 h; c) (±)-2 (from 3, *n*BuLi/TMEDA, Et₂O), Et₂O, -78°C, 2 h, 63% (2 steps b and c); d) *t*BuLi, Et₂O, -30°C \rightarrow -20°C, 30 min, 70%. BAIB=[bis(acetoxy)iodo]benzene.

failed, resulting only in the recovery of the starting material, despite encouraging Marshall precedent.^[34]

Finally, we were pleased to find that the use of TEMPOfree radicals and a suitable co-oxidant led to a significant improvement. Recourse to BAIB as a co-oxidant delivered cleanly the oxidation product.[35] Unfortunately, fast protodestannylation of 23 upon purification or even shelf storage was observed. Therefore, the crude product was directly subjected to crotyltitanation conditions with racemic (\pm) -2. The O-enecarbamate (\pm) -24 was then converted into terminal alkyne (\pm) -25 by *t*BuLi treatment. After the difficulties encountered during the oxidation step, it was gratifying to find that the (Z)-vinyl stannane function was sufficiently stable toward crotyltitanation conditions and especially upon tBuLi treatment. Thus, the 12-desmethyl C8-C14 fragment (\pm) -24 could be obtained in only four steps, from commercially available 2,3-dihydrofuran 8, in 26.9% overall yield.

With secured information on the C14 tin function stability in hand, the stage was set for the synthesis of DHF **9** bearing a methyl group at the C12-position (Scheme 5). Prepara-

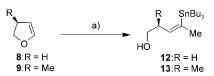


Scheme 5. a) NaCN, DMSO, 60°C, 24 h, 92%; b) DIBAL-H, $CH_2CI_{2,}$ -78°C, 2 h, 82%; c) PTSA, quinoline, 160°C \rightarrow 245°C, 45 min, 85%. DIBAL-H = diisobutylaluminium hydride.

tion of DHF 9 was achieved from commercially available (R)-3-bromo-2-methyl-1-propanol (26). Cyanide displacement of the bromide and partial reduction of the nitrile function led to lactol 27 after spontaneous cyclization of the γ -hydroxy aldehyde.

After several attempts, a straightforward dehydration with PTSA in quinoline with simultaneous distillation of the volatile elimination product from the reaction mixture cleanly afforded the desired DHF **9** in 85% yield (Scheme 5).^[36] This expeditious three-step sequence furnished the enantiopure DHF **9** in 64% overall yield.

DHF 9 was subjected to metallate rearrangement under the standard laboratory conditions with a cyano-Gilman dimethylcuprate to afford the expected trisubstituted (Z)alkene 13 in a disappointing 27% yield (Scheme 6; Table 1,



Scheme 6. Cuprate rearrangement with methyl transfer from DHF 8 or 9. a) 1) *t*BuLi (1.2 equiv), THF, -60 °C 10 min then 0 °C 50 min; 2) MeLi (5 equiv), CuCN (2 equiv), Et₂O, additive (see Table 1), 0 °C \rightarrow RT; 3) *n*Bu₃SnCl, -30 C \rightarrow RT, 5 h.

Table 1. Results of cuprate rearrangement with methyl transfer from DHF ${\bf 8}$ or ${\bf 9}.$

Entry	Starting material	Additive	Conditions	Yield of (Z)- 12 or (Z)- 13 [%]
1	8	none ^[a,b]	RT, 3 h	61
2	9	none ^[a,b]	RT, 3 h	27
3	8	$BF_3 \cdot OEt_2$ (2.5 equiv) ^[a,b]	RT, 3 h	13
4	8	TMSCl (4 equiv) ^[a,b]	RT, 3 h	46
5	8	HMPA (5 equiv) ^[a,b]	RT, 3 h	29
6	8	DMPU (5 equiv) ^[a,b]	RT, 3 h	0
7	8	DMF (20 equiv) ^[a,b]	RT, 3 h	0
8	8	TMEDA (10 equiv) ^[a,b]	RT, 3 h	45
9	8	pyridine (10 equiv) ^[a,b]	RT, 3 h	16
10	8	thiophene (20 equiv) ^[a,b]	RT, 12 h	55
11	8	tetrahydro-thiophene (20 equiv) ^[a,b]	RT, 12 h	67
12	8	DMS (Et ₂ O/DMS 4:1) ^[b]	RT, 12 h	76
13	8	none ^[a,c]	RT, 3 h	22
14	9	DMS (Et ₂ O/DMS 4:1) ^[b]	RT, 12 h	68

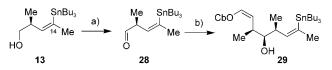
[a] THF/Et₂O 1:2. [b] MeLi/CuCN. [c] MeMgBr (4 equiv), CuBr-Me₂S (2 equiv). DMPU = N,N'-dimethylpropyleneurea; HMPA = hexamethylphosphoramide.

entry 2). To improve this result, it was decided to investigate the influence of different reaction parameters. To this end, the solvent and/or cuprate ligands were modified with regard to the classical 1:2 THF/Et₂O solvent mixture. The results, from commercial DHF 8 as a model, are summarized in Table 1. It was envisaged to activate the C-O bond of the dihydrofuran ring by addition of Lewis acids (Table 1, entries 3-4), to promote organometallic dissociation (Table 1, entries 5–7) or to modify the cuprate ligands in presence of TMEDA or pyridine (Table 1, entries 8-9), but without significant effect. However, the use of sulfur additives resulted in appreciable yields (Table 1, entries 10-12). Thereby, a 4:1 mixture of Et₂O and DMS (Table 1, entry 12) allowed the formation of the expected (Z)-alkene 12 in a satisfactory 76% yield. It was noted, however, that the magnesio-cuprates prepared from MeMgBr and CuBr·Me₂S (Table 1, entry 13), versus cyanocuprates, gave disappointing results and yields remained below 22%.

www.chemeurj.org

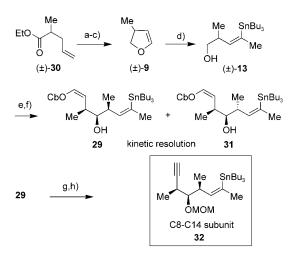
Application of optimized conditions to DHF 9 led to the (Z)-tin derivative 13 in a substantial 68% yield (Table 1, entry 14).

We developed a useful and efficient dimethylsulfide-promoted methyl transfer in the 1,2-cuprate rearrangement leading to (Z)-trisubstituted olefins. With an efficient supply of alkene **13** being assured, the synthesis of the C8–C14 subunit **B** was tackled. In agreement with the model C12-desmethyl series, compound **13** was converted to carbamate **29** as a single diastereomer in 87% yield after oxidation to aldehyde **28** with TEMPO/BAIB and crotyltitanation (Scheme 7). To confirm the (S)-C11 secondary hydroxy center configuration, **29** was derivatized to the (R)- and (S)- α -methoxy-phenylacetic acid (MPA) ester (see the Experimental Section for details).^[37]



Scheme 7. a) TEMPO (10 mol%), BAIB, CH₂Cl₂, RT, 2 h; b) (*R*)-**2**, cyclohexane/pentane 1:7, -78°C, 3 h, 87% (2 steps, a and b).

DHF 9 was also prepared in racemic form. After some experimentation, it was found that the most effective approach started from commercially available ethyl 2-methyl-4-pentenoate $((\pm)-30)$ (Scheme 8). Reduction of ester function followed by ozonolysis of the terminal double bond and dehydration of the corresponding lactol led to the desired DHF $(\pm)-9$. Oxidation of the 1,2-cuprate transfer product $(\pm)-13$

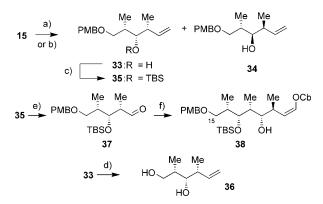


Scheme 8. a) LAH, Et₂O, 0°C \rightarrow RT, 3 h; b) O₃, Sudan III, CH₂Cl₂/MeOH 1:1, -78°C, 7 h then PPh₃, -78°C \rightarrow RT, overnight, 80% (2 steps, a and b), anomeric ratio 2:1; c) PTSA, quinoline, 160 \rightarrow 245°C, 45 min, 85%; d) *t*BuLi, Me₂CuLi-LiCN, Et₂O/DMS 4:1, 0°C \rightarrow RT, 18 h, then *n*Bu₃SnCl, -30°C \rightarrow RT, 5 h, 68%; e) TEMPO (10 mol %), BAIB, CH₂Cl₂, RT, 2 h; f) (*R*)-2, cyclohexane/pentane 1:7, -78°C, 3 h, compound 29 42% (2 steps, e and f), compound 31 14% (2 steps, e and f; g) *t*BuLi, THF, $-40\rightarrow-20$ °C, 20 min, 78%; h) MOMCl, TBAI, Hunig's base, CH₂Cl₂, RT, 18 h, 92%. DMS=dimethyl sulfide; Hunig's base=diisopropylethylamine; LAH=lithium aluminum hydride; MOM=methoxymethyl; TBAI=tetra-*n*-butylammonium iodide.

followed by crotyltitanation delivered the required Felkin-Anh homoallylic alcohols **29** in 42% yield and a minor diastereomer **31** in 14% yield. At this stage of the synthesis, it is important to point out that these C12-epimeric compounds **29/31** were easily separated by flash chromatography.

From carbamate 29, α -elimination reaction upon a *t*BuLi treatment and subsequent C11 secondary alcohol protection as the methoxymethyl ether furnished alkyne 32 (Scheme 8). Subunit 32 was obtained in eight steps in 13.9 or 26.3 % yields, respectively, from pentenoate (±)-30 (with a kinetic resolution step) or enantiopure propanol 26. However, taking into account the cost of the starting material 26 and the convenient separation of the C12-epimeric compounds, we preferred using the racemic DHF (±)-9 to produce C8-C14 subunit 32 on a multigram scale.

The synthesis of the C15–C24 subunit **A** started with the C16–C18 *syn–syn* motif preparation, planned by a substratebased crotylation reaction. The addition of the achiral tri*n*butylcrotylstannane on chiral aldehyde **15** under Keck's conditions (BF₃•OEt₂ in CH₂Cl₂),^[38] led to a 85:15 mixture of diastereomers **33** and **34**. Felkin–Anh selectivity was optimized to 95:5 when replacing CH₂Cl₂ by Et₂O and performing the reaction at a lower temperature (Scheme 9). Howev-



Scheme 9. a) Tri-*n*-butylcrotylstannane, BF₃·OEt₂, Et₂O, -100 °C, 2 h, 74%, d.r. 95:5; b) (*Z*)-2-butene, *n*BuLi, *t*BuOK, [(-)-Ipc]₂BOMe, BF₃·OEt₂, 65%, d.r. 100:0; c) TBSOTf, 2,6-lutidine, CH₂Cl₂, 0 °C \rightarrow RT, 2 h, 92%; d) DDQ, CH₂Cl₂/H₂O 95:5, 0 °C \rightarrow RT, 2 h, 75%; e) O₃, Sudan III, Py, CH₂Cl₂/MeOH 1:1, -78 °C, 2 h then DMS, -78 °C \rightarrow RT, overnight; f) (*R*)-**2**, cyclohexane/pentane 1:7, -78 °C, 3 h, 77%, d.r. 100:0. Ipc=isopinocampheyl, Py=pyridine.

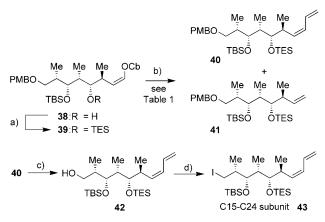
er, large-scale production of **33** was impeded by difficulties associated with diastereomer separation. This difficulty was readily overcome by using the chiral crotylborane procedure developed by Brown.^[39] The desired homoallylic alcohol **33** was cleanly formed in a stereospecific manner from the (*Z*)crotyldiisopinocampheylborane reagent derived from (–)-Ipc₂B(OMe).

The absolute configuration of the stereogenic centers of **33** was confirmed by chemical correlation. Cleavage of the PMB moiety was accomplished to provide the known triol **36**. The NMR spectroscopic data and the specific optical rotation were in agreement with the literature.^[40]

For the construction of the second C18–C20 *syn–anti* stereotriad, we selected a crotyltitanation reaction. After protection of the C17 hydroxy group of **33** as TBS ether, an oxidative cleavage of corresponding alkene **35** by ozonolysis delivered the aldehyde **37**. When the crotyltitanation reaction was realized with the chiral, non-racemic (R)-**2**, the desired isomer **38** was provided in 77% yield with total stereocontrol.

On the basis of extensive precedent, we predicted that the unique isomer from this addition would be the C18–C20 *syn–anti* stereoisomer **38**. This assignment was confirmed by applying the Rychnovsky method^[41] and analysis of the ¹H NMR vicinal coupling constants of the corresponding acetonides (see the Experimental Section for details).

Application of this crotyltitanation reaction was also instrumental in the preparation of the required terminal (Z)diene. As previously described by our group,^[11] the direct vinylation of a (Z)-O-enecarbamate moiety was carried out in the presence of [Ni(acac)₂] with commercially available vinylmagnesium bromide. In this way, from carbamate **39**, this reaction gave highly variable but mostly disappointing low yields (Scheme 10, see Table 2 entry 1). The cross-coupled product **40** was generated solely or as a mixture with the corresponding inseparable reduced compound **41** (selectivity varying from 100:0 to 0:100). The quality of commercial vi-



Scheme 10. a) TESOTf, 2,6-lutidine, CH_2CI_2 , 0°C \rightarrow RT, 1.5 h, 76%; b) [Ni(acac)₂], Et₂O, 0°C, overnight, vinylmagnesium bromide, see Table 1; c) DDQ, CH_2CI_2/H_2O 95:5, 0°C \rightarrow RT, 40 min, 72%; d) I₂, PPh₃, imid, C₆H₆/Et₂O, 0°C \rightarrow RT, 2 h, 79%. acac = acetylacetonate; imid=imidazole.

Table 2. Results of direct vinylation of a (Z)-O-enecarbamate by a [Ni-(acac)₂]-catalyzed cross-coupling reaction from compound **39**.

Entry	Conditions	40	41	Yield [%] ^[a]
1	CH ₂ =CH-MgBr ^[b]	$100 \rightarrow 0$	$0 \! \rightarrow \! 100$	$35 \rightarrow 80$
2	CH ₂ =CH-Br, Mg	80	20	65
3	CH ₂ =CH-Br, <i>t</i> BuLi, MgBr ₂ •OEt ₂	-	-	_[c]
4	(CH ₂ =CH) ₄ Sn, MeLi, MgBr ₂ •OEt ₂	100	0	50 ^[d]
5	(CH ₂ =CH) ₄ Sn, MeLi	100	0	80

[a] Isolated yield of the **40** and **41** mixture. [b] Commercial reagent. [c] Only the starting material was recovered. [d] 10% of starting material was recovered. nylmagnesium bromide solution was unreliable for this cross-coupling. Furthermore, even with the same batch, the selectivity between the cross-coupled product and the reduced compound decreased during the time of storage. We decided to prepare ourselves the vinylmagnesium bromide. Three methods were explored. First, direct insertion of magnesium into vinyl bromide^[42] led to reliable results but with a modest selectivity in favor of the desired compound 40, that is, 80:20 (Table 2, entry 2). The vinylmagnesium reagent was also generated in situ, by transmetallation of the vinyllithium derivative with magnesium bromide.^[43] When the vinyllithium was produced from vinyl bromide by metal-halogen exchange,^[44] no reaction occurred (Table 2, entry 3). On the other hand, the preparation of the vinyllithium from tetravinyltin and MeLi,^[45] cleanly afforded the cross-coupled product 40 in modest yield (Table 2, entry 4).

The reaction with solely vinyllithium species (without magnesium salts) produced from tetravinyltin and MeLi,^[45] led to a significant improvement (Table 2, entry 5). This method allowed the selective formation of the cross-coupled product **42** in a reproducible 80% yield (no reduction product was recovered). Use of tetravinylstannane and MeLi allowed the scaling up the cross-coupling reaction to 2.5 g. To the best of our knowledge, it is the first example of a C- (sp^2) -C (sp^2) cross-coupling between a vinyllithium and a vinylic electrophilic species.^[46]

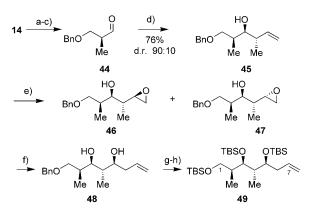
To achieve the synthesis of subunit **43**, the PMB ether was removed by treatment with DDQ and the resulting alcohol **42** was transformed into alkyl iodide **43** under optimized Garegg conditions in a benzene/diethyl ether mixture under high dilution.^[47] The C15–C24 subunit **43**, bearing five contiguous stereogenic centers and a terminal diene, was obtained in 13.7% overall yield for 11 steps.

Our initial approach to the synthesis of the third building block C1-C7 C started with standard Brown crotylation of aldehyde 44 (derived from (S)-Roche ester 14), to afford the desired syn-anti homoallylic alcohol 45 in 76% yield (Scheme 11). It was envisaged to rely upon a substrate-directable epoxidation/oxirane regioselective ring-opening sequence for the installation of the missing C5 chiral center. The epoxidation of homoallylic alcohol 45 turned out to be more difficult than expected. The use of mCPBA, [Mo(CO)₆]/TBHP^[48] or [VO(O*i*Pr)₃]/TBHP never gave more than 40% selectivity, whereas recourse to [VO-(acac)₂]/TBHP^[49] led to a significant improvement (78% yield of oxiranes 46 and 47 as a 90:10 mixture of separable diastereomers). However, upon attempted optimization and scaling up, this reaction was not reproducible. Pure 46 was then subjected to a regioselective ring opening by a cyano-Gilman divinyl cuprate, (CH₂=CH)₂CuLi·LiCN,^[50] to ensure the formation of the alkene 48 in 86% yield. Finally, cleavage of the OBn group with lithium in liquid ammonia followed by silulation afforded the C1-C7 subunit 49.

An alternative route allowed the generation of the C5 chiral center by hetero-Michael addition on α , β -unsaturated Weinreb amide 53, which in turn could be derived from aldehyde 50 through a crotyltitanation reaction and homolo-

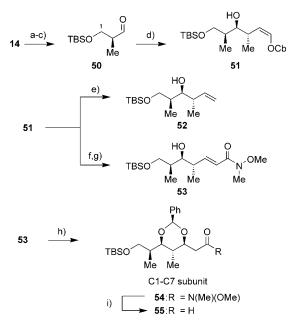
Chem. Eur. J. 2008, 14, 11092-11112

© 2008 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim



Scheme 11. a) CCl₃C(N)OBn, TfOH (5 mol%), CH₂Cl₂/cyclohexane 1:2, 0°C \rightarrow RT, 40 h, 84%; b) LAH, THF, 0°C \rightarrow RT, 3 h, 94% c) IBX, DMSO, RT, 3 h, 92%; d) (*E*)-2-butene, *n*BuLi, *t*BuOK, [(–)-Ipc]₂BOMe, BF₃·OEt₂, 76%, d.r. 90:10; e) [VO(acac)₂], TBHP, CH₂Cl₂, 78%, d.r. 90:10; f) CH₂=CH-Br, *t*BuLi, CuCN, THF, -78 \rightarrow 0°C, overnight, 86%; g) Li, NH₃, THF, -78°C, 1 h; h) TBSOTf, 2,6-lutidine, CH₂Cl₂, 0°C \rightarrow RT, 3 h, 99% (2 steps, g and h). TBHP=*tert*-butyl hydroperoxide.

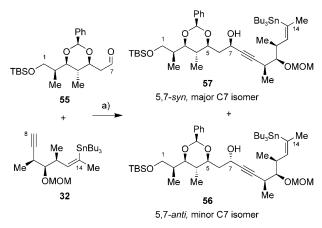
gation (Scheme 12). Thus, crotyltitanation of aldehyde **50** (which was prepared from (*S*)-Roche ester **14**) cleanly led to the *syn-anti* homoallylic alcohol **51**. A cross-metathesis (CM) reaction between alkene **51** and *N*-methoxy-*N*-methylacrylamide was tested.^[51] None of the desired product was formed and most of the starting material was recovered. We suspected that the *N*,*N*-diisopropyl-*O*-enecarbamate function could have deleterious effects for steric or/and electron-



Scheme 12. a) TBSCl, imid, DMF, RT, 3 h, 94%; b) DIBAL-H, CH_2Cl_2 , -20°C, 30 min, 92%; c) (COCl)₂, DMSO, Et₃N, -55°C \rightarrow RT, 1 h; d) (*R*)-**2**, cyclohexane/pentane 1:7, -78°C, 3 h, 66% (2 steps, c and d); e) [Ni-(acac)₂] (10 mol%), *i*PrMgBr, THF, RT overnight, 50%; f) O₃, Sudan III, CH₂Cl₂, -78°C, 1 h, then PPh₃, -78°C \rightarrow RT, 3 h, 87%; g) (EtO)₂P(O)CH₂C(O)NMe(OMe), NaH, THF, 0°C \rightarrow RT, 45 min, 86%; h) PhCHO, KHMDS, THF, -20°C, 45 min, 79%; i) DIBAL-H, CH₂Cl₂, -78°C, 1 h, 93%.

ic reasons. Thus, the vinylic function of compound 51 was deoxygenated by hydride reduction with iso-propylmagnesium chloride in a nickel-catalyzed reaction to furnish 52.^[52] More surprisingly, the CM reactions between N-methoxy-Nmethylacrylamide or methyl acrylate and terminal olefinic compound 52 failed when using Grubbs 2nd-generation or Hoveyda catalyst. Gratifyingly, a two-step sequence involving an oxidative cleavage of the double bond, followed by a Horner-Wadsworth-Emmons reaction (HWE) with commercially available diethyl (N-methoxy-N-methylcarbamoylmethyl)phosphonate, gave the α,β -unsaturated Weinreb amide 53. Reaction of 53 with benzaldehyde mediated by KHMDS afforded benzylidene 54 in 79% yield with an excellent diastereoselectivity (d.r.>95:5).[53] After reduction of 54 into aldehyde 55 by treatment with DIBAL-H,^[54] the last C1-C7 subunit, bearing four stereocenters, was obtained in 32.7% overall yield for eight steps.

Completion of the total synthesis of discodermolide: With access to the required fragments now reliable, the stage was set for the coupling. The first coupling reaction involved an addition of the C8–C14 acetylenic compound **32** to C1–C7 aldehyde **55** (Scheme 13). Previous observations on the sta-

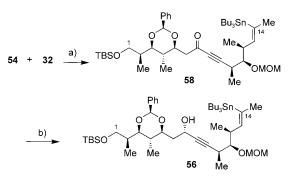


Scheme 13. a) **32**, *t*BuLi, THF, -78 °C, 30 min, then **55**, $-78 \rightarrow 0$ °C, 3 h, 50 %, d.r. 1:2.

bility of the sterically hindered trisubstituted (Z)-vinyl stannane provided encouraging precedent with regard to the chemoselectivity issue. Although the lithiated acetylide of **32** was chemoselectively produced by action of *t*BuLi at low temperature, its reaction with aldehyde **55** afforded a 2:1 mixture of separable C7 diastereomers **56** and **57** in a modest 50% yield.

To assign the C7 stereochemistry of both these epimers, each C7 isomeric secondary alcohol was independently subjected to the formation of (R)- and (S)-MPA ester derivatives (see the Experimental Section for details). To our surprise, the desired and expected 5,7-anti diol **56** was the minor product (Scheme 13).^[55]

A more appropriate solution was found with the addition of acetylenic derivative **32** to Weinreb amide **54** followed by stereoselective reduction of ynone **58** (Scheme 14). Preliminary assays were conducted by using *t*BuLi as a metallating agent and afforded cleanly the desired adduct **58** in 80% yield. However, upon attempted optimization and scale up,

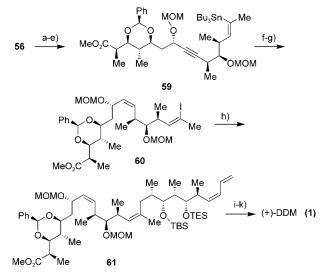


Scheme 14. a) **32**, *n*BuLi/Et₂O, THF, -40° C, 50 min, then **54**, $-40 \rightarrow >0^{\circ}$ C, 1 h, 81%; b) (*S*)-(-)-2-methyl-CBS-oxazaborolidine, BH₃·Me₂S, THF, -30° C, 2 h, 95%, d.r. 98:2. CBS = Corey–Bakshi–Shibata.

this reaction was not reproducible. During these initial attempts, however, it was noticed that Weinreb amide **54** was totally consumed with the production of a significant amount of benzaldehyde, which indicated a probable retrohetero-Michael reaction, followed by a vinylogous retroaldol reaction. After careful optimization, it was found that this addition reaction proceeded best when using a solution of *n*BuLi generated by treatment of butyl bromide by lithium metal in Et₂O,^[56] whereas protocols involving commercial *n*BuLi, LDA, or LiHMDS did not bring any significant improvement. This procedure led cleanly and faster to the desired ynone **58** in a reliable 81% yield. Subsequent reagent-controlled stereoselective reduction of **58** with (*S*)-CBS reagent^[57] provided alcohol **56** in high yield (95%) and diastereoselectivity (d.r. > 98:2).

The coupling of the two C1–C7 and C8–C14 fragments now in hand required the consecutive formation of vinyl iodide **60**, in view of an alkyl-Suzuki reaction with the C15– C24 subunit **43**. Alcohol **56** was smoothly converted to methyl ester **59** by a two-step MOM-protection reaction and TBS-ether cleavage protocol followed by oxidation of the resulting alcohol and esterification (Scheme 15). Subsequent standard Lindlar hydrogenation of the alkyne **59** seemed optimal relative to the discodermolide C8=C9 semi-reduction cases previously reported in the literature.^[4] However, this reduction only resulted in recovery of the starting material. After some experimentation, it was found that the reaction preceded best when using a PtO₂ protocol without overreduction. Consecutive iododestannylation furnished the desired (Z)-vinyl iodide **60** in 71 % for two steps.

The stage was set for the coupling of the vinyl iodide **60** and C15–C24 subunit **43**. An alkyl-Suzuki reaction was performed according to the Marshall et al. procedure,^[58] by using trialkyl boronate species prepared from alkyl iodide **43** with *t*BuLi and *B*-methoxy-9-BBN under [PdCl₂(dppf)] and AsPh₃ conditions to afford the discodermolide backbone **60** in 60 % yield.



FULL PAPER

Scheme 15. a) MOMCl, TBAI, Hunig's base, CH_2Cl_2 , RT, 18 h, 86%; b) HF/Py, Py/THF, RT, 4 h, 96%; c) TEMPO (10 mol %), BAIB, CH_2Cl_2 , RT, 3 h; d) NaClO₂, NaH₂PO₄, 2-methyl-but-2-ene, *t*BuOH/H₂O, RT, 1 h; e) TMSCHN₂, C₆H₆/MeOH, RT, 15 min, 66% (3 steps, c–e); f) H₂, PtO₂ (30 mol %), AcOEt, RT, 12 h; g) I₂, CH_2Cl_2 , 0°C, 10 min, 71% (2 steps, f and g); h) **43**, *t*BuLi, Et₂O, -78°C, 5 min, *B*-methoxy-9-BBN, THF, -78°C \rightarrow RT, then **60**, Cs₂CO₃, [PdCl₂(dppf)], AsPh₃, DMF, H₂O, 18 h, 60%; i) PSA, MeOH, 0°C, 1 h, 77%; j) Cl₃CC(O)NCO, CH₂Cl₂, RT, 15 min, then K₂CO₃, MeOH, RT, 1.5 h, 64%; k) HCl 4N, THF, RT, 72 h, 70%. BBN = borabicyclo[3.3.1]nonane; dppf = (diphenylphosphino)ferrocene.

Selective cleavage of the C19 TES-ether followed by reaction with trichloroacetylisocyanate furnished the corresponding carbamate.^[59] Final total deprotection with concomitant lactonization by following the procedure established in the previous total syntheses^[4] afforded discodermolide **1** in 70% yield. The spectroscopic and analytical data of the synthetic samples of **1** and their in vitro cytotoxicity levels were in full accord with those of the natural product reported in the literature.

Conclusions

The investigation outlined above resulted in a total synthesis of the anticancer marine metabolite discodermolide (1) in 1.6% overall yield for 21 linear steps. The chosen approach bears witness to the maturity of the crotyltitanation reaction with an enantiomerically defined secondary crotyltitane reagent. The repeated application of this reaction yielded homoallylic (Z)-O-enecarbamate alcohols 16, 29, 38, and 51 with excellent diastereoselectivity. Moreover, advantage was taken of these alkenyl carbamates to form the next carboncarbon bond. Particularly notable is the direct construction of the terminal C21–C24 (Z)-diene 40, with high geometric control, utilizing an original cross-coupling nickel-catalyzed reaction. Another highlight of our strategy was the 1,2-cuprate rearrangement of the dihydrofuran 9 leading to the stereocontrolled building of the C13=C14 trisubstituted (Z)alkene 13 with total selectivity.

For the key steps employed en route to 1, the most significant problems were posed by the methyl transfer during the C13=C14 trisubstituted (Z)-double bond installation and the construction of the middle C8-C14 fragment in two sequential stages (dyotropic rearrangement/crotyltitanation). After careful optimization, these reactions allowed high and reliable yields. Hence, this study illustrates the notion that the total synthesis of complex target molecules remains the ultimate test for the performance, scope and limitations of any newly development methodology.

The strategy is flexible enough to accommodate systematic structural variations and, therefore, to provide novel discodermolide structural analogues.

Experimental Section

General: All reactions were carried out in oven or flame-dried glassware under an argon atmosphere by employing standard techniques in handling air-sensitive materials. All solvents employed were reagent grade. THF and diethyl ether (Et₂O) were freshly distilled from sodium/benzophenone under nitrogen immediately prior to use. Dichloromethane, cyclohexane and pentane were freshly distilled over calcium hydride. All other reagents were used as supplied. Reactions were magnetically stirred and monitored by TLC with 0.20 mm SDS 60F254 pre-coated silica-gel plates. Visualization was accomplished with UV light then treatment with a 10% ethanolic phopshomolybdic acid solution followed by heating. Flash chromatography was performed with silica gel 60 (particle size 0.040-0.063 mm) supplied by SDS. Yield refers to chromatography and spectroscopically pure compounds, unless otherwise noted. ¹H NMR spectra were recorded by using an internal deuterium lock at ambient temperature on a JEOL JNM-ECX 270 or 400 MHz spectrometer. Internal references of $\delta_{\rm H}\!=\!7.26$ and 1.96 ppm were used for CDCl3 and CD₃CN, respectively. Spectroscopic data are represented as follows: chemical shift (in ppm), multiplicity (s=single, d=doublet, t=triplet, q=quartet), integration, coupling constant (J/Hz⁻¹). ¹³C NMR spectra were recorded on a Jeol 67.5 or 100.5 MHz spectrometer. Internals references of $\delta_{\rm C}$ = 77.16 and 118.26 ppm were used for CDCl₃ and CD₃CN, respectively. IR spectra were recorded on a Nicolet Impact-400 and wavelength (\tilde{v}) is given in cm⁻¹. Mass spectra were recorded on a GC/MS coupling unit with a MSD 5973 spectrometer and a Hewlett-Packard HP-GC 6890 chromatograph. Ionization was obtained either by electronic impact (EI) or chemical ionization with methane (CI, CH₄). Mass spectral data are reported as m/z. Optical rotations were recorded on a Jasco P-1010 digital polarimeter at 589 nm and are reported as follows: $[\alpha]_{\rm D}^{20}$, concentration (c in g100 mL), and solvent. Elemental analysis were performed on a CHN 240 Perkin-Elmer instrument by the Service de Microanalyses, Centre d'Etudes Pharmaceutiques, Chatenay-Malabry, F-92296. HRMS were obtained on a Thermo-Electron MAT-95 spectrometer in the ICMMO, Mass Spectrometry Laboratory, Orsay University, 91405 Orsay. IUPAC nomenclature was used for all compounds. RN refers CAS registration numbers. See Supporting Information for further experimental procedures.

(3Z)-4-(Tributylstannyl)pent-3-en-1-al (23): BAIB (10.4 g, 32.2 mmol, 1.1 equiv) was added to a solution of alcohol 12 (11.0 g, 29.3 mmol, 1.0 equiv) and TEMPO (457 mg, 2.93 mmol, 0.1 equiv) in CH₂Cl₂ (100 mL). The mixture was stirred until the starting material had disappeared (2 h). After this time, the mixture was diluted with cyclohexane, washed with an aqueous saturated Na₂S₂O₃ solution and extracted with cyclohexane. The combined organic phases were washed with a saturated aqueous NaHCO₃ solution and brine, dried over MgSO₄, and the solvent was removed under reduced pressure. The crude aldehyde 23 was directly used in the next step. ¹H NMR (400.0 MHz, CDCl₃): δ =9.67 (t, *J*= 2.1 Hz, 1H), 6.16 (tq, *J*=7.3, 1.9, *J*_{1H,175n}=*J*_{1H,175n}=12.1 Hz, 1H), 2.89 (dd, *J*=7.3, 2.1 Hz, 2H), 1.98 (d, *J*=1.9, *J*_{1H,175n}=*J*_{1H,175n}=3.8.9 Hz, 3H;

CH₃), 1.60–1.40 (m, 6H; 3CH₂), 1.39–1.20 (m, 6H; 3CH₂), 1.02–0.84 ppm (m, 15H; 3CH₂, 3CH₃).

(1Z,6Z,3R/S,4R/S)-1-[(N,N-Diisopropyl)carbamoyloxy]-4-hydroxy-3methyl-7-(tributylstannyl)oct-1,6-diene ((\pm)-24): A solution of *n*BuLi (1.6 M in hexane, 44 mL, 70.0 mmol, 2.04 equiv) was added to a quickstirred solution of the (E)-crotyl (diisopropyl)carbamate (11.7 g, 59.0 mmol, 2.0 equiv) and TMEDA (6.8 g, 59.0 mmol, 2.0 equiv) in Et_2O (70 mL) at -78 °C. After 30 min at -78 °C, Ti(OiPr)₄ (52.0 mL, 176.0 mmol, 6.0 equiv) was quickly added by cannula to the reaction mixture of lithio carbamate, which turned orange. After 30 min at -78 °C, aldehyde 23 (10.9 g, 29.3 mmol, 1.0 equiv) in Et₂O (15 mL) was slowly added to the orange solution, and the mixture was stirred for 3 h at -78 °C. The solution was then poured into a mixture of Et₂O-saturated aqueous NH₄Cl solution. After extraction with Et₂O, the organic layer was washed with brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/Et₂O 95:5 to 60:40) to give the title compound (\pm)-24 (7.4 g, 44% for 2 steps). ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.12$ (d, J = 6.4 Hz, 1 H), 6.10 (ddq, J = 5.9, 5.5, 1.4, $J_{^{1}H_{^{117}Sn}} =$ $J_{1_{\text{H},119}\text{Sn}} = 131.0 \text{ Hz}, 1 \text{ H}$), 4.67 (dd, J = 9.9, 6.4 Hz, 1 H), 4.19–4.11 (brs, 1H), 3.80–3.75 (brs, 1H), 3.51 (dddd, J=8.6, 5.0, 3.4, 3.0 Hz, 1H), 2.81 (ddq, J=9.9, 8.6, 7.3 Hz, 1 H), 2.30-2.15 (m, 1 H), 2.10-2.06 (m, 1 H), 1.92 (d, J = 1.4, $J_{1H,117Sn} = J_{1H,119Sn} = 41.7$ Hz, 3H; CH₃), 1.78 (br d, J = 3.0 Hz, 1H; OH), 1.58-1.38 (m, 6H; 3CH₂), 1.38-1.17 (m, 18H; 3CH₂, 4CH₃), 1.07 (d, J = 7.3 Hz, 3H; CH₃), 0.96–0.85 ppm (m, 15H; 3CH₂, 3CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 152.3$ (C), 141.9 (C), 136.4 ($J_{^{13}C_{117}Sn} =$ J_{13C,119Sn}=28.1 Hz; CH), 135.3 (CH), 111.9 (CH), 74.1 (CH), 45.4 (CH), 45.2 (CH), 39.6 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 30.7 \text{ Hz}; \text{ CH}_2), 35.4$ (CH), 28.8 $(J_{^{13}C_{117}Sn} = J_{^{13}C_{119}Sn} = 19.2 \text{ Hz}; 3 \text{ CH}_2), 27.5 \quad (J_{^{13}C_{117}Sn} = J_{^{13}C_{119}Sn} = 56.5 \text{ Hz};$ 3 CH_2), 26.8 ($J_{^{13}\text{C},^{117}\text{Sn}} = J_{^{13}\text{C},^{119}\text{Sn}} = 46.0 \text{ Hz}$; CH₃), 21.1 (2 CH₃), 20.3 (2 CH₃), 17.0 (CH₃), 13.2 (3 CH₃), 9.6 ppm ($J_{^{13}C^{117}Sn} = 313.4$, $J_{^{13}C^{119}Sn} = 326.8$ Hz; 3 CH₂); IR (film): $\tilde{\nu}$ = 3470, 2957, 2925, 2872, 2855, 1712, 1705, 1679, 1642, 1456, 1444, 1371, 1306, 1211, 1147, 1134, 1065, 1001, 961, 903, 885, 865, 762 cm^{-1} .

(3*R*/*S*,4*R*/*S*,6*Z*)-4-Hydroxy-3-methyl-7-(tributylstannyl)oct-6-en-1-yne

((±)-25): A solution of tBuLi (1.5м in pentane, 27.2 mL, 40.8 mmol, 3.0 equiv) was slowly added to a solution of the preceding carbamate (\pm) -24 (8.4 g, 13.6 mmol, 1.0 equiv) in Et₂O (100 mL) at -40 °C. The solution was stirred for 20 min at -20 °C and was then guenched by the addition of a saturated aqueous NH₄Cl solution, and extracted with Et₂O. The combined organic phases were washed with brine, dried over MgSO₄, filtered, and concentrated under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/Et₂O 90:10 to 80:20) to give (\pm)-25 (4.7 g, 78%). ¹H NMR (270.0 MHz, CDCl₃): $\delta = 6.00$ (ddq, J = 7.7, 6.5, 1.7, $J_{^{1}H_{1}^{117}Sn} = J_{^{1}H_{1}^{119}Sn} = 129.0$ Hz, 1H), 3.47-3.42 (m, 1H), 2.54 (qdd, J=6.9, 4.0, 2.6 Hz, 1H), 2.26 (m, 1H), 2.16 (m, 1H), 2.14 (d, J=2.3 Hz, 1H), 1.86 (d, J=1.7, $J_{^{1}H_{1}}{}^{_{17}S_{n}}=J_{^{1}H_{1}}{}^{_{19}S_{n}}=$ 41.6 Hz, 3H; CH₃), 1.50-1.33 (m, 6H; 3CH₂), 1.32-1.11 (m, 10H; 1OH, 1 CH₃, 3 CH₂), 0.98–0.74 ppm (m, 15 H, 3 CH₂, 3 CH₃); ¹³C NMR (67.5 MHz, CDCl₃): $\delta = 142.8$ (C), 136.1 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 26.8$ Hz; CH), 85.1 (C), 74.0 (CH), 71.0 (CH), 40.1 (CH₂), 32.2 (CH), 29.2 (J_{13C,117Sn}= $J_{^{13}C,^{119}Sn} = 19.5 \text{ Hz}; 3 \text{ CH}_2), 27.4 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 56.2 \text{ Hz}; 3 \text{ CH}_2), 17.3$ (CH₃), 13.7 (3 CH₃), 9.6 ppm ($J_{^{13}C_{1}^{11}Sn}$ = 314.4, $J_{^{13}C_{1}^{119}Sn}$ = 328.8 Hz; 3 CH₂); IR (film): $\tilde{\nu}$ =3311, 2956, 2926, 2871, 2854, 1712, 1463, 1376, 1260, 1192, 1072, 1039, 960 cm⁻¹.

(2*R*/S,4*R*/S)-4-Methyltetrahydrofuran-2-ol ((±)-27): A solution of commercial ethyl 4-methylpentenoate (±)-30 (30 g, 211 mmol, 1.0 equiv) in Et₂O (200 mL) was added to a solution of LiAlH₄ (6.4 g, 169 mmol, 0.8 equiv) in Et₂O (100 mL) at 0°C over 30 min. After the reaction mixture had been stirred for 4 h at 20°C, the mixture was cooled at 0°C and H₂O (6.5 mL), an aqueous NaOH 15% solution (6.5 mL), and further H₂O (19.5 mL) were added successively. The mixture was stirred for 2 h, and filtered through a pad of Celite. The filtrate was dried over MgSO₄ and the solvent was removed under reduced pressure. The crude 2-methylpent-4-en-1-ol was directly used in the next step. ¹H NMR (400.0 MHz, CDCl₃): δ =5.75 (dddd, *J*=10.1, 1.4 Hz, 1H), 3.55 (ddd, *J*=6.9, 6.0, 4.6 Hz, 1H), 3.45 (ddd, *J*=6.4, 6.0, 4.6 Hz, 1H), 2.10 (ddd, *J*=8.2, 6.9, 6.0 Hz,

1 H), 1.88 (ddd, J = 8.2, 7.3, 6.4 Hz, 1 H), 1.66 (quinttd, J = 6,9, 6.4, 6.0 Hz, 1 H), 1.40 (t, J = 4.6 Hz, 1 H; OH), 0.85 ppm (d, J = 6.9 Hz, 3 H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ = 136.9 (CH), 116.0 (CH₂), 67.7 (CH₂), 37.7 (CH₂), 35.5 (CH), 16.3 ppm (CH₃); MS (GC, EI): m/z: 82 [M-18]⁺, 67, 58.

A stream of ozone was bubbled into a solution of the preceding 2methyl-pent-4-en-1-ol (24.2 g, 241.5 mmol, 1.0 equiv) and Sudan III (small amount) in CH2Cl2/MeOH (1:1, 200 mL) at -78 °C until the pink solution became colorless (ca. 7 h). Then PPh3 (63.3 g, 241.5 mmol, 1.0 equiv) was cautiously added (over 30 min), the cold bath was removed and the mixture was stirred at 20 °C for one night. After this time, the solvents were removed under reduced pressure (20 mmHg) and the product was distilled out of the reaction mixture by using a microdistillation apparatus under reduced pressure (0.2 mmHg) at 70 °C to give lactol (\pm) -27 (19.7 g, 80%) as a colorless oil and a 1:2 mixture of two diastereomers. RN: 34314-85-7; first anomer: ¹H NMR (400.0 MHz, CDCl₃): $\delta = 5.46$ (d, J = 5.0 Hz, 1 H), 4.09 (t, J = 7.8 Hz, 1 H; H_a-11), 3.32 (dd, J =7.8, 7.3 Hz, 1H; H_b-11), 2.80 (brs, 1H; OH), 2.24 (dddad, J=9.2, 7.8, 7.3, 6.9, 6.9 Hz, 1 H), 1.97 (dd, J=12.8, 6.9 Hz, 1 H), 1.51 (ddd, J=12.8, 9.2, 5.0 Hz, 1 H), 1.03 ppm (d, J=6.9 Hz, 3 H; CH₃); second anomer: ¹H NMR (400.0 MHz, CDCl₃): $\delta = 5.46$ (d, J = 4.6 Hz, 1 H), 3.95 (dd, J =8.2, 6.6 Hz, 1 H), 3.55 (dd, J=8.7, 8.2 Hz, 1 H), 2.92 (brs, 1 H; OH), 2.58-2.48 (m, 1H), 1.76–1.70 (m, 1H), 1.47–1.40 (m, 1H), 1.09 ppm (d, J =6.4 Hz, 3H; CH₃); first anomer: 13 C NMR (100.5 MHz, CDCl₃): $\delta = 98.9$ (CH), 74.4 (CH₂), 41.8 (CH₂), 31.4 (CH), 17.8 ppm (CH₃); second anomer: ¹³C NMR (100.5 MHz, CDCl₃): δ=99.4 (CH), 73.5 (CH₂), 41.8 (CH₂), 33.3 (CH), 17.3 ppm (CH₃); IR (film): $\tilde{\nu}$ = 3406, 2960, 2935, 2875, 1455, 1379, 1346, 1290, 1122, 1095, 1056, 1002, 927 cm⁻¹; MS (GC, EI): m/z: 101, 84, 72, 56.

(45)-3-Methyl-2,3-dihydrofuran (9) and (4R/S)-3-methyl-2,3-dihydrofuran $[(\pm)-9]$

Preparation of optically pure 9: Sodium cyanide (4.62 g, 94.4 mmol, 1.2 equiv) was added to a solution of commercially available (R)-3bromo-2-methyl-propanol 26 (13.1 g, 85.8 mmol, 1.0 equiv) in DMSO (140 mL). The reaction was stirred for 24 h at 20 °C. After this time, the solution was poured out into water and extracted three times with EtOAc. To extract the remainder of the compound, the aqueous solution was carefully acidified to pH 5 by using 10% sulfuric acid. After extracting the aqueous layer three more times, the combined organic layers were washed with brine, dried over MgSO4, and the solvent was removed under reduced pressure. (2S)-3-Cyano-2-methyl propan-1-ol (7.6 g, 89%) was collected and used directly in the next step. ¹H NMR (400.0 MHz, CDCl₃): $\delta = 3.65$ (ddd, J = 10.5, 5.0, 4.6 Hz, 1 H), 3.50 (ddd, J = 10.5, 7.3, 5.0 Hz, 1 H), 2.62 (t, J=5.0 Hz, 1 H; OH), 2.50 (dd, J=16.7, 5.5 Hz, 1 H), 2.37 (dd, J=16.7, 6.9 Hz, 1 H), 2.06 (dquintdd, J=7.3, 6.9, 5.5, 4.6 Hz, 1 H), 1.08 ppm (d, J = 6.9 Hz, 3 H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 118.7$ (C), 65.6 (CH₂), 32.8 (CH), 20.8 (CH₂), 15.8 ppm (CH₃); IR (film): $\tilde{\nu} = 3416$, 2967, 2933, 2880, 2251, 1644, 1464, 1424, 1387, 1350, 1044, 993 cm⁻¹; MS (GC, EI): m/z: 99, 69, 59, 54.

A solution of DIBAL-H (1 M in CH₂Cl₂, 86 mL, 86 mmol, 2.5 equiv) was added to a solution of the preceding nitrile derivative (3.4 g, 34 mmol, 1.0 equiv) in CH₂Cl₂ (60 mL) at -78 °C. The mixture was stirred at -78 °C until the starting material had disappeared and it was then diluted with AcOEt and poured into a solution of Rochelle's salt (C₄H₄KNaO₆·4H₂O, 78 g, 8.0 equiv). After the reaction mixture had been stirred for 12 h, it was extracted with AcOEt, and the combined organic phases were washed with brine, dried over MgSO₄, and the solvent was removed under reduced pressure. (2*R*/S,4S)-4-Methyl tetrahydrofuran-2-ol **27** was obtained as a 1:2 diastereometic mixture (2.87 g, 82 %). RN: 34314-85-7 (see above for analytical data for racemic 4-methyltetrahydrofuran-2-ol (\pm)-**27**).

A solution of the above optically active lactol **27** (5.0 g, 49 mmol, 1.0 equiv) and PTSA (25 mg, 0.11 mmol, 0.0022 equiv) in quinoline (2.5 mL) was heated from 160 to 245 °C. Distillation of a mixture of the title compound with water was achieved in 45 min (b.p. 80–90 °C). After separation of water, dihydrofuran **9** was obtained (3.5 g, 85%). RN: 1708-27-6; ¹H NMR (400.0 MHz, CDCl₃): δ =6.30 (dd, *J*=2.8, 2.3 Hz, 1H), 4.94 (dd, *J*=2.8, 2.3 Hz, 1H), 4.38 (dd, *J*=9.6, 8.7 Hz, 1H), 3.85

(dd, J=8.7, 6.4 Hz, 1 H), 3.03 (dqdt, J=9.6, 6.9, 6.4, 2.3 Hz, 1 H), 1.07 ppm (d, J=6.9 Hz, 3 H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ = 145.3 (CH), 106.6 (CH), 76.8 (CH₂), 36.6 (CH), 20.8 ppm (CH₃).

Preparation of (\pm) -9: (\pm) -9 was obtained from racemic lactol (\pm) -27 using the procedure described for the preparation of 9 from lactol 27.

(25,3Z)-2-Methyl-4-tributylstannyl-pent-3-en-1-ol (13): MeLi (1.6 M solution in Et₂O, 94.4 mL, 151.0 mmol, 5.0 equiv) was added to a suspension of CuCN (5.41 g, 60.4 mmol, 2.0 equiv) in dry Et_2O (10 mL) and DMS (40 mL) at -30 °C. The solution was stirred at -30 °C for 5 min and then allowed to warm up to 0°C for 20 min (pale-yellow color). tBuLi (1.5 M solution in pentane, 24.2 mL, 36.2 mmol, 1.2 equiv) was added to a solution of dihydrofuran 9 (2.54 g, 30.2 mmol, 1.0 equiv) in dry Et₂O (50 mL) at -60 °C. The mixture was stirred at 0 °C for 50 min. After this time, the solution of the lithio-dihydrofuran prepared above was added by cannula to the cyanocuprate (Et₂O/DMS 4:1) and the reaction mixture was allowed to warm up to 20 °C (strong orange color) over 12 h. The mixture was cooled at -30 °C and tri-n-butyltin chloride (49.1 mL, 181.2 mmol, 5.0 equiv) was added. The reaction mixture was allowed to warm up to 20°C over 5 h. Finally, the reaction mixture was poured into a mixture of a saturated aqueous NH₄Cl solution and concentrated ammonia (4:1) at 0°C and stirred for 1 h at 20°C before extraction with Et₂O. The organic layer was washed with water and brine, dried over MgSO₄, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/Et₂O 100:0 to 60:40) to give the title compound 13 (8.21 g, 68%). ¹H NMR (400.0 MHz, CDCl₃): $\delta = 5.78$ (dq, J = 9.6, 1.8, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 128.7$ Hz, 1H), 3.47 (ddd, J=10.5, 8.7, 6.0 Hz, 1 H), 3.36 (ddd, J=10.5, 8.2, 3.7 Hz, 1 H), 2.16 (dddq, J=9.6, 8.7, 8.2, 6.9 Hz, 1 H), 1.92 (d, J=1.8, $J_{^{1}H,^{117}Sn}=J_{^{1}H,^{119}Sn}=$ 41.7 Hz, 3H; CH₃), 1.55–1.45 (m, 6H; 3CH₂), 1.39–1.28 (m, 7H; OH, 3CH₂), 0.95 (d, J=6.4 Hz, 3H; CH₃), 0.99–0.91 (m, 6H; 3CH₂), 0.90 ppm (m, 9H; 3CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 143.0$ ($J_{^{13}C_1^{117}Sn} =$ $J_{^{13}C,^{119}Sn} = 27.8 \text{ Hz}; \text{ CH}, 140.9 \text{ (C)}, 67.3 \text{ (CH)}, 42.4 \text{ } (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = J_$ 33.6 Hz; CH), 29.3 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 20.4$ Hz; 3CH₂), 27.5 $(J_{^{13}C,^{117}Sn} =$ $J_{^{13}C,^{119}Sn} = 57.5 \text{ Hz}; 3 \text{ CH}_2$, 27.2 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 46.0 \text{ Hz}; \text{ CH}_3)$, 17.4 (CH₃), 13.7 (3CH₃), 10.1 ppm ($J_{^{13}C_{1}^{117}Sn}$ =314.4, $J_{^{13}C_{1}^{119}Sn}$ =329.7 Hz; 3CH₂); IR (film): $\tilde{\nu} = 3327, 2956, 2928, 2920, 2871, 2853, 1678, 1623, 1483, 1456,$ 1377, 1340, 1292, 1072, 1037, 996, 864, 690, 668 cm⁻¹; MS (GC, CI, NH₃): m/z: 373 [MH-H₂O]+, 359 [MH-H₂O-Bu]+, 291 [Bu₃Sn]+, 261, 249, 235, 141, 113, 101, 85, 57; elemental analysis calcd (%) for C₁₈H₃₈OSn: C 55.55, H 9.84; found: C 55.73, H 10.02.

(2S,3Z)-2-Methyl-4-tributylstannyl-pent-3-en-1-al (28): BAIB (6.09 g, 18.9 mmol, 1.1 equiv) was added to a solution of alcohol 13 (6.7 g, 17.2 mmol, 1.0 equiv) and TEMPO (538 mg, 3.45 mmol, 0.2 equiv) in CH₂Cl₂ (40 mL). The mixture was stirred until the starting material had disappeared (2 h), and was then diluted with cyclohexane, washed with an aqueous saturated Na₂S₂O₃ solution, and extracted with cyclohexane. The combined organic phases were washed with a saturated aqueous NaHCO3 solution and brine, dried over MgSO4, and the solvent was removed under reduced pressure until the volume was approximately equal to 20 mL. This solution can be kept only for a few hours before use (the neat product **28** is more unstable). ¹H NMR (400.0 MHz, CDCl₃): $\delta =$ 9.48 (d, J = 1.7 Hz, 1 H), 5.72 (dq, J = 9.7, 1.6, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 122.1$ Hz, 1 H), 2.89 (dqd, J = 9.7, 6.4, 1.7 Hz, 1 H), 1.89 (d, $J = 1.6, J_{^{1}H_{,}^{117}Sn} = J_{^{1}H_{,}^{119}Sn} =$ 40.7 Hz, 3H; CH₃), 1.47–1.37 (m, 6H; 3CH₂), 1.24 (sext, J=7.3 Hz, 6H; 3CH₂), 1.08 (d, J=6.4 Hz, 3H; CH₃), 0.91-0.86 (m, 6H; 3CH₂), 0.83 ppm $(t, J = 7.3 \text{ Hz}, 9 \text{ H}; 3 \text{ CH}_3); {}^{13}\text{C} \text{ NMR} (100.5 \text{ MHz}, \text{CDCl}_3): \delta = 202.1 \text{ (CH)},$ 146.0 $(J_{^{13}C,^{117}Sn} = 346.0, J_{^{13}C,^{119}Sn} = 362.3 \text{ Hz}, C), 136.2 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} =$ 24.9 Hz; CH), 53.1 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 31.6$ Hz; CH), 29.1 $(J_{^{13}C,^{117}Sn} = 31.6$ Hz; CH), 29.1 (J_{^{13}C,^{117}Sn} = 31.6 $J_{^{13}C,^{119}Sn} = 19.2 \text{ Hz}; 3 \text{ CH}_2), 27.4 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 40.3 \text{ Hz}; \text{ CH}_3, \text{ CH}_3), 27.3$ $(J_{^{13}C_1^{117}Sn} = J_{^{13}C_1^{119}Sn} = 58.5 \text{ Hz}; 3 \text{ CH}_2), 14.4 \text{ (CH}_3), 13.7 \text{ (3 CH}_3), 10.1 \text{ ppm}$ $(J_{^{13}C,^{117}Sn} = 317.3, J_{^{13}C,^{119}Sn} = 332.6 \text{ Hz}; 3 \text{ CH}_2); \text{ IR (film): } \tilde{\nu} = 2958, 2927, 2872,$ 2854, 2809, 1726, 1614, 1463, 1426, 1377, 1072, 999, 854, 689 cm⁻¹; MS (GC, CI, NH₃): *m*/*z*: 331 [*M*-Bu]⁺, 291 [Bu₃Sn]⁺, 275, 255, 235, 217, 189, 177, 159, 147, 135, 121, 97.

(2*R/S*,3*Z*)-2-Methyl-4-tributylstannyl-pent-3-en-1-ol ((\pm)-13): The same protocol used before for the synthesis of 13 was employed to prepare (\pm)-13 (16.5 g, 67%) from (\pm)-9 (5 g). See above for data.

Chem. Eur. J. 2008, 14, 11092-11112

© 2008 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

J.-F. Betzer, A. Pancrazi, J. Ardisson et al.

(2*R/S*,3*Z*)-2-Methyl-4-tributylstannyl-pent-3-en-1-al ((\pm)-28): The same protocol used before for the synthesis of 28 was employed to prepare (\pm)-28 from (\pm)-13 (2.5 g). See above for data.

(1*Z*,3*S*,4*S*,5*S*,6*Z*)-1-[(*N*,*N*-Diisopropyl)carbamoyloxy]-3,5-dimethyl-7-tributylstannyl-octa-1,6-dien-4-ol (29) and (1*Z*,3*S*,4*S*,5*R*,6*Z*)-1-[(*N*,*N*-diisopropyl)carbamoyloxy]-3,5-dimethyl-7-tributylstannyl-octa-1,6-dien-4-ol (31)

From optically pure aldehyde 28: A solution of nBuLi (1.6 M in hexanes, 2.4 mL, 3.95 mmol, 3.0 equiv) was added to a quick-stirred solution of the (E)-crotyldiisopropylcarbamate (787 mg, 3.95 mmol, 3.0 equiv) and (-)sparteine (925 mg, 3.95 mmol, 3.0 equiv) in pentane (6 mL) and cyclohexane (1 mL) at -78 °C. After 10 min, white crystals appeared. After 3 h of crystallization at -78°C, a pre-cooled (-40°C) solution of Ti(OiPr)₄ (3.5 mL, 11.9 mmol, 9.0 equiv) in pentane (7 mL) was quickly added by cannula to the reaction mixture of lithio carbamate, which became limpid and turned orange. After 1 h at -78°C, aldehyde 28 (513 mg, 1.30 mmol, 1.0 equiv) in cyclohexane (5 mL) was slowly added to the orange solution, and the mixture was stirred for 3 h at -78 °C. The solution was then poured into a mixture of Et2O-saturated aqueous NH4Cl solution. After extraction with Et₂O, the organic layer was washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ AcOEt 95:5 to 80:20) to give the title compound 29 (669 mg, 87% for 2 steps) as a single diastereomer. $[\alpha]_D^{20} = +20.7$ (c=0.94 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): δ = 7.15 (d, J = 6.4 Hz, 1 H), 6.17 (dq, J = 9.9, 1.4, $J_{^{1}\text{H},^{117}\text{Sn}} = J_{^{1}\text{H},^{119}\text{Sn}} = 134.0 \text{ Hz}, 1 \text{ H}$), 4.66 (dd, J = 9.8, 6.4 Hz, 1 H), 4.19-4.02 (brs, 1H), 3.78-3.61 (brs, 1H), 3.25 (dt, J=8.0, 2.8 Hz, 1H), 2.86 (ddq, J=9.8, 8.0, 6.9 Hz, 1 H), 2.18 (dqd, J=9.9, 6.8, 2.8 Hz, 1 H), 1.90 (d, J = 1.4, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 42.6$ Hz, 3H; CH₃), 1.72 (br d, J = 2.8 Hz, 1H; OH), 1.55–1.45 (m, 6H; 3CH₂), 1.38–1.16 (sext, J=7.3 Hz, 6H; 3CH₂), 1.20–1.12 (m, 12H; 4CH₃), 1.00 (d, J=6.9 Hz, 3H; CH₃), 0.98 (d, J = 6.8 Hz, 3H; CH₃), 0.99–0.92 (m, 6H; 3CH₂), 0.89 ppm (t, J = 7.3 Hz, 9H; 3CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 152.9$ (C), 144.3 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 27.8 \text{ Hz}; \text{ CH}), 137.7 (J_{^{13}C,^{117}Sn} = 375.7, J_{^{13}C,^{119}Sn} = 392.0 \text{ Hz};$ C), 136.7 (CH), 113.2 (CH), 78.6 (CH), 47.1 (CH), 45.7 (CH), 41.3 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 32.6 \text{ Hz}; \text{ CH}), 34.5 \text{ (CH)}, 29.3 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} =$ 19.2 Hz; 3CH₂), 27.5 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 57.5$ Hz; 3CH₂), 27.2 ($J_{^{13}C,^{117}Sn} = 57.5$ Hz; 27.2 ($J_{^{13}C,^{117}Sn} = 57.5$ ($J_{^{13}C,^{119}Sn} = 46.0 \text{ Hz}; CH_3), 21.7 (2 CH_3), 20.4 (2 CH_3), 17.6 (CH_3), 14.2$ (CH₃), 13.8 (3 CH₃), 10.0 ppm ($J_{13C,117Sn}$ =313.4, $J_{13C,119Sn}$ =326.8 Hz; 3 CH₂); IR (film): $\tilde{\nu} = 3490$, 2957, 2929, 2872, 2851, 1705, 1463, 1455, 1441, 1370, 1301, 1289, 1210, 1135, 1065, 997, 862, 762 cm⁻¹; MS (GC, CI, NH₃): m/z: 530, 385, 327, 291 [Bu₃Sn]⁺, 257, 235, 199, 177, 12, 86, 69; elemental analysis calcd (%) for C29H57NO5Sn: C 59.39, H 9.802, N 2.39; found: C 59.51, H 9.90, N 2.30.

From racemic aldehyde (\pm) -28: A solution of *n*BuLi (1.6 M in hexane, 23 mL, 36.8 mmol, 2.14 equiv) was added to a quick-stirred solution of the (E)-crotyldiisopropylcarbamate (6.84 g, 34.4 mmol, 2.0 equiv) and (-)-sparteine (8.30 g, 35.4 mmol, 2.06 equiv) in pentane (36 mL) and cyclohexane (6 mL) at -78 °C. After 10 min, white crystals appeared. After 3 h of crystallization at -78°C, a pre-cooled (-40°C) solution of Ti-(OiPr)₄ (30.5 mL, 103.2 mmol, 6.0 equiv) in pentane (30 mL) was quickly added by cannula to the reaction mixture of lithio carbamate, which became limpid and turned orange. After 1 h at -78 °C, aldehyde (±)-28 (6.66 g, 17.2 mmol, 1.0 equiv) in cyclohexane (5 mL) was slowly added to the orange solution, and the mixture was stirred for 3 h at -78 °C. The solution was then poured into a mixture of Et₂O-saturated aqueous NH_4Cl solution. After extraction with Et₂O, the organic layer was washed with brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/AcOEt 95:5 to 80:20) to give the title compound 29 (8.50 g, 42% for 2 steps; see above for data) and its 12R minor isomer **31** (1.41 g, 14%). ¹H NMR (400.0 MHz, CDCl₃): δ = 7.10 (d, J = 6.9 Hz, 1H), 5.88 (dq, J = 9.8, 1.7, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 128.0$ Hz, 1H), 4.86 (dd, J=10.1, 6.9 Hz, 1 H), 4.12-3.99 (br s, 1 H), 3.91-3.78 (br s, 1 H), 3.19 (dd, J=8.4, 2.6 Hz, 1 H), 2.92 (dqd, J=10.1, 6.9, 2.6 Hz, 1 H), 2.00 (dqd, J=9.8, 8.4, 6.4 Hz, 1 H), 1.92 (d, J=1.7, $J_{1H,117Sn}=J_{1H,119Sn}=40.3$ Hz, 3H; CH₃), 1.55–1.45 (m, 6H; 3CH₂), 1.38–1.16 (sext, J = 7.3 Hz, 19H; $3CH_2 + OH + 4CH_3$, 1.13 (d, J = 6.9 Hz, 3H; CH₃), 0.96 (d, J = 6.4 Hz, 3 H; CH₃), 0.99–0.92 (m, 6 H, 3 CH₂), 0.89 ppm (t, J = 7.3 Hz, 9 H; 3 CH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ = 152.6 (C), 144.3 ($J_{^{12}C^{117}Sn} = J_{^{12}C^{119}Sn} =$ 28.8 Hz, CH), 142.2 ($J_{^{12}C^{117}Sn} = 355.1$, $J_{^{12}C^{119}Sn} = 372.4$ Hz, C), 135.3 (CH), 111.0 (CH), 77.7 (CH), 46.6 (CH), 46.0 (CH), 44.0 ($J_{^{12}C^{117}Sn} = J_{^{12}C^{119}Sn} =$ 29.8 Hz; CH), 32.1 (CH), 29.1 ($J_{^{12}C^{117}Sn} = J_{^{12}C^{119}Sn} = 20.2$ Hz; 3 CH₂), 27.4 ($J_{^{12}C^{117}Sn} = J_{^{12}C^{119}Sn} = 57.6$ Hz; 3 CH₂), 27.2 ($J_{^{12}C^{117}Sn} = J_{^{12}C^{119}Sn} = 42.2$ Hz; CH₃), 21.4 (2 CH₃), 20.3 (2 CH₃), 18.7 (CH₃), 17.3 (CH₃), 13.6 (3 CH₃), 10.1 ppm ($J_{^{12}C^{117}Sn} = 315.7$, $J_{^{12}C^{119}Sn} = 330.1$ Hz; 3 CH₂).

(3S,4R,5S,6Z)-3,5-Dimethyl-4-[(methoxymethyl)oxy]-7-(tributylstannyl)-

oct-6-en-1-yne (32): A solution of tBuLi (1.5 m in pentane, 27.2 mL, 40.8 mmol, 3.0 equiv) was slowly added to a solution of carbamate 29 (8.4 g, 13.6 mmol, 1.0 equiv) in Et₂O (100 mL) at -40 °C. The solution was stirred for 20 min at -20 °C, and was then quenched by the addition of a saturated aqueous NH₄Cl solution and extracted with Et₂O. The combined organic phases were washed with brine, dried over $MgSO_4$, filtered, and concentrated under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/Et₂O 90:10 to 80:20) to give (3S,4R,5S,6Z)-3,5-dimethyl-4-hydroxy-7-(tributylstannyl)oct-6-en-1-yne (4.7 g, 78%). $[\alpha]_{\rm D}^{20} = -26.8$ (c=1.0 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 5.87$ (dq, J = 9.6, 1.8, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 133.3$ Hz, 1H), 3.17 (ddd, J=9.0, 7.5, 3.2 Hz, 1H), 2.73 (dqd, J=7.5, 6.9, 2.3 Hz, 1H), 2.16 (dqd, J=9.6, 6.9, 3.2 Hz, 1H), 2.14 (d, J=2.3 Hz, 1H), 1.90 (d, J = 1.8, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 40.8$ Hz, 3H; CH₃), 1.74 (d, J = 9.0 Hz, 1H; OH), 1.53–1.42 (m, 6H; 3CH₂), 1.34 (sext, J = 7.3 Hz, 6H; 3CH₂), 1.26 (d, J =6.9 Hz, 3H; CH₃), 1.08 (d, J = 6.9 Hz, 3H; CH₃), 0.98–0.91 (m, 6H; 3CH₂), 0.89 ppm (t, J=7.3 Hz, 9H; 3CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 142.8 \ (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 28.8 \text{ Hz}; \text{ CH}), \ 138.8 \ (C), \ 84.8 \ (C),$ 78.0 (CH), 71.5 (CH), 43.2 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 33.6$ Hz; CH), 30.4 (CH), 29.0 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 19.2 \text{ Hz}; 3 \text{ CH}_2), 27.1 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 58.7 \text{ Hz};$ 3 CH), 27.0 (*J*_{13C,117Sn}=*J*_{13C,119Sn}=44.0 Hz; CH₃), 18.4 (CH₃), 17.0 (CH₃), 13.4 (3CH₃), 9.7 ppm ($J_{^{13}C,^{117}Sn}$ = 314.4, $J_{^{13}C,^{119}Sn}$ = 328.8 Hz; 3 CH₂); IR (film): $\tilde{\nu}$ = 3560, 350, 3456, 3310, 2957, 2926, 2872, 2854, 1463, 1455, 1376, 1247, 1115, 1070, 1050, 997, 981, 960, 873, 862, 690, 663, 631 cm⁻¹; elemental analysis calcd (%) for C22H42OSn: C 59.88, H 9.59; found: C 59.99, H 9.77.

Diisopropylethylamine (27.7 mL, 159 mmol, 20 equiv) and MOMCl (6.9 mL, 79.5 mmol, 10 equiv) were added to a solution of the above alcohol (3.51 g, 7.95 mmol, 1.0 equiv), dimethylaminopyridine (194 mg, 1.59 mmol, 0.2 equiv), and TBAI (294 mg, 0.795 mmol, 0.1 equiv) in dried CH₂Cl₂ (40 mL) at 0°C. After the reaction mixture had been stirred for 12 h at 20°C, it was quenched with a saturated aqueous NaHCO3 solution and extracted with Et2O. The organic layers were washed with brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/Et₂O 98:2 to 90:10) to give the title compound 32 (3.53 g, 92%) as a yellow oil. $[\alpha]_D^{20} = -20.6$ (c = 0.98 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 5.87$ (d, J = 9.6, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 133.3$ Hz, 1 H), 4.80 (d, J=7.3 Hz, 1H), 4.72 (d, J=7.3 Hz, 1H), 3.45 (s, 3H; CH₃), 3.19 (dd, J=7.8, 3.6 Hz, 1H), 2.74 (qdd, J=7.3, 3.6, 2.7 Hz, 1H), 2.33 (ddq, J=9.6, 7.8, 6.9 Hz, 1 H), 2.09 (d, J=2.7 Hz, 1 H), 1.88 (s, $J_{1H,117Sn}=J_{1H,119Sn}=$ 40.8 Hz, 3H; CH₃), 1.52–1.42 (m, 6H; 3CH₂), 1.32 (sext, J=7.3 Hz, 6H; 3 CH₂), 1.25 (d, J=7.3 Hz, 3H; CH₃), 1.02 (d, J=6.9 Hz, 3H; CH₃), 0.97-0.93 (m, 6H; 3CH₂), 0.89 ppm (t, J = 7.3 Hz, 9H; 3CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 143.6$ ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 28.7$ Hz, CH), 138.5 (C), 98.4 (CH₂), 86.1 (C), 85.9 (CH), 70.0 (CH), 56.3 (CH₃), 42.8 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 30.1 \text{ Hz}; \text{ CH}), 30.0 \text{ (CH)}, 27.3 \text{ } (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} =$ 20.1 Hz; 3CH₂), 27.3 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 59.4$ Hz; 3CH₂), 27.2 ($J_{^{13}C,^{117}Sn} = 59.4$ Hz J_{13C,119Sn}=40.2 Hz; CH₃), 18.8 (CH₃), 17.7 (CH₃), 13.7 (3 CH₃), 9.9 ppm $(J_{^{13}C_1^{17}Sn} = 314.4, J_{^{13}C_1^{19}Sn} = 328.7 \text{ Hz}; 3 \text{ CH}_2); \text{ IR (film): } \tilde{\nu} = 3312, 2956, 2926,$ 2872, 2823, 1458, 1376, 1146, 1096, 1035, 996, 668, 634 $\rm cm^{-1};$ elemental analysis calcd (%) for C24H46O2Sn: C 59.39, H 9.55; found: C 59.55, H 9.73.

(3*R*,4*S*,5*S*)-3,5-Dimethyl-6-(4-methoxybenzyloxy)hex-1-en-4-ol (33) and (3*S*,4*R*,5*S*)-3,5-dimethyl-6-(4-methoxybenzyloxy)hex-1-en-4-ol (34)

Brown reaction: A solution of cis-2-butene (12.5 mL) in THF (10 mL) was slowly added to a solution of freshly sublimed potassium tert-butoxide (7.56 g, 67.4 mmol, 1.2 equiv) in THF (100 mL) at -78 °C. nBuLi (1.6 m in hexane, 44 mL, 70.3 mmol, 1.25 equiv) was then added dropwise

and the resulting yellow mixture was stirred at -78°C for 5 min and at -45°C for 20 min. The resulting orange solution was cooled to -78°C, and a solution of (-)-B-methoxydiisopinocamphenylborane (25 g, 78.7 mmol, 1.4 equiv) in Et₂O (46 mL) was added dropwise over ca. 15 min. The resulting white solution was stirred at -78°C for 40 min. BF₃·OEt₂ (11.8 mL, 95.5 mmol, 1.7 equiv) was added dropwise followed after 5 min by the addition of a solution of the aldehyde 13 (11.7 g, 56.2 mmol, 1.0 equiv) in THF (25 mL). The resulting solution was stirred at -78°C for 5 h and the reaction was then quenched by the successive addition of an aqueous NaOH solution (2.5N, 66 mL) and an aqueous H₂O₂ solution (30%, 20 mL). The acetone/dry ice bath was then removed, and the mixture was heated at 45 °C for 45 min. The cloudy solution was cooled to 20°C, diluted with Et₂O (45 mL), washed with brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was distilled (85-90°C, 3.3 mmHg) to remove most of the byproduct isopinocampheol. The remaining oil was then purified by chromatography on silica gel to give title compound 33, which was used in the next step.

Crotylation reaction with crotylstannane—crotylstannane preparation

Method 1: *n*BuLi (1.4 m in hexane, 49 mL, 69 mmol, 1.0 equiv) was added to a solution of diisopropylamine (8.0 mL, 79 mmol, 1.15 equiv) in THF (130 mL) at 0 °C. After 10 min, HS*n*Bu₃ (20.0 g, 69 mmol, 1.0 equiv) was added over 10 min to the LDA solution, and after stirring for 20 min at 0 °C, the mixture was cooled to -40 °C and a solution of (*E*)-crotylchloride (6.2 g, 69 mmol, 1.0 equiv) in THF (70 mL) was added by cannula. The mixture was stirred for 15 min at -20 °C and then treated with a saturated aqueous NH₄Cl solution. After extraction with Et₂O, the organic layer was washed with water and brine, dried over MgSO₄, and the solvent was removed under reduced pressure. The residue was then distilled under reduced pressure (154–158 °C, 0.3 mbar) to give a mixture of (*E*/*Z*)-crotylstannane 50:50 (19.0 g, 81%).

Method 2: A solution of *trans*-butene (7 mL) in THF was added by cannula to freshly sublimed potassium *tert*-butoxide (4.0 g, 36 mmol, 1.2 equiv) in THF (70 mL) at -78 °C. *n*BuLi was then added dropwise (1.4 m in hexane, 26.6 mL, 37.2 mmol, 1.25 equiv). After the reaction mixture had been stirred for 1 h at -45 °C, tributyltin chloride (9.7 g, 30 mmol, 1.0 equivalent) was added slowly and the resulting mixture was allowed to warm to 20 °C for 12 h. The reaction was then quenched by the addition of a saturated aqueous NH₄Cl solution. After extraction with Et₂O, the organic layer was washed with water and brine, dried over MgSO₄, and the solvent was removed under reduced pressure. The residue was then distilled under reduced pressure (154–158 °C, 0.3 mbar) to give a mixture of (*E/Z*)-crotylstannane 75:25 (3.25 g, 32 %).

BF₃·OEt₂ (9.0 g, 8.1 mL, 63.8 mmol, 2.2 equiv) was added to a solution of aldehyde **13** (6.0 g, 28.9 mmol, 1.0 equiv) in Et₂O (90 mL) at -100° C. After the reaction mixture had been stirred for 10 min, crotyltributylstannane (13.3 g, 38.6 mmol, 1.1 equiv) was added and the resulting mixture was stirred for 3 h at $-100/-95^{\circ}$ C. After this time, a saturated aqueous NaHCO₃ solution was added and the resulting mixture was extracted with Et₂O. The organic layer was washed with water and brine, dried over MgSO₄, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 90:10 to 70:30) to give the title compound **33** as a yellow oil (5.64 g, 74%) and its diastereomer **34** (d.r. 91:9).

Compound **33**: $[a]_{20}^{D} = -2.7$ (c = 9.6 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.26$ (d, J = 8.9 Hz, 2H), 6.88 (d, J = 8.9 Hz, 2H), 5.63 (ddd, J = 17.2, 10.2, 8.6 Hz, 1H), 5.07 (dd, J = 17.2, 2.0 Hz, 1H), 4.99 (dd, J = 10.2, 2.0 Hz, 1H), 4.46 (d, J = 11.6 Hz, 1H), 4.42 (d, J = 11.6 Hz, 1H), 3.82 (s, 3H), 3.55–3.45 (m, 2H), 3.43 (dd, J = 9.1, 5.0 Hz, 1H), 2.70 (d, J = 3.0 Hz, 1H; OH), 2.23 (dqd, J = 8.6, 7.3, 6.6 Hz, 1H), 1.92–1.79 (m, 1H), 1.10 (d, J = 6.6 Hz, 3H; CH₃), 0.96 ppm (d, J = 6.6 Hz, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 159.2$ (C), 141.3 (CH), 130.3 (C), 129.2 (2CH), 114.5 (CH₂), 113.8 (2CH), 75.4 (CH), 73.1 (CH₂), 72.0 (CH₂), 55.3 (CH₃), 42.1 (CH), 35.4 (CH), 17.2 (CH₃), 9.8 ppm (CH₃); IR (film): $\tilde{\nu} = 2960$, 2907, 2870, 1612, 1513, 1462, 1365, 1302, 1210, 1173, 1093, 1037, 982, 914 cm⁻¹; MS (GC, CI, CH₄): m/z: 293 $[M+29]^+$, 264, 246, 190, 161, 149, 137, 121, 109, 69, 57.

Compound **34**: ¹H NMR (400.0 MHz, CDCl₃): δ = 7.26 (d, *J* = 8.9 Hz, 2H), 6.88 (d, *J* = 8.9 Hz, 2H), 5.86 (ddd, *J* = 17.9, 10.1, 7.3 Hz, 1H), 5.07 (dd, *J* = 17.9, 1.4 Hz, 1H), 4.99 (dd, *J* = 10.1, 1.4 Hz, 1H), 4.46 (d, *J* = 11.6 Hz, 1H), 4.42 (d, *J* = 11.6 Hz, 1H), 3.82 (s, 3H), 3.67–3.60 (m, 1H), 3.55–3.45 (m, 2H), 2.70 (d, *J* = 3.0 Hz, 1H; OH), 2.25–2.22 (m, 1H), 1.92–1.79 (m, 1H), 1.04 (d, *J* = 6.9 Hz, 3H; CH₃), 0.96 ppm (d, *J* = 6.6 Hz, 3H; CH₃).

(3R,4S,5S)-4-(tert-Butyldimethylsilyloxy)-3,5-dimethyl-6-(4-methoxybenzyloxy)hex-1-ene (35): 2,6-Lutidine (6.9 g, 7.5 mL, 64 mmol, 3.0 equiv) and TBSOTf (8.5 g, 7.35 mL, 32 mmol, 1.5 equiv) were added to a solution of alcohol 33 (5.64 g, 21.3 mmol, 1.0 equiv) in dried CH₂Cl₂ (90 mL) at 0°C. After the reaction mixture had been stirred for 2 h at 20°C, it was partitioned between Et2O and a saturated aqueous solution of NH₄Cl and extracted with Et₂O. The organic layer was washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 98:2 to 95:5) to give the title compound 35 (6.30 g, 78%) as a yellow oil. $[\alpha]_D^{20} = +1.3$ (c=1.5 in CHCl₃); ¹H NMR $(400.0 \text{ MHz}, \text{CDCl}_3): \delta = 7.25 \text{ (d}, J = 8.6 \text{ Hz}, 2 \text{ H}), 6.87 \text{ (d}, J = 8.6 \text{ Hz}, 2 \text{ H}),$ 5.79 (ddd, J=17.6, 10.2, 7.8 Hz, 1 H), 4.96 (dd, J=17.6, 1.7 Hz, 1 H), 4.92 (dd, J=10.2, 1.7 Hz, 1 H), 4.42 (d, J=11.6 Hz, 1 H), 4.38 (d, J=11.6 Hz, 1H), 3.81 (s, 3H), 3.63 (dd, J=6.9, 2.3 Hz, 1H), 3.35 (dd, J=8.9, 7.9 Hz, 1 H), 3.19 (dd, J=8.9, 6.3 Hz, 1 H), 2.34 (dqd, J=7.8, 6.9, 6.9 Hz, 1 H), 1.97 (dqdd, J = 7.9, 6.6, 6.3, 2.3 Hz, 1 H), 0.99 (d, J = 6.9 Hz, 3 H; CH₃), 0.89 (s, 9H; 3CH₃), 0.85 (d, J=6.6 Hz, 3H; CH₃), 0.04 (s, 3H; CH₃), 0.02 ppm (s, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 159.0$ (C), 142.0 (CH), 130.8 (C), 129.1 (2CH), 113.6 (2CH), 113.4 (CH₂), 75.5 (CH), 73.4 (CH₂), 72.4 (CH₂), 55.2 (CH₃), 42.6 (CH), 36.5 (CH), 26.1 (3CH₃), 18.4 (C), 16.9 (CH₃), 11.0 (CH₃), -3.6 ppm (2CH₃); IR (film): $\tilde{\nu} = 2959, 2855, 1653, 1615, 1513, 1462, 1360, 1301, 1246, 1172, 1103, 1039,$ 911, 836 cm⁻¹; MS (GC, CI, CH₄): m/z: 407 [M+29]⁺, 377, 363, 323, 271, 241, 199, 187, 161, 145, 121, 109, 89, 75; elemental analysis calcd (%) for C22H38O3Si: C 69.79, H 10.12; found: C 70.16, H 10.35.

(2S,3S,4R)-2,4-Dimethyl-hex-5-ene-1,3-diol (36): At 0°C, a solution of compound 33 (100 mg, 0.38 mmol, 1.0 equiv) in CH₂Cl₂ (3 mL) was treated with water (0.15 mL) and DDQ (72 mg, 0.317 mmol, 1.2 equiv). The mixture was stirred for 10 min at 0°C, warmed to 20°C and then stirred for an additional 30 min. The mixture was quenched with a saturated aqueous NaHCO3 solution, diluted with CH2Cl2, and washed with water and brine. The combined organic layers were dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 60:40 then 50:50) to give the title compound 36 as white crystals (41 mg, 75%). RN: 108867-45-4; $[\alpha]_{D}^{20} = +40$ (c=1.5 in CHCl₃); ¹H NMR $(270.0 \text{ MHz}, \text{ CDCl}_3): \delta = 6.03-5.49 \text{ (m, 1 H)}, 5.03-4.89 \text{ (m, 2 H)}, 3.69-3.57$ (m, 2H), 3.51 (dd, J=9.2, 2.3 Hz, 1H), 2.49 (brs, 1H; OH), 2.28-2.19 (m, 1 H), 1.79–1.73 (m, 1 H), 1.03 (d, J = 6.6 Hz, 3 H; CH₃), 0.87 ppm (d, J =6.9 Hz, 3H; CH₃); ¹³C NMR (67.5 MHz, CDCl₃): $\delta = 140.8$ (CH), 114.7 (CH₂), 76.5 (CH), 67.8 (CH₂), 42.2 (CH), 36.5 (CH), 17.1 (CH₃), 8.9 ppm (CH₃); IR (film): $\tilde{\nu}$ =3464, 2957, 1655, 1264, 1105, 1094, 1023, 1016, 798 cm⁻¹; MS (GC, CI, CH₄): m/z: 127 [MH-H₂O]⁺, 125, 109, 97, 89, 83, 71, 57; elemental analysis calcd (%) for C₈H₁₆O₂: C 66.63, H 11.18; found C 66.46, H 11.07.

(2S,3R,4S)-3-(*tert*-Butyldimethylsilyloxy)-2,4-dimethyl-5-(4-methoxybenzyloxy)pentanal (37): A stream of ozone was bubbled through a solution of compound 35 (6.3 g, 16.6 mmol, 1.0 equiv), pyridine (4 mL, 50 mmol, 3.0 equiv), and a small amount of Sudan III in MeOH/CH₂Cl₂ (1:1, 200 mL) at -78 °C until the pink solution became colorless. The solution was treated with DMS (36.5 mL, 498 mmol, 30 equiv) and was then warmed to 20 °C for 12 h. The solution was concentrated under reduced pressure, diluted with Et₂O, washed with water (3×), dried over MgSO₄, and the solvent was removed under reduced pressure. The crude 37 produced (6.0 g, 95%) was used in the next step without further purification. ¹H NMR (400.0 MHz, CDCl₃): δ =9.87 (d, *J*=0.8 Hz, 1H), 7.25 (d, *J*= 8.6 Hz, 2H), 6.89 (d, *J*=8.6 Hz, 2H), 4.41 (d, *J*=11.6 Hz, 1H), 4.36 (d, *J*=11.6 Hz, 1H), 4.24 (dd, *J*=5.1, 3.4 Hz, 1H), 3.82 (s, 3H), 3.37 (dd, *J*= 9.0, 7.3 Hz, 1H), 3.24 (dd, *J*=9.0, 5.9 Hz, 1H), 2.54 (qdd, *J*=6.9, 5.1, 0.8 Hz, 1H), 1.94 (dqdd, *J*=7.3, 6.9, 5.9, 3.4 Hz, 1H), 1.05 (d, *J*=6.9 Hz,

Chem. Eur. J. 2008, 14, 11092-11112

© 2008 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

3H; CH_3), 0.91 (s, 9H; $3CH_3$), 0.86 (d, J=6.9 Hz, 3H; CH_3), 0.07 (s, 3H; CH₃), 0.05 ppm (s, 3H; CH₃); 13 C NMR (100.5 MHz, CDCl₃): $\delta = 205.5$ (C), 129.2 (2CH), 113.7 (2CH), 72.6 (CH₂), 72.5 (CH₂), 72.3 (CH), 55.2 (CH₃), 51.3 (CH), 37.0 (CH), 25.9 (3CH₃), 18.2 (C), 12.1 (CH₃), 9.3 (CH₃), -4.0 ppm (2CH₃), one C atom was not detected; MS (GC, EI): *m*/*z*: 323 [*M*-*t*Bu]⁺, 251, 199, 187, 137, 121, 107, 98, 89, 73, 59; elemental analysis calcd (%) for $C_{21}H_{36}O_4Si$: C 66.67, H 9.53; found: C 66.60, H 9.67.

(1Z,3S,4S,5R,6R,7S)-6-(tert-Butyldimethylsilyloxy)-1-[(N,N-diisopropyl)carbamoyloxy]-8-(4-methoxybenzyloxy)-3,5,7-trimethyl oct-1-en-4-ol (38)

(E)-Crotyldiisopropylcarbamate preparation: A solution of 2-buten-1-ol (E/Z mixture, 18.4 g, 255 mmol, 1.0 equiv) in THF (50 mL) was added to a suspension of NaH (60%, 12.2 g, 306 mmol, 1.2 equiv) in THF (50 mL) at 0°C over 20 min. After the reaction mixture had been stirred for 20 min at 20 °C, it was cooled to 0 °C and a solution of N,N-diisopropylcarbamoyl chloride (50 g, 306 mmol, 1.2 equiv) in THF (70 mL) was slowly added. The reaction mixture was stirred for 3 h at 20°C, and was then poured into an aqueous HCl 1N solution (150 mL) and extracted with Et2O. The combined organic layers were washed with a saturated aqueous NaHCO3 solution and brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was distilled (64°C, 0.1 mbar) to give (E)-crotyldiisopropylcarbamate (50.0 g, 98%).

Crotylation reaction with optically active crotyltitane: A solution of nBuLi (1.4 m in hexanes, 25.5 mL, 35.5 mmol, 2.1 equiv) was added to a quickstirred solution of the (E)-crotyl diisopropylcarbamate (6.6 g, 33.2 mmol), 2.0 equiv) and (-)-sparteine (8.01 g, 34.2 mmol, 2.1 equiv), in pentane (30 mL) and cyclohexane (5 mL) at -78 °C, and after 10 min white crystals appeared. After 3 h of crystallization at -78°C, a pre-cooled (-40°C) solution of Ti(OiPr)4 (29.5 mL, 99.6 mmol, 6.0 equiv) in pentane (30 mL) was quickly added by cannula to the reaction mixture of lithio carbamate, which became limpid and turned orange. After 1 h at -78 °C, a solution of aldehyde 37 (6.32 g, 16.6 mmol, 1.0 equiv) in pentane (10 mL) was slowly added to the orange solution, and the mixture was stirred for 3 h at -78 °C. The solution was then poured into a mixture of Et₂O (250 mL) and aqueous HCl (0.5 N, 250 mL). After extraction with Et_2O_1 , the organic layer was washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cvclohexane/ethyl acetate 90:10 to 80:20) to give the title compound 38 (7.1 g, 77%) as a single diastereomer.

4.6-Acetonide derivative of 38: A solution of tetrabutylammonium fluoride (1.0 m in THF, 345 µL, 345 µmol, 2.0 equiv) was added to a solution of compound 38 (100 mg, 172 µmol, 1.0 equiv) in THF (0.6 mL) at 20 °C. The reaction mixture was stirred for 4 h, quenched by the addition of water, and extracted with ethyl acetate. The organic layer was washed with brine, dried over MgSO4, filtered, and concentrated. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 80:20 to 70:30) to give the corresponding (2S,3R,4R,5S,6S)-3,5-dihydroxy-8-{[(N,N-diisopropyl)carbamoyloxy]-1-(4-methoxy-benzyloxy)-2,4,6-tri-

methyl oct-7-ene (73 mg, 91%). $[a]_{\rm D}^{20} = +0.8$ (c=0.75 in CHCl₃); ¹H NMR (270.0 MHz, CDCl₃): $\delta = 7.18$ (d, J = 8.6 Hz, 2H), 7.08 (d, J =6.3 Hz, 1 H), 6.80 (d, J=8.6 Hz, 2 H), 4.53 (dd, J=9.9, 6.3 Hz, 1 H), 4.35 (s, 2H), 4.13-3.97 (br s, 1H), 3.73 (s, 3H; CH₃), 3.72-3.60 (br s, 1H), 3.66-3.63 (m, 1H), 3.35-3.33 (m, 2H), 3.32-3.28 (m, 1H), 2.88-2.73 (m, 1H), 2.00-1.85 (m, 1H; OH), 1.94-1.85 (m, 2H), 1.57-1.48 (m, 1H; OH), 1.21–1.16 (m, 12H; 4CH₃), 0.98 (d, J = 6.9 Hz, 3H; CH₃), 0.88 (d, J =6.9 Hz, 3H; CH₃), 0.85 ppm (d, J = 6.9 Hz, 3H; CH₃); ¹³C NMR (67.5 MHz, CDCl₃): δ=158.9 (C), 152.5 (C), 137.2 (CH), 130.2 (C), 129.1 (2CH), 113.7 (2CH), 112.9 (CH), 78.5 (CH), 77.9 (CH), 73.9 (CH₂), 72.3 (CH₂), 55.2 (CH₃), 45.7 (2CH), 36.1 (CH), 35.6 (CH), 34.2 (CH), 21.0 (4CH₃), 16.2 (CH₃), 13.1 (CH₃), 5.8 ppm (CH₃); IR (film): $\tilde{\nu}$ = 2969, 2934, 2873, 1708, 1692, 1513, 1441, 1370, 1305, 1247, 1210, 1135, 1063, 971 cm⁻¹. 2,2-Dimethoxypropane (210 µL, 1.67 mmol, 12 equiv) and PPTS (10%) were added to a solution of the above diol (65 mg, 0.14 mmol, 1.0 equiv) in CH2Cl2 at 20°C. The reaction mixture was stirred for 2 h. Then the so-

lution was cooled to 0°C and triethylamine (1 mL) was added. The reaction mixture was concentrated under reduced pressure. The crude residue

was purified by chromatography on silica gel (cyclohexane/ethyl acetate 70:30) to give the title product (59 mg, 84%). $[a]_{D}^{20} = +29.6$ (c=2.9 in CHCl₃); ¹H NMR (270.0 MHz, CDCl₃): $\delta = 7.18$ (d, J = 8.6 Hz, 2 H), 6.93 (d, J=6.3 Hz, 1 H), 6.79 (d, J=8.6 Hz, 2 H), 4.49 (dd, J=8.9, 6.3 Hz, 1 H), 4.33 (s, 2H), 4.16-4.02 (brs, 1H), 3.75-3.58 (brs, 1H), 3.72 (s, 3H), 3.51 (dd, J=9.6, 2.0 Hz, 1H), 3.36 (dd, J=9.6, 2.0 Hz, 1H), 3.24 (d, J=4.6 Hz)2H), 2.78-2.63 (m, 1H), 1.83-1.73 (m, 1H), 1.50-1.42 (m, 1H), 1.27-1.16 (m, 12H; 4CH₃), 1.27 (s, 3H; CH₃), 1.25 (s, 3H; CH₃), 0.96 (d, J =6.6 Hz, 3 H), 0.81 (d, J = 6.9 Hz, 3 H), 0.75 ppm (d, J = 6.9 Hz, 3 H); ^{13}C NMR (67.5 MHz, CDCl₃): $\delta\!=\!159.0$ (C), 153.2 (C), 134.9 (CH), 130.5 (C), 129.0 (2CH), 114.6 (CH), 113.6 (2CH), 98.8 (C), 77.5 (CH), 75.7 (CH), 72.7 (CH₂), 71.1 (CH₂), 55.1 (CH₃), 46.1 (2CH), 34.9 (CH, C-16), 31.8 (CH), 30.9 (CH), 29.8 (CH₃), 21.0 (4 CH₃), 19.4 (CH₃), 15.8 (CH₃), 14.8 (CH₃), 4.9 ppm (CH₃); IR (film): $\tilde{\nu}$ =2968, 2936, 1708, 1513, 1462, 1439, 1377, 1310, 1288, 1248, 1135, 1057, 1012 cm⁻¹; MS (GC, CI, CH₄): m/z: 534 $[M+29]^+$, 506 $[M+1]^+$; elemental analysis calcd (%) for C₂₉H₄₇NO₆: C 68.88, H 9.37, N 2.77; found: C 68.75, H 9.32, N 2.85.

2,4-Acetonide of 38: A stream of ozone was bubbled into a cooled (-78°C) solution of the carbamate 38 (100 mg, 172 µmol, 1.0 equiv), pyridine (40 µL), and a small amount of Sudan III in MeOH/CH₂Cl₂ (1:1, 10 mL) until the pink solution became colorless. Sodium borohydride (39 mg, 1.03 mmol, 6.0 equiv) was added at -78 °C, the mixture was allowed to warm to 20°C and was then stirred for 6 h. After this time, the reaction mixture was cooled, treated with a saturated aqueous solution of NH₄Cl and extracted with Et₂O. The organic layers were washed with brine, dried over MgSO4, filtered, and concentrated. The residue was purified by flash chromatography on silica gel (cyclohexane/ethyl acetate 70:30) to give (2S.3R.4R.5S.6S)-3-(tert-butyldimethylsilyloxy)-5.7-dihydroxy-1-(4-methoxy-benzyloxy)-2,4,6-trimethyl heptane (60 mg, 79%). $[a]_{D}^{20} = -0.73$ (c = 1.1 in CHCl₃); ¹H NMR (270.0 MHz, CDCl₃): $\delta = 7.22$ -7.18 (m, 2H), 6.85-6.79 (m, 2H), 4.36 (s, 2H), 3.75 (s, 3H), 3.86-3.67 (m, 3H), 3.80-3.70 (m, 1H; OH), 3.60-3.50 (m, 1H; OH), 3.50 (t, J= 10.6 Hz, 1 H), 3.40-3.31 (m, 1 H), 3.22-3.16 (m, 1 H), 2.06-2.01 (m, 1 H), 1.90-1.65 (m, 2H), 0.83 (s, 9H; 3CH₃), 0.87-0.70 (m, 9H; 3CH₃), 0.03 ppm (s, 6H; 2CH₃); ¹³C NMR (67.5 MHz, CDCl₃): $\delta = 159.0$ (C), 130.7 (C), 129.1 (2 CH), 113.6 (2 CH), 77.6 (CH), 73.9 (CH₂), 73.6 (CH), 72.4 (CH₂), 68.4 (CH₂), 55.2 (CH₃), 38.5 (CH), 35.8 (CH), 32.8 (CH), 26.1 (3 CH₃), 18.4 (C), 12.1 (CH₃), 10.5 (CH₃), 9.6 (CH₃), -3.6 ppm (2 CH₃); IR (film): $\tilde{\nu} = 2958$, 2933, 2856, 1653, 1618, 1559, 1514, 1423, 1344, 1270, 1103, 1037, 837 cm⁻¹

2,2-Dimethoxypropane (180 µL, 1.50 mmol, 12 equiv) and PPTS (10 mol %) were added to a solution of the preceding diol (55 mg, 125 $\mu mol,$ 1.0 equiv) in CH₂Cl₂ (3 mL) at 20 °C. The reaction mixture was stirred for 2 h and then the solution was cooled to 0°C and triethylamine (1 mL) was added. The reaction mixture was concentrated under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 90:10 to 60:40) to give the title product, the 2,4acetonide of **38**, (46 mg, 78 %) and the starting diol (22 mg). $[\alpha]_{D}^{20} = +12.2$ $(c=1.2 \text{ in CHCl}_3)$; ¹H NMR (270.0 MHz, CDCl₃): $\delta = 7.21$ (d, J = 8.6 Hz, 2H), 6.82 (d, J = 8.6 Hz, 2H), 4.36 (s, 2H), 3.76 (s, 3H), 3.72 (dd, J = 7.6, 1.3 Hz, 1H), 3.63–3.49 (m, 2H), 3.43 (t, J=11.3 Hz, 1H), 3.36–3.30 (m, 1H), 3.22-3.16 (m, 1H), 1.86-1.70 (m, 3H), 1.38-1.19 (m, 6H; 2CH₃), 0.87 (d, *J*=5.3 Hz, 3 H; CH₃), 0.84 (s, 9H; 3 CH₃), 0.80 (d, *J*=6.9 Hz, 3 H; CH₃), 0.63 (d, J = 6.6 Hz, 3H; CH₃), 0.00 ppm (s, 6H; 2CH₃); ¹³C NMR (67.5 MHz, CDCl₃): δ=159.5 (C), 131.5 (C), 129.5 (2CH), 114.5 (2CH), 98.5 (C), 75.2 (CH), 74.2 (CH₂), 73.6 (CH), 72.8 (CH₂), 67.0 (CH₂), 55.8 (CH₃), 38.2 (CH), 36.8 (CH), 31.3 (CH), 29.3 (CH₃), 26.7 (3CH₃), 19.4 (CH₃), 19.0 (C), 12.8 (CH₃), 11.0 (CH₃), 10.5 (CH₃), -3.6 ppm (2 CH₃); IR (film): $\tilde{\nu}$ = 3256, 1684, 1653, 1550, 1501, 1445, 1359, 1248, 1101, 1032, 846 cm⁻¹; MS (GC, CI, CH₄): m/z: 481 [M+H]⁺; elemental analysis calcd (%) for C₂₇H₄₈O₅Si: C 67.45, H 10.06; found: C 67.32, H 10.24.

(1Z,3S,4S,5R,6S,7S)-6-(tert-Butyldimethylsilyloxy)-1-[(N,N-diisopropyl)carbamoyloxy]-8-(4-methoxybenzyloxy)-4-(triethylsilyloxy)-3,5,7-trimethyl oct-1-ene (39): 2,6-Lutidine (0.9 mL, 7.50 mmol, 4.0 equiv) and TESOTf (0.9 mL, 3.76 mmol, 2.5 equiv) were added to a solution of 38 (1.09 g, 1.88 mmol, 1.0 equiv) in dried CH₂Cl₂ (15 mL) at 0°C. After the reaction mixture had been stirred for 2 h at 20°C, it was partitioned between Et₂O and a saturated aqueous solution of NH₄Cl and was extract-

```
11104 -
```

ed with Et2O. The organic layer was washed with brine, dried over magnesium sulfate, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 95:5 to 90:10) to give the title compound 39 (985 mg, 76%) of as a colorless oil. $[\alpha]_{D}^{20} = +14.4$ (c=0.8 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.22$ (d, J = 8.6 Hz, 2H), 6.90 (d, J =6.6 Hz, 1 H), 6.84 (d, J=8.6 Hz, 2 H), 5.01 (dd, J=9.2, 6.6 Hz, 1 H), 4.38 (d, J=11.1 Hz, 1 H), 4.35 (d, J=11.1 Hz, 1 H), 4.11-3.95 (br s, 1 H), 3.82-3.69 (brs, 1H), 3.78 (s, 3H), 3.67 (dd, J=6.3, 2.3 Hz, 1H), 3.53 (dd, J= 6.3, 5.3 Hz, 1 H), 3.37 (dd, J=8.6, 5.6 Hz, 1 H), 3.16 (t, J=8.6 Hz, 1 H), 2.78-2.72 (m, 1H), 1.87-1.80 (m, 1H), 1.74-1.67 (m, 1H), 1.28-1.14 (m, 12H; 4CH₃), 1.01 (d, J=7.2 Hz, 3H; CH₃), 0.97-0.87 (m, 12H; 4CH₃), 0.85 (s, 9H; 3CH₃), 0.80 (d, *J*=7.2 Hz, 3H; CH₃), 0.60 (q, *J*=7.6 Hz, 6H; 3CH₂), 0.02 (s, 3H; CH₃), 0.01 ppm (s, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 159.0$ (C), 153.0 (C), 133.7 (CH), 130.7 (C), 129.2 (2CH), 113.6 (2 CH), 112.7 (CH), 77.0 (CH), 73.9 (CH), 73.6 (CH₂), 72.6 (CH₂), 55.2 (CH₃), 45.4 (2 CH), 42.9 (CH), 38.7 (CH), 32.4 (CH), 26.1 (3 CH₃), 21.5 (4CH₃), 20.3 (CH₃), 18.4 (C), 12.2 (CH₃), 11.1 (CH₃), 7.0 (3CH₃), 5.5 (3 CH₂), 3.2 ppm (2 CH₃); IR (film): $\tilde{\nu}$ =2966, 2937, 2877, 1710, 1514, 1460, 1439, 1306, 1249, 1060, 1042, 1005, 837 $\rm cm^{-1};$ elemental analysis calcd (%) for C38H71NO6Si2: C 65.75, H 10.31, N 2.02; found: C 65.72, H 10.27, N 1.97.

(3Z,5S,6S,7R,8S,9S)-8-(*tert*-Butyldimethylsilyloxy)-10-(4-methoxybenzyloxy)-6-(triethylsilyloxy)-5,7,9-trimethyl deca-1,3-diene (40) and (3Z,5S,6S,7R,8S,9S)-6-(*tert*-butyldimethylsilyloxy)-8-(4-methoxybenzyloxy)-6-(triethylsilyloxy)-3,5,7-trimethyl-1-decene (41)

Reaction with vinylmagnesium bromide: A solution of the vinyl carbamate **39** (700 mg, 1.01 mmol, 1.0 equiv) in Et₂O (10 mL, 3 mL rinse) and vinylmagnesium bromide (sol. 1 M in THF, 15.1 mL, 15.1 mmol, 15 equiv) were added to a stirred suspension of [Ni(acac)₂] (26 mg, 0.10 mmol, 0.1 equiv) in Et₂O (10 mL) at 0 °C. The resulting mixture was stirred for 24 h at 0 °C. After this time, a saturated aqueous NH₄Cl solution was added and the solution was extracted with Et₂O. The organic layer was washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 100:0 to 90:10) to give diene **40** (460 mg, 80%) and compound **41** (less than 5% yield).

Reaction with vinyllithium: MeLi (1.6 m in Et₂O, 18 mL, 28.8 mmol, 10 equiv) was added to a solution of tetravinyltin (1.7 g, 7.48 mmol, 2.6 equiv) in Et₂O (12 mL) at 0°C. After the reaction mixture had been stirred for 30 min, it was added by cannula to a stirred solution of the carbamate **39** (2.0 g, 2.88 mmol, 1.0 equiv) and [Ni(acac)₂] (74 mg, 0.288 mmol, 0.1 equiv) in Et₂O (4 mL) at -5/0°C. After the reaction mixture had been stirred for a further 6 h, the same amount of [Ni(acac)₂] was added and the resulting mixture was stirred for another 12 h at 0°C. Then a saturated aqueous NH₄Cl solution was added and the solution was extracted with Et₂O. The organic layer was washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 98:2 to 80:20) to give **40** (1.33 g, 80%).

Compound **40**: $[a]_{D}^{20} = +8.9$ (c=2.1 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.24$ (d, J=8.3 Hz, 2H), 6.86 (d, J=8.3 Hz, 2H), 6.57 (dt, J=16.9, 10.0 Hz, 1H), 5.97 (t, J=10.0 Hz, 1H), 5.48 (t, J=10.0 Hz, 1H), 5.14 (d, J=10.0 Hz, 1H), 5.07 (d, J=16.9 Hz, 1H), 4.40 (d, J=11.4 Hz, 1H), 4.35 (d, J=11.4 Hz, 1H), 3.80 (s, 3H), 3.63 (dd, J=5.5, 3.0 Hz, 1H), 3.51 (t, J=4.0 Hz, 1H), 3.38 (dd, J=16.5, 9.0 Hz, 1H), 3.16 (dd, J=16.5, 9.0 Hz, 1H), 2.82 (m, 1H), 1.99–1.88 (m, 1H), 1.73–1.58 (m, 1H), 1.03–0.85 (m, 27 H; 9 CH₃), 0.59 (q, J=7.8 Hz, 6H; 3 CH₂), 0.04 ppm (s, 6H; 2 CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 159.0$ (C), 135.0 (CH), 136. (2 CH), 73.1 (CH), 72.5 (CH₂), 55.2 (CH₃), 40.2 (CH), 38.0 (CH), 36.5 (CH), 26.2 (3 CH₃), 18.7 (C), 18.5 (CH₃), 11.9 (CH₃), 11.5 (CH₃), 7.2 (3 CH₃), 5.6 (3 CH₂), -3.3 ppm (2 CH₃); IR (film): $\bar{\nu} = 2957$, 2857, 1616, 1490 cm⁻¹; elemental analysis calcd (%) for C₃₃H₆₀O₄Si₂: C 68.69, H 10.48; found: C 68.48, H 10.67.

Compound **41**: ¹H NMR (400.0 MHz, CDCl₃): δ =7.24 (d, *J*=8.7 Hz, 2H), 6.86 (d, *J*=8.7 Hz, 2H), 5.80 (ddd, *J*=15.1, 11.9, 8.2 Hz, 1H), 4.97 (d, *J*=11.9 Hz, 1H), 4.96 (d, *J*=15.1 Hz, 1H), 4.41 (d, *J*=11.9 Hz, 1H),

FULL PAPER

4.35 (d, J = 11.9 Hz, 1 H), 3.80 (s, 3 H), 3.68 (dd, 1 H, J = 5.9, 3.2 Hz), 348 (dd, J = 5.0, 4.8 Hz, 1 H), 3.38 (dd, J = 8.7, 6.4 Hz, 1 H), 3.18 (dd, J = 8.7, 7.3 Hz, 1 H), 2.34 (dqd, J = 8.2, 6.9, 5.0 Hz, 1 H), 1.93 (dqdd, J = 7.3, 6.9, 6.4, 3.2 Hz, 1 H), 1.69 (qdd, J = 6.9, 5.9, 4.8 Hz, 1 H), 0.99 (d, J = 6.9 Hz, 3H; CH₃), 0.93 (t, J = 7.8 Hz, 9H; 3 CH₃), 0.88 (s, 9H; 3 CH₃), 0.86 (d, J = 6.9 Hz, 3H; CH₃), 0.85 (d, J = 6.9 Hz, 3H; CH₃), 0.03 ppm (s, 6H; 2 CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 159.0$ (C), 141.2 (C), 131.2 (CH), 129.3 (2 CH), 114.6 (CH₂), 113.6 (2 CH), 77.0 (CH), 73.3 (CH₂), 73.2 (CH), 72.6 (CH₂), 55.3 (CH₃), 39.6 (CH), 37.9 (CH), 33.5 (CH), 26.2 (3 CH₃), 18.5 (C), 18.5 (CH₃), 11.7 (CH₃), 11.5 (CH₃), 7.2 (3 CH₃), 5.6 (3 CH₂), -3.4 ppm (2 CH₃).

(3Z,5S,6S,7R,8S,9S)-8-(tert-Butyldimethylsilyloxy)-10-hydroxy-6-(triethylsilyloxy)-5,7,9-trimethyl deca-1,3-diene (42): DDQ (117 mg, 0.52 mmol, 1.1 equiv) was added to a solution of compound 40 (249 mg, 0.43 mmol, 1.0 equiv) in CH2Cl2 (6 mL)/H2O (300 µL) at 0°C. The mixture was stirred for 10 min at 0 °C, warmed to 20 °C and was then stirred for an additional 30 min. The mixture was quenched with a saturated aqueous NaHCO3 solution, diluted with Et2O, and washed with water and brine. The combined organic layers were dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 95:5 to 80:20) to give the title compound 42 (142 mg, 72%). $[\alpha]_{\rm D}^{20} = +11.4$ (c = 1.2 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 6.58$ (ddd, J = 16.9, 11.0, 10.5 Hz, 1 H), 6.02 (t, J=11.0 Hz, 1 H), 5.60 (dd, J=11.0, 10.1 Hz, 1 H), 5.22 (d, J = 16.9 Hz, 1 H), 5.13 (d, J = 10.5 Hz, 1 H), 3.68 (t, J = 3.9 Hz, 1H), 3.60–3.56 (m, 1H), 3.51 (dd, J=7.3, 3.9 Hz, 1H), 3.45–3.40 (m, 1H), 2.88 (dqd, J=10.1, 7.3, 3.9 Hz, 1 H), 2.86 (m, 1 H), 1.75 (dd, J=5.5, 5.0 Hz, 1H; OH), 1.64 (dqd, J=7.3, 6.9, 3.9 Hz, 1H), 1.04 (d, J=7.3 Hz, 3H; CH₃), 0.98 (t, J=7.8 Hz, 9H; 3CH₃), 0.94 (d, J=6.9 Hz, 3H; CH₃), 0.85 (s, 9H; 3CH₃), 0.83 (d, *J*=6.9 Hz, 3H; CH₃), 0.65 (q, *J*=7.8 Hz, 6H; 3CH₂), 0.11 ppm (s, 3H; CH₃), 0.08 ppm (s, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 134.6$ (CH), 132.5 (CH), 129.2 (CH), 117.7 (CH₂), 78.0 (CH), 73.7 (CH), 66.2 (CH₂), 40.8 (CH), 40.2 (CH), 35.9 (CH), 26.2 (3CH₃), 19.2 (CH₃), 18.5 (C), 12.5 (CH₃), 11.9 (CH₃), 7.4 $(3 CH_3)$, 5.8 $(3 CH_2)$, -3.5 (CH_3) , -3.7 ppm (CH_3) ; IR (film): $\tilde{v} = 3405$, 2956, 2930, 2878, 2858, 1461, 1252, 1096, 1076, 1006, 837, 772, 737 cm⁻¹; elemental analysis calcd (%) for C25H52O3Si2: C 65.73, H 11.47; found: C 65.56, H 11.62.

(3Z,5S,6S,7R,8S,9S)-8-(tert-Butyldimethylsilyloxy)-10-iodo-6-(triethylsilyloxy)-5,7,9-trimethyl deca-1,3-diene (43): A solution of iodine (167 mg, 0.66 mmol, 1.5 equiv) in Et₂O (2 mL) was added slowly to a solution of compound 42 (200 mg, 0.44 mmol, 1.0 equiv), triphenylphosphine (PPh₃, 218 mg, 0.83 mmol, 1.9 equiv), and imidazole (57 mg, 0.83 mmol, 1.9 equiv) in a benzene/Et2O mixture (1.5 mL:3 mL) at 0 °C. After the reaction mixture had been stirred for 2 h at 20°C, a 1 M Na₂S₂O₃ solution was added and the reaction mixture was extracted with Et2O. The organic layers were washed with, brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/Et₂O 100:0 to 80:20) to give the title compound **43** (197 mg, 79%). $[\alpha]_D^{20} = +11.3$ (c=0.68 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 6.58$ (ddd, J = 16.9, 11.0, 10.1 Hz, 1 H), 6.02 (t, J=11.0 Hz, 1 H), 5.55 (dd, J=11.0, 10.1 Hz, 1 H), 5.21 (d, J= 16.9 Hz, 1 H), 5.12 (d, J=10.1 Hz, 1 H), 3.58 (t, J=4.6 Hz, 1 H), 3.50 (dd, J=6.4, 3.7 Hz, 1H), 3.32 (dd, J=9.6, 4.6 Hz, 1H), 2.98 (dd, J=9.6, 9.2 Hz, 1 H), 2.84 (dqd, J=10.1, 6.9, 4.6 Hz, 1 H), 1.97 (dqdm, J=9.2, 6.9, 4.6 Hz, 1H), 1.65 (qdm, J=6.9, 4.6 Hz, 1H), 1.01 (d, J=6.9 Hz, 3H; CH₃), 0.98 (t, J=7.8 Hz, 9H; 3CH₃), 0.98 (d, J=6.9 Hz, 3H; CH₃), 0.91 (s, 9H; 3CH₃), 0.90 (d, J=6.9 Hz, 3H; CH₃), 0.64 (q, J=7.8 Hz, 6H; 3 CH₂), 0.10 (s, 3H; CH₃), 0.09 ppm (s, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ = 134.1 (CH), 132.5 (CH), 129.0 (CH), 117.4 (CH₂), 77.1 (CH), 75.0 (CH), 41.9 (CH), 40.4 (CH), 35.8 (CH), 25.9 (3 CH₃), 18.8 (CH₃), 18.4 (C), 15.0 (CH₃), 13.4 (CH₂), 11.9 (CH₃), 7.0 (3CH₃), 5.5 (3CH₂), $-3.8 \text{ ppm} (2 \text{ CH}_3)$; IR (film): $\tilde{\nu} = 2950, 2930, 2877, 2857, 1461, 1097, 1028,$ 1005, 836 cm⁻¹; elemental analysis calcd (%) for C₂₅H₅₁IO₂Si₂: C 52.98, H 9.07; found: C 53.17, H 9.25.

(1Z,3S,4S,5S)-6-(tert-Butyldimethylsilyloxy)-1-[(N,N-diisopropyl)carba-

moyloxy]-3,5-dimethyl-hex-1-en-4-ol (51): A solution of nBuLi (1.37 M in hexanes, 30.6 mL, 41.9 mmol, 2.1 equiv) was added to a quick-stirred so-

lution of the (E)-crotyl diisopropylcarbamate (7.80 g, 39.2 mmol, 2.0 equiv) and (-)-sparteine (9.50 g, 40.4 mmol, 2.1 equiv) in pentane (36 mL) and cyclohexane (6 mL) at -78 °C. After 10 min, white crystals appeared, and after 3 h of crystallization at -78°C, a pre-cooled (-40°C) solution of Ti(OiPr)₄ (34.7 mL, 117.6 mmol, 6.0 equiv) in pentane (30 mL) was quickly added by cannula to the reaction mixture of lithio carbamate, which became limpid and turned orange. After 1 h at -78 °C, aldehyde 50 (3.96 g, 19.6 mmol, 1.0 equiv) in pentane (10 mL) was slowly added to the orange solution, and the mixture was stirred for 3 h at -78°C. The solution was then poured into a mixture of Et₂O (250 mL)/ aqueous HCl (0.5 N, 250 mL). After extraction with Et₂O, the organic layer was washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 95:5 to 70:30) to give the title compound 51 (4.30 g, 58%) and its 3,4-bis-epi diastereomer (de = 95:5; de = diastereometric excess).

Compound 51: $[\alpha]_D^{20} = +20.1$ (c=1.0 in CHCl₃); H RMN (400.0 MHz, CDCl₃): $\delta = 7.13$ (d, J = 6.4 Hz, 1H), 4.73 (dd, J = 9.6, 6.4 Hz, 1H), 4.22– 4.11 (brs, 1H), 3.82–3.70 (brs, 1H), 3.71 (d, J=4.6 Hz, 2H), 3.54 (dt, J= 8.2, 2.3 Hz, 1 H), 2.86 (ddq, J = 9.6, 8.2, 6.9 Hz, 1 H), 2.69 (d, J = 2.3 Hz, 1H; OH), 1.84 (qtd, J=6.9, 4.6, 2.3 Hz, 1H), 1.28–1.10 (m, 12H; 4CH₃), 0.97 (d, J=6.9 Hz, 3 H; CH₃), 0.94 (d, J=6.9 Hz, 3 H; CH₃), 0.90 (s, 9 H; $3 CH_3$, 0.08 ppm (s, 6 H; 2 CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 153.0$ (C), 135.9 (CH), 113.9 (CH), 77.0 (CH), 68.0 (CH₂), 47.0 (CH), 45.2 (CH), 36.6 (CH), 34.0 (CH), 26.0 (3 CH₃), 21.6 (2 CH₃), 20.5 (2 CH₃), 18.3 (C), 17.4 (CH₃), 9.4 (CH₃), -5.50 ppm (2 CH₃); IR (film): \tilde{v} = 3493, 2959, 2930, 2858, 1708, 1472, 1439, 1370, 1304, 1290, 1256, 1211, 1156, 1134, 1092, 1061, 837 cm⁻¹; MS (GC, EI): m/z: 344 [M-tBu]⁺, 326, 302, 228, 212, 199, 184, 145, 128, 115, 86, 73; elemental analysis calcd (%) for C21H43NO4Si: C 62.80, H 10.79, N 3.49; found: C 62.61, H 10.87, N 3.35. 3,4-bis-epi isomer of 51: ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.07$ (d, J =6.4 Hz, 1 H), 4.97 (dd, J=9.6, 6.4 Hz, 1 H), 4.25-4.05 (br s, 1 H), 4.20 (s, 1H; OH), 3.85-3.67 (brs, 1H), 3.75 (dd, J=10.1, 4.1 Hz, 1H), 3.60 (dd, J=10.1, 9.4 Hz, 1 H), 3.47 (dd, J=8.7, 2.3 Hz, 1 H), 2.85 (ddq, J=9.6, 8.7, 6.9 Hz, 1 H), 1.77 (dqdd, J=9.4, 6.9, 4.1, 2.3 Hz, 1 H), 1.33-1.19 (m, 12 H; 4 CH₃), 1.14 (d, J = 6.9 Hz, 3H; CH₃), 0.90 (s, 9H; 3 CH₃), 0.76 (d, J =6.9 Hz, 3H; CH₃), 0.09 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 153.5$ (C), 135.1 (CH), 111.5 (CH), 80.5 (CH), 69.5 (CH₂), 46.9 (CH), 45.5 (CH), 38.1 (CH), 33.5 (CH), 25.9 (3CH₃), 21.6 (2CH₃), 20.5 (2 CH₃), 18.3 (C), 18.2 (CH₃), 13.2 (CH₃), -5.5 ppm (2 CH₃).

(2S,3S,4S)-1-(*tert*-Butyldimethylsilyloxy)-2,4-dimethyl-hex-5-en-3-ol (52): A solution of isopropylmagnesium chloride in THF (1.55 M, 8.0 mL, 12.4 mmol, 10 equiv) was slowly added to a solution of carbamate 51 (500 mg, 1.24 mmol, 1.0 equiv) and [Ni(acac)₂] (32 mg, 0.12 mmol, 10 mol%) in THF (5 mL) at 20°C. After the reaction mixture had been stirred for 12 h, a saturated aqueous NH4Cl solution was added, and the reaction mixture was extracted with Et₂O. The combined organic layers were washed with water and brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 98:2 to 85:15) to give the title compound 52 (162 mg, 50%). RN: 106296-58-6; ¹H NMR (400.0 MHz, CDCl₃): $\delta = 5.84$ (ddd, J = 16.5, 10.5, 8.2 Hz, 1 H), 5.12 (dd, J=16.5, 1.8 Hz, 1 H), 5.09 (dd, J=10.5, 1.8 Hz, 1 H), 3.74 (dd, J=9.8, 4.1 Hz, 1H), 3.71 (dd, J=9.8, 5.0 Hz, 1H), 3.55 (ddd, J=8.7, 2.3, 11.8 Hz, 1H), 2.89 (d, J=2.3 Hz, 1H; OH), 2.27 (ddq, J=8.7, 8.2, 6.9 Hz, 1H), 1.80 (qddd, J=6.9, 5.0, 4.1, 1.8 Hz, 1H), 0.95 (d, J=6.9 Hz, 6H; 2CH₃), 0.90 (s, 9H; 3CH₃), 0.07 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 142.3$ (CH), 115.3 (CH₂), 77.2 (CH), 68.8 (CH₂), 42.0 (CH), 36.1 (CH), 26.0 (3CH₃), 18.4 (C), 16.9, 9.5 (2CH₃), -5.4 (CH₃), -5.5 ppm (CH₃).

(2E,4S,5S,6S)-N-Methyl-N-methoxy-7-(*tert*-butyldimethylsilyloxy)-4,6-dimethyl-5-hydroxy-hept-2-enamide (53): A stream of ozone was bubbled into a cooled (-78 °C) solution of 51 (1.05 g, 2.61 mmol, 1.0 equiv) and a small amount of Sudan III in CH₂Cl₂ (50 mL) until the pink solution became colorless. The solution was treated with triphenylphosphine (686 mg, 2.61 mmol, 1.0 equiv) and the mixture was allowed to warm to 20 °C for 2 h. The solvent was removed under reduced pressure and the residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 90:10 to 60:40) to give (2R,3R,4S)-5-(*tert*-butyldimethylsilyloxy)-2,4-dimethyl-3-hydroxy-pentanal (590 mg, 87%). RN: 209251-54-7; ¹H NMR (400.0 MHz, CDCl₃): δ = 9.83 (d, J = 2.3 Hz, 1H), 4.06 (dt, J = 9.6, 1.8 Hz, 1H), 3.84 (dd, J = 9.8, 3.4 Hz, 1H), 3.73 (dd, J = 9.8, 4.3 Hz, 1H), 3.49 (d, J = 1.8 Hz, 1H; OH), 2.52 (dqd, J = 9.6, 7.3, 2.3 Hz, 1H), 1.77 (qddd, J = 6.9, 4.3, 3.4, 1.8 Hz, 1H), 1.01 (d, J = 7.3 Hz, 3H; CH₃), 0.98 (d, J = 6.9 Hz, 3H; CH₃), 0.90 (s, 9H; 3CH₃), 0.08 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ = 205.9 (CH), 75.4 (CH), 67.8 (CH₂), 49.7 (CH), 35.8 (CH), 26.0 (3CH₃), 18.3 (C), 10.6 (CH₃), 9.4 (CH₃), -5.5 ppm (2 CH₃).

A solution of commercial diethyl N-methoxy-N-methylphosphonoacetamide (3.9 mL, 19.1 mmol, 2.0 equiv) was added to a suspension of sodium hydride (powder, 60% in oil, 763 mg, 19.1 mmol, 2.0 equiv) in dry THF (80 mL) at 0 °C. The solution was stirred at 20 °C for 1 h (until hydrogen evolution was not observed). The solution was cooled to 0°C, and the preceding aldehyde (2R,3R,4S)-5-(tert-butyldimethylsilyloxy)-2,4-dimethyl-3-hydroxy-pentanal (2.48 g, 9.52 mmol, 1.0 equiv) in dry THF (20 mL) was added dropwise. After 5 min, the mixture was allowed to warm to 20°C for 1 h and was then guenched by the addition of a saturated agueous NH4Cl solution. The aqueous layer was separated and extracted with Et₂O (2×). The combined organic phases were washed with water and brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 70:30 to 40:60) to give 53 (2.83 g, 86%) as a pale-yellow oil. $[\alpha]_D^{20} = -3.3$ (c=1.0 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 6.99$ (dd, J = 15.6, 8.7 Hz, 1 H), 6.49 (d, J =15.6 Hz, 1 H), 3.76 (dd, J=9.8, 3.9 Hz, 1 H), 3.71 (s, 3 H; CH₃), 3.69 (dd, J=6.4, 5.0 Hz, 1 H), 3.68 (dd, J=9.8, 5.0 Hz, 1 H), 3.24 (s, 3 H; CH₃), 2.84 (brs, 1H; OH), 2.50 (dqd, J=8.7, 6.9, 6.4 Hz, 1H), 1.82 (qddd, J=7.3, 5.0, 5.0, 3.9 Hz, 1 H), 1.01 (d, J=6.9 Hz, 3H; CH₃), 0.96 (d, J=7.3 Hz, 3H; CH₃), 0.90 (s, 9H; 3CH₃), 0.09 ppm (s, 6H, 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): δ=167.0 (C), 150.9 (CH), 119.1 (CH), 77.3 (CH), 68.6 (CH2), 61.8 (CH3, CH3), 40.9 (CH), 36.2 (CH), 32.4 (CH3), 26.0 (3 CH_3) , 18.3 (C), 16.6 (CH₃), 9.5 (CH₃), -5.5 ppm (2 CH_3) ; IR (film): $\tilde{\nu} =$ 3444, 2958, 2930, 2857, 1650, 1627, 1471, 1417, 1385, 1255, 1093, 1004, 850 cm^{-1} ; MS (GC, EI): m/z: 288 $[M-tBu]^+$, 267, 246, 227, 203, 187, 172, 145, 115, 89, 75, 55; elemental analysis calcd (%) for C₁₇H₃₅NO₄Si: C 59.09, H 10.21, N 4.05; found: C 58.95, H 10.28, N 3.94.

(2S.4S.5S.6R)-2-{6-[2-(tert-Butyldimethylsilyloxy)-1-methyl ethvll-5methyl-2-phenyl-1,3-dioxinan-4-yl}-N-methoxy-N-methyl-acetamide (54): KHMDS (0.5 M solution in toluene, 175 µL, 87.5 µmol, 0.1 equiv) was added to a solution of compound 53 (300 mg, 0.87 mmol, 1.0 equiv) and freshly distilled benzaldehyde (97 µL, 0.96 mmol, 1.1 equiv) in dry THF (9 mL) at 0°C. The yellow solution obtained was stirred for 15 min. This sequence (addition of KHMDS/stirring) was repeated twice, after which the solution was quenched by the addition of a saturated aqueous NH₄Cl solution. The organic layer was washed with water and brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by flash chromatography on silica gel (cyclohexane/ethyl acetate 95:5 to 50:50) to give 54 (311 mg, 79%) as a pale-yellow oil. $[\alpha]_{D}^{20} = +21.9$ (c=0.6 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.46-7.43$ (m, 2H), 7.36-7.29 (m, 3H), 5.58 (s, 1H), 4.12 (ddd, J=10.0, 9.6, 3.0 Hz, 1 H), 3.74 (dd, J=9.8, 1.7 Hz, 1 H), 3.69 (dd, J=9.6, 9.2 Hz, 1 H), 3.65 (s, 3 H; CH₃), 3.50 (dd, J=9.6, 5.5 Hz, 1 H), 3.22 (s, 3H), 2.92 (dd, J=15.6, 9.6 Hz, 1H), 2.64 (dd, J=15.6, 3.0 Hz, 1H), 2.01 (dqdd, J=9.2, 6.9, 5.5, 1.7 Hz, 1 H), 1.77 (ddq, J=10.0, 9.8, 6.9 Hz, 1H), 0.90 (d, J=6.9 Hz, 3H; CH₃), 0.89 (s, 9H; 3CH₃), 0.85 (d, J=6.9 Hz, 3H; CH₃), 0.03 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 172.2$ (C), 139.1 (C), 128.4 (CH), 128.0 (2 CH), 126.1 (2 CH), 100.0 (CH), 80.3 (CH), 79.2 (CH), 65.0 (CH₂), 61.6 (CH₃), 36.7 (CH₂), 36.0 (CH), 35.4 (CH), 32.2 (CH₃), 26.0 (3 CH₃), 18.4 (C), 12.0 (CH₃), 9.8 (CH₃), -5.3 ppm (2 CH₃); IR (film): $\tilde{\nu} = 2956$, 2930, 2857, 1666, 1483, 1403, 1387, 1256, 1149, 1090, 1029, 837 cm⁻¹; MS (GC, EI): *m*/*z*: 394 [M-tBu]⁺, 330, 288, 258, 243, 221, 199, 185, 172, 145, 115, 89, 75, 55; elemental analysis calcd (%) for C₂₄H₄₁NO₅Si: C 63.82, H 9.15, N 3.10; found: C 63.79. H 9.21. N 3.01.

(25,45,55,6R)-2-{6-[2-(*tert*-Butyldimethylsilyloxy)-1-methyl ethyl]-5methyl-2-phenyl-1,3-dioxinan-4-yl}acetaldehyde (55): DIBAL-H (1 M so-

lution in CH₂Cl₂, 10.0 mL, 9.96 mmol, 3.0 equiv) was added to a solution of Weinreb amide 54 (1.50 g, 3.32 mmol, 1.0 equiv) in CH₂Cl₂ (20 mL) at -78°C. The yellow solution was stirred at -78°C for 1 h. After this time, the mixture was quenched by the addition of ethyl acetate (15 mL). The mixture was allowed to warm to 20 °C and a solution of Rochelle's salt (17.0 g, 60.2 mmol, 18.1 equiv, in 20 mL water) was added. The mixture was stirred at 20°C for 2 h and then the organic layer was extracted with $Et_2O(2\times)$. The combined organic extracts were washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The crude residue was purified by chromatography on silica gel (cyclohexane/ethyl acetate 95:5 to 80:20) to give 55 (1.21 g, 93%) as a paleyellow oil. $[\alpha]_{D}^{20} = +12.3$ (c=1.0 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 9.83$ (dd, J = 2.3, 1.8 Hz, 1 H), 7.48–7.43 (m, 2 H), 7.40–7.31 (m, 3H), 5.58 (s, 1H), 4.07 (ddd, J=9.6, 7.8, 3.7 Hz, 1H), 3.75 (dd, J=10.6, 1.8 Hz, 1 H), 3.66 (t, J=9.2 Hz, 1 H), 3.47 (dd, J=9.2, 6.0 Hz, 1 H), 2.74 (ddd, J=10.5, 3.7, 1.8 Hz, 1 H), 2.71 (ddd, J=10.5, 7.8, 2.3 Hz, 1 H), 2.02 (dqdd, J=9.2, 7.3, 6.0, 1.8 Hz, 1 H), 1.78 (ddq, J=10.6, 9.6, 6.9 Hz, 1 H), 0.90 (s, 9 H; 3 CH₃), 0.90 (d, J=7.3 Hz, 3 H; CH₃), 0.88 (d, J= $6.9~\text{Hz},~3\,\text{H};~CH_3),~0.04~\text{ppm}$ (s, $6\,\text{H};~2\,\text{CH}_3);~^{13}\text{C}\,\text{NMR}$ (100.5 MHz, CDCl₃): δ = 201.6 (C), 138.6 (C), 128.7 (CH), 128.2 (2 CH), 126.1 (2 CH), 100.3 (CH), 80.1 (CH), 77.4 (CH), 64.9 (CH₂), 47.0 (CH₂), 36.6 (CH), 35.1 (CH), 26.0 (3 CH₃), 18.4 (C), 11.8 (CH₃), 9.7 (CH₃), -5.3 ppm (2 CH_3) ; IR (film): $\tilde{\nu} = 2955$, 2929, 2882, 2856, 1728, 1471, 1462, 1402, 1256, 1101, 1073, 1029, 837 cm⁻¹; MS (GC, EI): m/z: 335 [M-tBu]⁺, 281, 263, 243, 229, 211, 199, 185, 171, 157, 145, 131, 115, 101, 89, 75.

$$\label{eq:second} \begin{split} & [2S,4R(1S),5S,6S(2S,5S,6R,7S,8Z)]-4-[2-(tert-Butyldimethylsilyloxy)-1- \\ methyl ethyl-1-yl]-6-{5,7-dimethyl-2-hydroxy-6-[(methoxymethyl)oxy]-9- \\ tributylstannyl-dec-8-en-3-yn-1-yl]-5-methyl-2-phenyl-1,3-dioxinan (56) \\ & and [2S,4R(1S),5S,6S(2R,5S,6R,7S,8Z)]-4-[2-(tert-butyldimethylsilyloxy)-1- \\ methyl ethyl-1-yl]-6-{5,7-dimethyl-2-hydroxy-6-[(methoxymethyl)oxy]-9- \\ tributylstannyl-dec-8-en-3-yn-1-yl]-5-methyl-2-phenyl-1,3-dioxinan (57) \end{split}$$

Coupling reaction between B fragment 32 and C aldehyde 55: A solution of *t*BuLi (1.5 m in pentane, 295 µL, 0.44 mmol, 1.98 equiv) was added to a solution of alkyne 32 (305 mg, 0.62 mmol, 2.5 equiv) in THF (3 mL) at -78° C. The resulting mixture was stirred for 30 min at -78° C and then a solution of aldehyde 55 (100 mg, 0.25 mmol, 1.0 equiv) in THF (1 mL) was slowly added. After the reaction mixture had been stirred for 3 h at 0° C, a saturated aqueous NH₄Cl solution was added and the reaction mixture was extracted with Et₂O. The organic layers were washed with water and brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/Et₂O 98:2 to 80:20) to furnish the title compound 56 (30 mg, 14%), its diastereomer 57 (74 mg, 34%), and the starting aldehyde 55 (14 mg, 14%).

Compound 56: $[a]_{D}^{20} = -1.33$ (c = 0.64 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.49-7.43$ (m, 2H), 7.39–7.30 (m, 3H), 5.97 (dq, J = 9.6, 1.4, $J_{1H,117Sn} = J_{1H,119Sn} = 132.8 \text{ Hz}, 1 \text{ H}$, 5.59 (s, 1 H), 4.80 (d, J = 6.9 Hz, 1 H), 4.70 (d, J = 6.9 Hz, 1H), 4.70–4.67 (ddd, J = 7.8, 7.3, 2.7 Hz, 1H), 4.03 (ddd, J=10.1, 10.1, 2.3 Hz, 1 H), 3.73 (dd, J=10.1, 1.4 Hz, 1 H), 3.69 (dd, J=9.6, 9.2 Hz, 1H), 3.50 (dd, J=9.6, 6.0 Hz, 1H), 3.43 (s, 3H; CH₃), 3.23 (dd, J=5.9, 5.0 Hz, 1 H), 2.98 (d, J=7.3 Hz, 1 H; OH), 2.77 (qdd, J=6.9, 5.0, 1.8 Hz, 1 H), 2.24 (dqd, J=9.6, 6.9, 5.9 Hz, 1 H), 2.12 (ddd, J=14.6, 7.8, 2.3 Hz, 1 H), 2.10 (dddm, J=9.2, 6.0, 1.4 Hz, 1 H), 1.95 (ddd, J=14.6, 10.1, 2.7 Hz, 1 H), 1.89 (d, J = 1.4, $J_{^{1}H_{,}^{117}Sn} = J_{^{1}H_{,}^{119}Sn} = 44.8$ Hz, 3 H; CH₃), 1.76 (ddq, J=10.1, 10.1, 6.4 Hz, 1 H), 1.54–1.43 (m, 6 H; 3 CH₂), 1.33–1.29 (sext, J = 7.8 Hz, 6H; 3CH₂), 1.22 (d, J = 6.9 Hz, 3H; CH₃), 1.02 (d, J =6.9 Hz, 3 H; CH₃), 1.0–0.81 (m, 18 H; 4 CH₃, 3 CH₂), 0.89 (s; 3 CH₃), 0.81 (d, J=6.4 Hz, 3H; CH₃), 0.04 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 143.9$ ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 26.8$ Hz; CH), 138.6 ($J_{^{13}C,^{117}Sn} = 26.8$ Hz; CH), 138.6 (J_{^{13}C,^{117}Sn} = 26.8 Hz; CH), 138.6 (J_{^{13}C,^{117}Sn} = J_{13C,19Sn}=385.3 Hz; C), 137.7 (C), 128.3 (CH), 127.9 (2 CH), 125.7 (2 CH), 99.8 (CH), 98.1 (CH2), 87.0 (CH), 85.5 (C), 82.5 (C), 80.1 (CH), 79.6 (CH), 64.7 (CH₂), 60.0 (CH), 59.6 (CH₃), 42.2 (*J*_{13C,117Sn}=*J*_{13C,119Sn}=30.6 Hz; CH), 42.1 (CH₂), 36.4 (CH), 34.7 (CH), 30.0 (CH), 29.1 (J_{13C,117Sn}= $J_{^{13}C,^{119}Sn} = 19.1 \text{ Hz}; 3 \text{ CH}_2), 27.3 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 57.5 \text{ Hz}; 3 \text{ CH}_2), 26.7$ $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 44.1 \text{ Hz}; \text{ CH}_3), 25.8 (3 \text{ CH}_3), 18.6 (\text{CH}_3), 18.1 (\text{C}), 16.6$ (CH₃), 13.6 (3 CH₃), 11.5 (CH₃), 9.8 ($J_{13C,117Sn} = 313.4$, $J_{13C,119Sn} = 327.8$ Hz; 3 CH_2), 9.5 (CH₃), -5.5 ppm (2 CH₃); IR (film): $\tilde{\nu} = 2956$, 2928, 2855, 2360, 2341, 1594, 1455, 1404, 1375, 1255, 1149, 1095, 1050, 836, 775, 695, 663, 661 cm $^{-1}\!;$ elemental analysis calcd (%) for $C_{46}H_{82}O_6SiSn\colon C$ 62.93, H 9.41; found: C 63.10, H 9.55.

Compound 57: ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.47-7.24$ (m, 2H), 7.39–7.30 (m, 3 H), 6.00 (dq, J = 9.6, 1.4, $J_{^{1}H_{1}^{117}Sn} = J_{^{1}H_{1}^{119}Sn} = 137.4$ Hz, 1 H), 5.53 (s, 1H), 4.81 (d, J=6.9 Hz, 1H), 4.77-4.70 (m, 1H), 4.71 (d, J= 6.9 Hz, 1 H), 3.78-3.65 (m, 3 H), 3.49 (dd, J=9.6, 6.0 Hz, 1 H), 3.43 (s, 3H; CH₃), 3.26 (t, J=5.5 Hz, 1H), 2.89 (d, J=3.2 Hz, 1H; OH), 2.74 (qdd, J=6.9, 5.5, 2.7 Hz, 1H), 2.23 (dqd, J=9.6, 6.9, 5.5 Hz, 1H), 2.12 (ddd, J=13.7, 6.0, 1.8 Hz, 1 H), 2.06-1.96 (m, 2 H; H-2), 1.90 (d, J=1.4, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 42.6 \text{ Hz}, 3 \text{ H}; \text{ CH}_{3}, 1.80 - 1.71 \text{ (m, 1 H)}, 1.53 - 1.45 \text{ (m, }$ 6H; 3CH₂), 1.32 (sext, J = 7.3 Hz, 6H; 3CH₂), 1.21 (d, J = 6.9 Hz, 3H; CH₃), 1.00 (d, J=6.9 Hz, 3H; CH₃), 1.00–0.86 (m, 18H; 4CH₃, 3CH₂), 0.90 (s, 9H; 3CH₃), 0.82 (d, J=6.9 Hz, 3H), 0.04 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 144.2$ (CH), 138.7 (C), 137.8 (C), 128.7 (CH), 128.2 (2CH), 126.0 (2CH), 100.5.1 (CH), 98.3 (CH₂), 87.4 (CH), 85.7 (C), 82.5 (C), 81.7 (CH), 80.4 (CH), 65.0 (CH₂), 61.8 (CH), 56.3 (CH₃), 42.3 (CH), 41.2 (CH₂), 36.7 (CH), 35.2 (CH), 30.3 (CH), 29.3 $(J_{^{13}C_1}, I_{^{17}Sn} = J_{^{13}C_1}, I_{^{19}Sn} = 20.1 \text{ Hz}; 3 \text{ CH}_2), 27.6 (J_{^{13}C_1}, I_{^{17}Sn} = J_{^{13}C_1}, I_{^{19}Sn} = 57.5 \text{ Hz};$ 3 CH_2), 27.3 ($J_{^{13}\text{C},^{117}\text{Sn}} = J_{^{13}\text{C},^{119}\text{Sn}} = 46.0 \text{ Hz}$; CH₃), 26.1 (3 CH₃), 18.8 (CH₃), 18.5 (C), 16.5 (CH₃), 13.9 (3 CH₃), 11.9 (CH₃), 10.1 ($J_{^{13}C_1^{117}Sn} = 312.5$, *J*_{13C,119Sn} = 329.8 Hz; 3 CH₂), 9.8 (CH₃), -5.5 ppm (2 CH₃).

Reduction of ynone **58** (vide infra): A solution of (S)-CBS-oxazaborolidine in toluene (1 M, 8.22 mL, 8.22 mmol, 2.0 equiv) and BH₃-Me₂S (2 M in THF, 10.3 mL, 20.5 mmol, 5.0 equiv) were added to a solution of compound **58** (3.6 g, 4.1 mmol, 1.0 equiv) in THF (20 mL) at -30 °C. The resulting mixture was stirred for 2 h at -30 °C and was then quenched with EtOH and extracted with Et₂O. The organic layers were washed with water and brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/AcOEt 95:5 to 70:30) to give the title compound **56** (6.88 g, 95%).

Compound 57 was also obtained by enantioselective reduction of ynone 58: A solution of (*R*)-CBS-oxazaborolidine in toluene (1 m, 171 µL, 0.171 mmol, 2.0 equiv) and BH₃·Me₂S (2 m in THF, 214 µL, 0.43 mmol, 5.0 equiv) were added to a solution of ynone 58 (75 mg, 0.085 mmol, 1.0 equiv) in THF (0.5 mL) at -30 °C. The resulting mixture was stirred for 2 h at -30 °C and was then quenched with EtOH and extracted with Et₂O. The organic layers were washed with water and brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ AcOEt 95:5 to 70:30) to give 57 (63 mg, 90%).

[2*S*,4*R*(1*S*),5*S*,6*S*(5*S*,6*R*,7*S*,8*Z*)]-4-[2-(*tert*-Butyldimethylsilyloxy)-1methyl ethyl-1-yl]-6-{5,7-dimethyl-6-[(methoxymethyl)oxy]-9-tributylstannyl-2-oxo-dec-8-en-3-yn-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (58)

Coupling reaction with tBuLi: A solution of tBuLi (1.5 m in pentane, 295 µL, 0.44 mmol, 1.98 equiv) was added to a solution of alkyne **32** (215 mg, 0.44 mmol, 2.0 equiv) in THF (3 mL) at $-78 \,^{\circ}$ C. The resulting mixture was stirred for 30 min at $-78 \,^{\circ}$ C and then a solution of amide **54** (100 mg, 0.22 mmol, 1.0 equiv) in THF (1 mL) was slowly added. After the reaction mixture had been stirred for 3 h at 0 $\,^{\circ}$ C, a saturated aqueous NH₄Cl solution was added and the reaction mixture was extracted with Et₂O. The organic layers were washed with water and brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/Et₂O 98:2 to 80:20) to furnish the title compound **58** (156 mg, 81 %).

Coupling reaction with nBuLi—preparation of nBuLi in Et₂O: Small pieces of lithium (wire containing 0.5–1% Na, 4.3 g, 619 mmol, 2.48 equiv) were introduced to a four-necked flask (500 mL) containing dry Et₂O (100 mL) and equipped with a thermometer, a mechanical stirrer, and a dropping funnel under argon. After starting the stirrer, 30 drops of a *n*butylbromide (27 mL, 250 mmol, 1.0 equiv) solution in Et₂O (50 mL) were added from the dropping funnel. After a few minutes, the solution became slightly cloudy and bright spots appeared on the lithium. The reaction mixture was then cooled to -10° C by immersing in a -30° C nitrogen/acetone bath. The remainder of the *n*-butylbromide solution was added over 20 min while keeping the internal temperature below -10° C. After the addition was complete, the reaction mixture was allowed to warm up gradually to 0 and then 10° C with stirring over 1.5 h.

The pale-blue solution was then filtered whilst being transferred to a flask by cannula, titrated by a standard method (1.23 M, 74 %), and kept under argon at 4°C.

A solution of nBuLi in Et₂O (1.23 M, 2.8 mL, 3.44 mmol, 1.98 equiv) was added to a solution of alkyne 32 (1.7 g, 3.5 mmol, 2.0 equiv) in THF (25 mL) at -40 °C. The resulting mixture was stirred for 50 min at -40 °C and then a solution of amide 54 (789 mg, 1.75 mmol, 1.0 equiv) in THF (12 mL) was slowly added. After the reaction mixture had been stirred for 1 h at 0 °C, a saturated aqueous NH4Cl solution was added and the reaction mixture was extracted with Et2O. The organic layers were washed with water and brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/Et₂O 98:2 to 80:20) to furnish the title compound **58** (1.22 g, 81%). $[a]_{D}^{20} = +13.4$ (c = 0.94 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.43-7.40$ (m, 2H), 7.38–7.34 (m, 3H), 5.92 (dq, J = 6.9, 1.4 Hz, 1 H), 5.54 (s, 1 H), 4.71 (d, J = 6.9 Hz, 1 H), 4.66 (d, J = 6.9 Hz) 6.9 Hz, 1 H), 4.20 (ddt, J=9.6, 6.4, 5.9 Hz, 1 H), 3.73 (dd, J=9.6, 1.4 Hz, 1H), 3.69 (dd, J=9.6, 9.2 Hz, 1H), 3.51 (dd, J=9.6, 5.9 Hz, 1H), 3.39 (s, 3H; CH₃), 3.27 (dd, J=6.4, 4.1 Hz, 1H), 2.87 (qd, J=6.9, 4.1 Hz, 1H), 2.85 (d, J=5.9 Hz, 2 H), 2.22 (dqd, J=9.6, 6.9, 6.4 Hz, 1 H), 1.98 (dddm, J=9.2, 5.9, 1.4 Hz, 1 H), 1.88 (d, J=1.4, $J_{^{1}H,^{117}Sn}=J_{^{1}H,^{119}Sn}=41.2$ Hz, 3 H; CH₃), 1.72 (ddq, J=9.6, 9.6, 6.9 Hz, 1 H), 1.54–1.43 (m, 6H; 3 CH₂), 1.32 (sext, J = 7.3 Hz, 6H; 3CH₂), 1.24 (d, J = 6.9 Hz, 3H; CH₃), 1.00 (d, J =6.9 Hz, 3H; CH₃), 0.95–0.87 (m, 27H; 7CH₃+3CH₂), 0.80 (d, J=6.9 Hz, 3H; CH₃), 0.04 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta =$ 185.3 (C), 143.4 (CH), 138.8 (C), 138.7 (C), 128.4 (CH), 128.0 (2 CH), 126.0 (2 CH), 99.9 (CH), 98.3 (CH₂), 96.5 (C), 85.2 (CH), 83.0 (C), 80.3 (CH), 78.2 (CH), 65.0 (CH₂), 56.4 (CH₃), 49.5 (CH₂), 42.8 ($J_{^{13}C^{117}Sn} =$ J_{13C,119Sn}=30.6 Hz; CH), 36.7 (CH), 35.0 (CH), 30.8 (CH), 29.4 (3 CH₂), 27.5 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 58.4 \text{ Hz}; \text{ CH}_3, 3 \text{ CH}_2), 26.0 (3 \text{ CH}_3), 18.4 (C), 17.8$ (CH₃), 17.1 (CH₃), 13.8 (3 CH₃), 11.9 (CH₃), 10.1 ($J_{13C,117Sn} = 313.4$, $J_{^{13}C_1^{119}Sn} = 328.8 \text{ Hz}; 3 \text{ CH}_2), 9.7 \text{ (CH}_3), -5.2 \text{ ppm} (2 \text{ CH}_3); \text{ IR} \text{ (film)}: \tilde{\nu} =$ 2956, 2855, 2212, 1678, 1465, 1403, 1376, 1345, 1255, 1215, 1147, 1092, 1032, 837, 758, 699 cm⁻¹; elemental analysis calcd (%) for $C_{46}H_{80}O_6SiSn$: C 63.08, H 9.21; found: C 63.18, H 9.32.

 $[2S\!,\!4R(1R)\!,\!5S\!,\!6S(2S\!,\!5S\!,\!6R\!,\!7S\!,\!8Z)]\text{-}4\text{-}(2\text{-}Methoxycarbonyl\text{-}1\text{-}methyl$ ethyl-1-yl)-6-{5,7-dimethyl-2-hydroxy-2,6-bis[(methoxymethyl)oxy]-9-tributylstannyl-dec-8-en-3-yn-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (59): Diisopropylethylamine (34.1 mL, 196 mmol, 20 equiv) and MOMCl (5.95 mL, 78.4 mmol, 10 equiv) were added to a solution of compound 56 (6.88 g, 7.84 mmol, 1.0 equiv), dimethylaminopyridine (192 mg, 1.57 mmol, 0.2 equiv) and TBAI (290 mg, 0.784 mmol, 0.1 equiv) in dried CH₂Cl₂ (35 mL) at 0 °C. After the reaction mixture had been stirred for 12 h at 20°C, it was quenched with a saturated aqueous NaHCO3 solution and extracted with Et₂O. The organic layers were washed with brine. dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/AcOEt 95:5 to 80:20) give to [2S,4R(1S),5S,6S(2S,5S,6R,7S,8Z)]-4-[2-(tert-butyldimethylsilyloxy)-1methyl ethyl-1-yl]-6-{5,7-dimethyl-2,6-bis[(methoxymethyl)oxy]-9-tributylstannyl-dec-8-en-3-yn-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (6.2 g. 86%) as a yellow oil. $[\alpha]_D^{20} = -17.2$ (c=1.52 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.49-7.47$ (m, 2H), 7.37-7.30 (m, 3H), 6.00 (d, 1 H, J = 9.6, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 133.7$ Hz), 5.51 (s, 1 H), 5.04 (d, J = 6.4 Hz, 1H), 4.82–4.78 (m, 1H), 4.78 (d, J=6.9 Hz, 1H), 4.70 (d, J=6.9 Hz, 1H), 4.70 (d, J=6.4 Hz, 1 H), 3.75-3.67 (m, 3 H), 3.51 (dd, J=9.6, 5.9 Hz, 1 H), 3.43 (s, 3H; CH₃), 3.36 (s, 3H; CH₃), 3.24 (t, J = 5.5 Hz, 1H), 2.72 (qd, J=6.9, 5.5 Hz, 1 H), 2.28-2.18 (dqd, J=9.6, 6.9, 5.5 Hz, 1 H), 2.28-2.18 (m, 1 H), 2.04–1.98 (m, 1 H), 1.89 (s, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 41.7$ Hz, 3 H; CH₃), 1.87-1.83 (m, 1H), 1.66-1.63 (m, 1H), 1.52-1.45 (m, 6H; 3CH₂), 1.33 (sext, J=7.3 Hz, 6H; 3CH₂), 1.18 (d, J=7.3 Hz, 3H; CH₃), 0.98 (d, J=6.9 Hz, 3H; CH₃), 0.98–0.85 (m, 18H; 4CH₃+3CH₂), 0.89 (s, 9H; 3CH₃), 0.81 (d, J=6.4 Hz, 3H; CH₃), 0.05 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 144.2$ ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 29.7$ Hz; CH), 139.1 (C), 137.5 (C), 128.3 (CH), 128.0 (2 CH), 125.8 (2 CH), 99.6 (CH), 98.1 (CH₂), 93.6 (CH₂), 87.7 (C), 85.3 (C), 80.1 (C), 80.0 (CH), 77.6 (CH), 65.0 (CH₂), 61.0 (CH), 56.1 (CH₃), 55.5 (CH₃), 42.0 (CH), 39.9 (CH₂), 36.6 (CH), 35.2 (CH), 30.2 (CH), 29.2 (*J*_{13C,117Sn}=*J*_{13C,119Sn}=20.1 Hz; 3 CH₂), 27.4 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 58.4 \text{ Hz}; \text{ CH}_3), 26.9 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 53.6 \text{ Hz};$ 3CH₂), 25.9 (3CH₃), 18.5 (CH₃), 18.3 (C), 16.3 (CH₃), 13.7 (3CH₃), 11.6 (CH₃), 9.9 $(J_{^{13}C_{117}S_n} = 313.4, J_{^{13}C_{119}S_n} = 327.8 \text{ Hz}; 3 \text{ CH}_2), 9.6 (CH_3),$ $-5.4 \text{ ppm} (2 \text{ CH}_3)$; IR (film): $\tilde{\nu} = 2956$, 2928, 2380, 2341, 1456, 1402, 1375, 1350, 1250, 1158, 1098, 1015, 920, 837, 775, 690, 663 cm⁻¹; elemental analysis calcd (%) for C₄₈H₈₆O₇SiSn: C 62.53, H 9.40; found: C 62.33, H 9.59. A HF-pyridine stock solution (500 µL, prepared from commercial HF-pyridine 2 mL/pyridine 4 mL/THF 10 mL) was added to a solution of the preceding di-MOM ether (100 mg, 0.108 mmol, 1.0 equiv) in THF (1 mL) at 0°C. After the reaction mixture had been stirred for 4 h at 20°C, the resulting mixture was quenched with a saturated aqueous NaHCO₃ solution and extracted with Et₂O. The organic layers were washed with brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/AcOEt 70:30 to 40:60) to give [2*S*,4*R*(1*S*),5*S*,6*S*(2*S*,5*S*,6*R*,7*S*,8*Z*)]-4-(2-hydroxy-1-methyl ethyl-1-yl)-6-{5,7-dimethyl-2,6-bis[(methoxymethyl)oxy]-9-tributylstannyl-dec-8-en-3yn-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (84 mg, 96%). $[\alpha]_{D}^{20} = -35.9$ (c = 1.19 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.46 - 7.43$ (m, 2H), 7.38–7.33 (m, 3H), 6.01 (dq, 1H, J=9.6, 1.4, $J_{1H,117Sn}=J_{1H,119Sn}=132.8$ Hz), 5.58 (s, 1H), 5.02 (d, J=6.9 Hz, 1H), 4.79 (d, J=6.4 Hz, 1H), 4.76 (d, J= 11.0 Hz, 1 H), 4.70 (d, J=6.4 Hz, 1 H), 4.59 (d, J=6.9 Hz, 1 H), 3.83–3.73 (m, 4H), 3.44 (s, 3H; CH₃), 3.34 (s, 3H; CH₃), 3.24 (t, J = 5.5 Hz, 1H), 2.72 (qd, J=6.9, 5.5 Hz, 1 H), 2.28-2.23 (dqd, J=9.6, 6.9, 5.5 Hz, 1 H), 2.28–2.23 (m, 1H), 2.06–2.02 (m, 1H), 1.89 (d, J=1.4, $J_{1H,117Sn}=J_{1H,119Sn}=$ 43.0 Hz, 3H; CH₃), 1.89–1.85 (m, 1H), 1.83 (dqd, J=6.9, 6.4, 3.4 Hz, 1 H), 1.49–1.38 (m, 6H; 3 CH₂), 1.37–1.28 (sext, J = 7.3 Hz, 6H; 3 CH₂), 1.18 (d, J = 6.9 Hz, 3H; CH₃), 1.43 (d, J = 6.9 Hz, 3H; CH₃), 1.03 (d, J =6.9 Hz, 3H; CH₃), 1.02–0.88 (m, 15H; 3CH₂+3CH₃), 0.85 ppm (d, J= 6.9 Hz, 3 H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 144.1$ (CH), 138.5 (C), 137.5 (C), 128.5 (CH), 128.1 (2CH), 125.7 (2CH), 99.9 (CH), 98.0 (CH₂), 93.6 (CH₂), 87.7 (C), 85.2 (CH), 83.5 (CH or CH₂), 80.3 (C), 77.5 (CH or CH₂), 66.7 (CH or CH₂), 60.9 (CH), 56.1 (CH₃), 55.4 (CH₃), 41.9 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 34.4 \text{ Hz}; \text{ CH}), 39.8 (CH_2), 35.5 (CH), 35.3 (CH), 30.1$ (CH), 29.2 $(J_{^{13}C_{^{117}Sn}} = J_{^{13}C_{^{119}Sn}} = 19.2 \text{ Hz}; 3 \text{ CH}_2), 27.4 (J_{^{13}C_{^{117}Sn}} = J_{^{13}C_{^{119}Sn}} =$ 58.5 Hz; 3 CH₂), 27.2 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 46.0$ Hz; CH₃), 18.5 (CH₃), 16.3 (CH₃), 13.6 (3 CH₃), 11.5 (CH₃), 9.8 ($J_{13C,117Sn} = 314.4$, $J_{13C,119Sn} = 327.8$ Hz; 3 CH_2), 9.5 ppm (CH₃); IR (film): $\tilde{\nu} = 3380$, 2956, 2927, 2872, 2852, 1590, 1454, 1436, 1376, 1351, 1154, 1098, 1029, 699 cm⁻¹; elemental analysis calcd (%) for C42H72O7Sn: C 62.46, H 8.98; found: C 62.58, H 9.00.

BAIB (1.44 g, 4.46 mmol, 1.1 equiv) and TEMPO (160 mg, 0.81 mmol, 0.2 equiv) were added to a solution of the preceding alcohol (3.3 g, 4.05 mmol, 1.0 equiv) in dried CH₂Cl₂ (20 mL). After the reaction mixture had been stirred for 3 h at 20°C, a 1 M Na₂S₂O₃ solution was added, and the reaction mixture was extracted with Et₂O. The organic layers were washed with a saturated aqueous NaHCO3 solution, brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The crude [2S,4R(1R),5S,6S(2S,5S,6R,7S,8Z)]-4-(2-formyl-1-methyl ethyl-1-yl)-6-{5,7-dimethyl-2,6-bis[(methoxymethyl)oxy]-9-tributylstannyldec-8-en-3-yn-1-yl}-5-methyl-2-phenyl-1,3-dioxinan was used without further purification for the next step. ¹H NMR (400.0 MHz, CDCl₃): δ = 9.80 (s, 1H), 7.42–7.32 (m, 5H), 6.00 (dq, J=8.2, 1.4, $J_{^{1}H_{1}^{117}Sn}=J_{^{1}H_{1}^{119}Sn}=$ 135.1 Hz, 1H), 5.59 (s, 1H), 5.03 (d, J=6.9 Hz, 1H), 4.80-4.76 (m, 1H), 4.78 (d, J=6.9 Hz, 1 H), 4.69 (d, J=6.9 Hz, 1 H), 4.58 (d, J=6.9 Hz, 1 H), 4.15 (dd, J = 10.1, 2.3 Hz, 1 H), 3.84 (ddd, J = 9.6, 9.2, 1.4 Hz, 1 H), 3.43 (s, 3H; CH₃), 3.35 (s, 3H; CH₃), 3.23 (t, J=5.5 Hz, 1H), 2.72 (qd, J=7.3, 5.5 Hz, 1H), 2.64 (qd, J=6.9, 2.3 Hz, 1H), 2.28-2.18 (dqd, J=8.2, 6.9, 5.5 Hz, 1 H), 2.28–2.18 (m, 1 H), 1.89 (d, J = 1.4, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 41.2$ Hz, 3H; CH₃), 1.89-1.84 (m, 1H), 1.84 (ddq, J=10.1, 9.6, 6.4 Hz, 1H), 1.51-1.44 (m, 6H; 3CH₂), 1.33 (sext, J = 7.3 Hz, 6H; 3CH₂), 1.23 (d, J =6.9 Hz, 3H; CH₃), 1.19 (d, J=7.3 Hz, 3H; CH₃), 0.99 (d, J=6.9 Hz, 3H; CH₃), 0.96–0.88 (m, 15H; 3CH₂+3CH₃), 0.90 ppm (d, J=6.4 Hz, 3H); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 204.0$ (C), 144.0 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} =$ 32.6 Hz; CH), 137.9 (C), 137.2 (C), 128.0 (2 CH), 127.3 (CH), 125.6 (2 CH), 99.6 (CH), 98.0 (CH₂), 94.2 (CH₂), 87.7 (C), 86.2 (CH), 80.6 (CH), 80.2 (C), 77.0 (CH), 60.7 (CH), 56.1 (CH₃), 55.4 (CH₃), 47.2 (CH), 41.9 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 34.4 \text{ Hz}; \text{ CH}), 39.7 (CH_2), 34.9 (CH), 30.1 (CH),$ 29.1 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 21.0 \text{ Hz}; 3 \text{ CH}_2), 27.3 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 57.4 \text{ Hz};$ 3 CH_2), 27.2 ($J_{^{13}\text{C},^{117}\text{Sn}} = J_{^{13}\text{C},^{119}\text{Sn}} = 46.0 \text{ Hz}$; CH₃), 18.4 (CH₃), 16.3 (CH₃), 13.6 (3 CH₃), 13.5 (CH₃), 9.8 (3 CH₂), 6.8 (CH₃).

A solution of NaClO₂ (549 mg, 6.07 mmol, 1.5 equiv) and NaH₂PO₄ (729 mg, 6.07 mmol, 1.5 equiv) in H₂O (21 mL) was added to a solution of the preceding aldehyde (3.26 g, 4.05 mmol, 1.0 equiv) in tert-butanol (245 mL) and 2-methyl-2-butene (55 mL). After the reaction mixture had been stirred for 1 h at 20°C, the same quantity of NaClO2 and NaH2PO4 solution was added. The resulting mixture was stirred for 1 h and was then quenched by the addition of a 0.1 N HCl solution. The mixture was extracted with CH₂Cl₂ and the organic layers were washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The crude [2S,4R(1R),5S,6S(2S,5S,6R,7S,8Z)]-4-(2-carboxy-1methyl ethyl-1-yl)-6-{5,7-dimethyl-2,6-bis[(methoxymethyl)oxy]-9-tributylstannyl-dec-8-en-3-yn-1-yl}-5-methyl-2-phenyl-1,3-dioxinan was used in the next step without further purification. ${}^{1}H$ NMR (400.0 MHz, CDCl₃): $\delta = 7.43 - 7.40$ (m, 2H), 7.33-7.28 (m, 3H), 6.00 (d, J = 8.2 Hz, 1H, $J_{1H_{117Sn}} = J_{1H_{119Sn}} = 133.3 \text{ Hz}$, 5.54 (s, 1 H), 5.02 (d, J = 6.9 Hz, 1 H), 4.79 (d, J = 6.4 Hz, 1H), 4.80–4.78 (m, 1H), 4.69 (d, J = 6.4 Hz, 1H), 4.09 (dd, J = 6. 10.1, 2.3 Hz, 1 H), 4.00 (d, J=6.9 Hz, 1 H), 3.79 (td, J=9.6, 1.4 Hz, 1 H), 3.43 (s, 3H; CH₃), 3.35 (s, 3H; CH₃), 3.23 (t, J = 5.5 Hz, 1H), 2.86 (qd, J=6.9, 2.3 Hz, 1H), 2.72 (qd, J=6.9, 5.5 Hz, 1H), 2.20 (dqd, J=8.2, 6.4, 5.5 Hz, 1H), 2.22–2.18 (m, 1H), 1.89 (s, $J_{^{1}H,^{117}Sn} = J_{^{1}H,^{119}Sn} = 42.6$ Hz, 3H; CH₃), 1.89–1.85 (m, 1 H), 1.64 (ddq, J=10.1, 9.6, 6.4 Hz, 1 H), 1.53–1.42 (m, 6H; 3CH₂), 1.31 (d, J = 6.9 Hz, 3H; CH₃), 1.30 (sext, J = 7.3 Hz, 6H; 3 CH₂), 1.17 (d, J=6.9 Hz, 3H; CH₃), 0.98 (d, J=6.4 Hz, 3H; CH₃), 0.94-0.85 ppm (m, 18H; 3CH₂, 4CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta =$ 178.0 (C), 144.2 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 29.0 \text{ Hz}$; CH), 138.2 (C), 137.6 (C), 128.5 (CH), 128.1 (2 CH), 125.7 (2 CH), 99.7 (CH), 98.1 (CH₂), 93.7 (CH₂), 88.0 (C), 85.3 (CH), 81.8 (CH), 80.0 (C), 77.0 (CH), 61.0 (CH), 56.2 (CH₃), 55.5 (CH₃), 42.1 (CH), 40.7 (CH), 39.8 (CH₂), 35.5 (CH), 30.2 (CH), 29.2 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 19.2 \text{ Hz}; 3 \text{ CH}_2), 27.3 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} =$ 57.4 Hz; 3 CH₂), 27.2 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 36.4$ Hz; CH₃), 18.6 (CH₃), 16.4 (CH₃), 15.2 (CH₃), 13.7 (3 CH₃), 11.5 (CH₃), 9.9 ppm ($J_{^{13}C,^{117}Sn}$ = 310.6, $J_{^{13}C,^{119}Sn} = 331.8 \text{ Hz}; 3 \text{ CH}_2$).

A (trimethylsilyl)diazomethane solution (2 m in hexane, 4.05 mmol, 1.0 equiv) was added to a solution of the preceding acid (3.3 g, 4.05 mmol, 1.0 equiv) in benzene (160 mL)/MeOH (45 mL) until a yellow tint persisted. The resulting mixture was stirred for 15 min and was then concentrated in vacuum. The residue was purified by chromatography on silica gel (cyclohexane/AcOEt 90:10 to 80:20) to give the expected product **59** (2.32 g, 69% over 3 steps). $[\alpha]_D^{20} = -34.5$ (c = 0.96 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): δ = 7.45–7.42 (m, 2H), 7.37–7.31 (m, 3H), 6.00 (dq, J=9.6, 1.8, $J_{1H,117Sn}=J_{1H,119Sn}=132.8$ Hz, 1 H), 5.55 (s, 1 H), 5.03 (d, J=6.4 Hz, 1H), 4.81-4.77 (m, 1H), 4.79 (d, J=6.9 Hz, 1H), 4.70 (d, J=6.9 Hz, 1 H), 4.56 (d, J=6.4 Hz, 1 H), 4.05 (dd, J=10.5, 3.2 Hz, 1 H), 3.78 (ddd, J=9.6, 9.1, 1.8 Hz, 1 H), 3.72 (s, 3 H; CH₃), 3.43 (s, 3 H; CH₃), 3.36 (s, 3H; CH₃), 3.23 (t, J=5.5 Hz, 1H), 2.82 (qd, J=7.3, 3.2 Hz, 1H), 2.72 (qd, J=6.9, 5.5 Hz, 1 H), 2.30-2.17 (dqd, J=9.6, 6.9, 5.5 Hz, 1 H), 2.30–2.17 (m, 1H), 1.89 (d, J=1.8, $J_{1H,117Sn}=J_{1H,119Sn}=42.1$ Hz, 3H; CH₃), 1.87-1.83 (m, 1H), 1.75 (ddq, J=10.5, 9.6, 6.9 Hz, 1H), 1.57-1.51 (m, 6H; 3CH₂), 1.37 (sext, J=7.3 Hz, 6H; 3CH₂), 1.24 (d, J=7.3 Hz, 3H; CH₃), 1.19 (d, J=6.9 Hz, 3H; CH₃), 1.00–0.80 (m, 15H; 3CH₂, 3CH₃), 0.85 (d, J = 6.9 Hz, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 174.8$ (C), 144.4 $(J_{13C,117Sn} = J_{13C,119Sn} = 27.8 \text{ Hz}; \text{ CH})$, 138.6 (C), 137.7 $(J_{13C,117Sn} =$ J_{13C119Sn}=369.0 Hz; C), 128.6 (CH), 128.5 (2CH), 125.8 (2CH), 99.7 (CH), 98.2 (CH₂), 93.8 (CH), 87.9 (C), 85.4 (C), 82.2 (CH), 80.5 (C), 77.0 (CH), 61.1 (CH), 56.2 (CH₃), 55.7 (CH₃), 52.1 (CH₃), 42.2 ($J_{13C,117Sn} = J_{13C,119Sn} =$ 42.6 Hz; CH), 41.1 (CH), 39.9 (CH₂), 35.8 (CH), 30.3 (CH), 29.4 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 19.2 \text{ Hz}; 3 \text{ CH}_2), 27.6 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 57.4 \text{ Hz};$ 3 CH_2), 27.4 ($J_{^{13}\text{C},^{117}\text{Sn}} = J_{^{13}\text{C},^{119}\text{Sn}} = 44.1 \text{ Hz}$; CH₃), 18.7 (CH₃), 16.5 (CH₃), 13.8 (3 CH₃), 11.6 (CH₃), 10.0 ($J_{^{13}C,^{117}Sn}$ =312.4, $J_{^{13}C,^{119}Sn}$ =327.8 Hz; 3 CH₂), 9.3 ppm (CH₃); IR (film): $\tilde{\nu} = 2955$, 2927, 2872, 2853, 1745, 1454, 1403, 1349, 1375, 1207, 1154, 1110, 1029, 700 cm⁻¹; elemental analysis calcd (%) for $C_{43}H_{72}O_8Sn: C$ 61.80, H 8.68; found: C 61.92, H 8.83.

[25,4*R*(1*R*),55,65(25,3*Z*,55,6*R*,75,8*Z*)]-4-(2-Methoxycarbonyl-1-methyl ethyl-1-yl)-6-(5,7-dimethyl-9-iodo-2,6-bis[(methoxymethyl)oxy]deca-3,8-dien-1-yl]-5-methyl-2-phenyl-1,3-dioxinan (60): PtO₂ (20 mg, 0.088 mmol, 0.35 equiv) was added to a solution of the preceding ester (200 mg, 0.24 mmol, 1.0 equiv) in AcOEt (2.5 mL). The resulting suspension was stirred for 12 h at 20°C under a H₂ atmosphere. After this time, the mix-

ture was filtered and the solvent was removed under reduced pressure.

The crude [2S,4R(1R),5S,6S(2S,3Z,5S,6R,7S,8Z)]-4-(2-methoxycarbonyl-1methyl ethyl-1-yl)-6-{5,7-dimethyl-2,6-bis[(methoxymethyl)oxy]-9-tributylstannyl-deca-3,8-dien-1-yl}-5-methyl-2-phenyl-1,3-dioxinan was used without further purification in the next step. ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.49-7.44$ (m, 2H), 7.35-7.29 (m, 3H), 5.93 (dq, J = 8.2, 1.4, $J_{1H,117Sn} = J_{1H,119Sn} = 136.0 \text{ Hz}, 1 \text{ H}$), 5.60 (s, 1 H), 5.55 (t, J = 10.5 Hz, 1 H), 5.25 (dd, J=10.5, 10.1 Hz, 1 H), 4.87 (dd, J=10.1, 9.2 Hz, 1 H), 4.72 (d, J = 6.9 Hz, 1H), 4.54 (s, 2H), 4.49 (d, J = 6.9 Hz, 1H), 4.05 (dd, J = 10.1, 3.2 Hz, 1 H), 3.82 (dd, J=10.1, 9.2 Hz, 1 H), 3.74 (s, 3 H; CH₃), 3.32 (s, 6H; 2CH₃), 3.14 (dd, J=6.9, 4.1 Hz, 1H), 2.85-2.79 (qd, J=6.4, 3.2 Hz, 1H), 2.85-2.79 (dqd, J=10.9, 6.9, 6.9 Hz, 1H), 2.17 (dqd, J=8.2, 6.9, 4.1 Hz, 1 H), 2.00–1.91 (m, 1 H), 1.85 (d, J = 1.4, $J_{1H,117Sn} = J_{1H,119Sn} = 42.1$ Hz, 3H; CH₃), 1.70-1.60 (m, 2H), 1.51-1.44 (m, 6H; 3CH₂), 1.32 (sext, J= 7.3 Hz, 6H; 3CH₂), 1.23 (d, *J*=6.9 Hz, 3H; CH₃), 0.98 (d, *J*=6.9 Hz, 3H; CH₃), 0.94 (d, J = 6.4 Hz, 3H; CH₃), 0.92–0.82 ppm (m, 18H; 3CH₂+ 4CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 174.8$ (C), 144.7 (CH), 138.7 (CH), 137.3 (C), 136.5 (C), 129.0 (CH), 128.2 (CH), 127.9 (2 CH), 125.7 (2CH), 99.5 (CH), 97.9 (CH₂), 93.1 (CH₂), 86.3 (CH), 82.2 (CH), 77.3 (CH), 66.2 (CH), 56.1 (CH₃), 55.2 (CH₃), 51.9 (CH₃), 41.0 (2 CH), 39.1 (CH₂), 35.9 (CH), 35.8 (CH), 29.2 ($J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 19.2 \text{ Hz}; 3 \text{ CH}_2$), 27.4 $(J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 57.5 \text{ Hz}; 3 \text{ CH}_2), 27.2 (J_{^{13}C,^{117}Sn} = J_{^{13}C,^{119}Sn} = 42.2 \text{ Hz};$ CH₃), 17.7 (CH₃), 15.6 (CH₃), 14.2 (CH₃), 13.7 (3 CH₃), 11.6 (CH₃), 9.9 ppm ($J_{^{13}C,^{117}Sn} = 313.4, J_{^{13}C,^{119}Sn} = 326.8 \text{ Hz}; 3 \text{ CH}_2$).

An iodine solution (48 mg, 0.19 mmol, 0.98 equiv) in CH₂Cl₂ (16 mL) was added to a solution of the preceding compound (168 mg, 0.2 mmol, 1.0 equiv) in CH₂Cl₂ (4 mL) at 0°C over 30 min. The resulting solution was immediately added to a 1 M Na₂S₂O₃ solution and the organic phase was then concentrated under reduced pressure. The residue was taken up with Et₂O (3 mL) and a 0.7 M KF solution (600 µL). After the reaction mixture had been stirred for 3 h at 20 °C, the solution was filtered and extracted with Et2O. The organic layers were washed with brine, dried over MgSO₄, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/ Et₂O 80:20 to 50:50) to give the title compound 60 (96 mg, 71% for 2 steps). $[\alpha]_{D}^{20} = +3.70 \ (c = 0.78 \ \text{in CHCl}_{3}); {}^{1}\text{H NMR} \ (400.0 \ \text{MHz}, \ \text{CDCl}_{3}):$ $\delta = 7.49 - 7.44$ (m, 2H), 7.35-7.30 (m, 3H), 5.60 (t, J = 10.5 Hz, 1H), 5.59 (s, 1H), 5.28 (t, J=10.5 Hz, 1H), 5.24 (dq, J=8.2, 1.4 Hz, 1H), 4.83 (ddd, J=10.5, 9.2, 1.8 Hz, 1 H), 4.72 (d, J=6.9 Hz, 1 H), 4.57 (s, 2 H), 4.51 (d, J = 6.9 Hz, 1 H), 4.08 (dd, J = 10.1, 3.2 Hz, 1 H), 3.84 (dd, J = 9.6, 9.2, 1.8 Hz, 1 H), 3.74 (s, 3 H; CH₃), 3.35 (s, 6 H; 2 CH₃), 3.26 (t, J = 5.5 Hz, 1H), 2.86-2.78 (dqd, J=10.5, 6.9, 5.5 Hz, 1H), 2.86-2.78 (qd, J=6.9, 3.2 Hz, 1H), 2.47 (dqd, J=8.2, 6.9, 5.5 Hz, 1H), 2.36 (d, J=1.4 Hz, 3H; CH₃), 1.99 (ddm, J=11.0, 1.8 Hz, 1 H), 1.70 (ddq, J=10.1, 9.6, 6.9 Hz, 1 H), 1.72–1.64 (m, 1 H), 1.25 (d, J = 6.9 Hz, 3 H; CH₃), 1.08 (d, J = 6.9 Hz, 3H; CH₃), 0.96 (d, J=6.9 Hz, 3H; CH₃), 0.89 ppm (d, J=6.9 Hz, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 174.8$ (C), 138.8 (C), 137.9 (C), 135.8 (CH), 129.4 (CH), 128.2 (2CH), 127.9 (2CH), 125.7 (CH), 100.4 (CH), 99.4 (CH₂), 97.9 (C), 93.1 (CH₂), 85.6 (CH), 82.1 (CH), 77.6 (CH), 66.1 (CH), 56.1 (CH₃), 55.2 (CH₃), 51.9 (CH₃), 43.9 (CH), 41.0 (CH), 39.2 (CH or CH₂), 35.7 (CH), 35.3 (CH₃), 33.6 (CH or CH₂), 18.4 (CH₃), 14.7 (CH₃), 11.7 (CH₃), 9.2 ppm (CH₃); IR (film): $\tilde{\nu}$ = 3428, 2952, 2928, 2880, 2821, 1744, 1454, 1435, 1398, 1207, 1152, 1090, 1045 $\rm cm^{-1};$ elemental analysis calcd (%) for C₃₁H₄₇IO₈: C 55.19, H 7.02; found: C 55.31, H 7.15. [2S,4R(1R),5S,6S(2S,3Z,5S,6R,7S,8Z,11S,12R,13S,14S,15S,16Z)]-4-(2-Methoxycarbonyl-1-methyl ethyl-1-yl)-6-{12-(tert-butyldimethylsilyloxy)-5,7,9,11,13,15-hexamethyl-2,6-bis[(methoxymethyl)oxy]-14-(triethylsilyloxy)nonadeca-3,8,16,18-tetraen-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (61): tBuLi (1.5м in pentane, 355 µL, 0.53 mmol, 3.8 equiv) was rapidly added to a solution of the alkyliodide 43 (145 mg, 0.26 mmol, 1.8 equiv) in Et₂O (3.7 mL) at -80 °C. After 2 min, 9-MeOBBN (1 m in hexane, 602 µL, 0.60 mmol, 4.3 equiv) followed by THF (3.7 mL) was added. The resulting mixture was stirred 10 min at -78 °C and was then allowed to warm to 20°C over a period of 1 h 15 min. An aqueous 3.5 M Cs₂CO₃ solution was then added (220 µL, 0.77 mmol, 5.5 equiv), followed by the addition

of the vinyliodide **60** (96 mg, 0.14 mmol, 1.0 equiv), followed by the addition of the vinyliodide **60** (96 mg, 0.14 mmol, 1.0 equiv) in DMF (3.7 mL). Finally, [PdCl₂(dppf)] (12 mg, 0.014 mmol, 0.10 equiv) and triphenylarsine (7 mg, 0.021 mmol, 0.12 equiv) were added, and the resulting dark solution was stirred at 20 °C for 16 h. After extraction with Et_2O , the organic layers were washed with brine, dried over MgSO₄, filtered, and the sol-

Chem. Eur. J. 2008, 14, 11092-11112

© 2008 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

www.chemeurj.org

vent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/Et₂O 80:20 to 60:40) to give the title compound **61** (87 mg, 60%). $[\alpha]_D^{20} = +13.8$ (c=0.90 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.47 - 7.44$ (m, 2 H), 7.35–7.30 (m, 3 H), 6.57 (ddd, J=16.9, 11.0, 10.5 Hz, 1H), 6.03 (dd, J=11.0, 10.5 Hz, 1H), 5.58 (s, 1H), 5.54 (dd, J=11.9, 10.5 Hz, 1H), 5.54 (t, J=10.5 Hz, 1H), 5.20 (d, J=16.9 Hz, 1 H), 5.22 (dd, J=10.5, 10.1 Hz, 1 H), 5.11 (d, J= 10.5 Hz, 1 H), 4.97 (d, J=10.1 Hz, 1 H), 4.84 (dd, J=11.0, 10.1 Hz, 1 H), 4.71 (d, J=6.9 Hz, 1 H), 4.54 (s, 2 H), 4.50 (d, J=6.9 Hz, 1 H), 4.06 (dd, J = 10.1, 3.2 Hz, 1 H), 3.82 (dd, J = 10.1, 8.7 Hz, 1 H), 3.74 (s, 3 H, CH₃), 3.55 (dd, J=6.9, 3.2 Hz, 1 H), 3.44 (dd, J=4.6, 4.1 Hz, 1 H), 3.34 (s, 3 H; CH₃), 3.33 (s, 3H; CH₃), 3.07 (t, J=5.5 Hz, 1H), 2.91–2.76 (m, 3H), 2.54 (dqd, J=10.1, 6.4, 5.5 Hz, 1 H), 2.08-1.55 (tq, J=10.1, 6.4 Hz, 1 H), 2.08-1.55 (m, 6H), 1.55 (s, 3H; CH₃), 1.22 (d, J=6.9 Hz, 3H; CH₃), 1.06-0.83 (m; 8CH₃), 0.91 (s, 9H; 3CH₃), 0.68 (q, J = 7.8 Hz, 6H; 3CH₂), 0.63 (d, J = 6.4 Hz, 3H; CH₃), 0.07 ppm (s, 6H; 2CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 174.8$ (C), 138.8 (C), 136.8 (CH), 134.4 (CH), 132.3 (C), 132.1 (CH), 130.4 (CH), 128.9 (3 CH), 128.2 (CH), 127.9 (2 CH), 125.8 (CH), 117.4 (CH₂), 99.6 (CH), 97.8 (CH₂), 92.5 (CH₂), 86.9 (CH), 82.2 (CH), 78.0 (CH), 77.0 (2CH), 66.4 (CH), 56.0 (CH₃), 55.2 (CH₃), 51.9 (CH₃), 41.9, 41.0, 40.3, 39.0, 36.2, 35.8, 35.3, 35.0, 34.4 (7 CH, 2 CH₂), 26.9 (3CH₃), 23.1 (CH₃), 18.5 (C), 18.5, 17.4, 16.5, 14.2, 11.6, 11.3, 9.3 (7CH₃, CH₃-2), 7.2 (3 CH₃), 5.7 (3 CH₂), -3.1 (CH₃), -3.2 ppm (CH₃); IR (film): $\tilde{\nu} = 3383, 2957, 2931, 2878, 2857, 2821, 2359, 2340, 1747, 1733, 1657, 1455,$ 1462, 1435, 1396, 1374, 1328, 1252, 1143, 1094, 1045, 919, 835, 772, 739, 701 cm⁻¹; HRMS (ESI): *m*/*z*: calcd for C₅₆H₉₈O₁₀NaSi₂: 1009.6591; found: 1009.6591.

Discodermolide 1: PTSA (3 mg, 0.015 mmol, 0.28 equiv) was added to a solution of compound 61 (53 mg, 0.054 mmol, 1.0 equiv) in MeOH (6.5 mL) at 0°C. After the reaction mixture had been stirred for 1 h at 0°C, triethylamine was added (0.3 equiv), and the solvent was removed under reduced pressure. The residue was purified by chromatography on gel (cyclohexane/AcOEt, 95:5 to 70:30) to give the silica [2S,4R(1R),5S,6S(2S,3Z,5S,6R,7S,8Z,11S,12R,13S,14S,15S,16Z)]-4-(2-meethyl-1-yl)-6-{12-(tert-butyldimethylsilyloxy)thoxycarbonyl-1-methyl 5,7,9,11,13,15-hexamethyl-14-hydroxy-2,6-bis[(methoxymethyl)oxy]nonadeca-3,8,16,18-tetraen-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (36 mg. 77%). $[\alpha]_{D}^{20} = +11.0$ (c = 0.90 in CHCl₃); ¹H NMR (400.0 MHz, CDCl₃): $\delta = 7.47 - 7.43$ (m, 2H), 7.33-7.30 (m, 3H), 6.64 (ddd, J = 16.9, 11.0, 10.1 Hz, 1 H), 6.16 (dd, J=11.0, 10.5 Hz, 1 H), 5.59 (s, 1 H), 5.54 (dd, J= 11.0, 10.1 Hz, 1 H), 5.33 (dd, J=10.5, 10.1 Hz, 1 H), 5.23 (dd, J=10.1, 9.6 Hz, 1 H), 5.20 (d, J=16.9 Hz, 1 H), 5.16 (d, J=10.1 Hz, 1 H), 4.99 (d, J=10.1 Hz, 1 H), 4.84 (dd, J=10.1, 9.6 Hz, 1 H), 4.71 (d, J=6.9 Hz, 1 H), 4.54 (s, 2H), 4.50 (d, J=6.9 Hz, 1H), 4.06 (dd, J=10.5, 3.2 Hz, 1H), 3.82 (t, J=9.6 Hz, 1 H), 3.74 (s, 3H; CH₃), 3.63 (dd, J=5.5, 2.8 Hz, 1 H), 3.33 (s, 6H; 2CH₃), 3.35–3.33 (m, 1H), 3.09 (t, J=5.5 Hz, 1H), 2.86–2.75 (dqd, J = 11.0, 7.3, 5.5 Hz, 1H), 2.86–2.75 (m, 2H), 2.54 (dqd, J = 10.1, 6.4, 5.5 Hz, 1 H), 2.19 (t, J=12.4 Hz, 1 H), 2.05–1.71 (m, 4 H), 1.58 (s, 3 H; CH₃), 1.65–1.55 (ddq, J=10.5, 9.6, 6.4 Hz, 1H), 1.65–1.55 (m, 1H), 1.44 (s, 1 H; OH), 1.22 (d, J=7.3 Hz, 3 H; CH₃), 1.01 (d, J=6.9 Hz, 3 H; CH₃), 0.96 (d, J = 6.4 Hz, 3H; CH₃), 0.95–0.91 (m, 6H; 2CH₃), 0.91 (s, 9H; 3CH₃), 0.87 (d, J=6.4 Hz, 3H; CH₃), 0.70 (d, J=6.9 Hz, 3H; CH₃), 0.10 ppm (s, 6H; 2CH₃); 13 C NMR (100.5 MHz, CDCl₃): $\delta = 174.8$ (C), 138.8 (C), 137.0 (CH), 134.7 (CH), 132.7 (C), 132.1 (CH), 131.1 (CH), 130.0 (CH), 129.1 (CH), 128.2 (2CH), 127.9 (2CH), 125.8 (CH), 118.5 (CH₂), 99.6 (CH), 97.9 (CH₂), 93.2 (CH₂), 86.8 (CH), 82.0 (CH), 79.0 (CH), 77.8 (CH), 76.7 (CH), 66.4 (CH), 56.1 (CH₃), 55.3 (CH₃), 51.9 (CH₃), 41.1, 39.1, 38.0, 36.7, 36.3, 35.9, 35.3, 34.7, 34.4 (7 CH, 2 CH₂), 26.3 (3CH₃), 23.3 (CH₃), 18.5 (C), 17.6, 17.1, 16.4, 13.4, 11.7, 9.5, 9.3 (7CH₃), -3.1 (CH₃), -3.6 ppm (CH₃); IR (film): $\tilde{\nu} = 2377$, 2957, 2929, 2883, 2857, 1591, 1462, 1434, 1395, 1257, 1153, 1093, 1036, 874 cm⁻¹; HRMS (ESI): m/z: calcd for C₅₀H₈₄O₁₀NaSi: 895.5726; found: 895.5721.

Trichloroacetylisocyanate (16 μ L, 0.14 mmol, 1.05 equiv) was added to a solution of the preceding alcohol (114 mg, 0.13 mmol, 1.0 equiv) in CH₂Cl₂ (10 mL). After the reaction mixture had been stirred for 15 min at 20 °C, the resulting mixture was concentrated in vacuo and the residue taken up in MeOH (13 mL). K₂CO₃ (99 mg, 0.71 mmol, 5.5 equiv) was added and the resulting solution was stirred for 1 h 15 min and then concentrated in vacuo and extracted with AcOEt. The organic layers were

washed with water, brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (cyclohexane/AcOEt 90:10 to 50:50) to give [2S,4R(1R),5S,6S(2S,3Z,5S,6R,7S,8Z,11S,12R,13S,14S,15S,16Z)]-4-(2-methoxycarbonyl-1-methyl ethyl-1-yl)-6-{12-(tert-butyldimethylsilyloxy)-14-(carbamoyloxy)-5,7,9,11,13,15-hexamethyl-2,6-bis[(methoxymethyl)oxy]nonadeca-3,8,16,18-tetraen-1-yl}-5-methyl-2-phenyl-1,3-dioxinan (76 mg, 64%) as a pale colorless oil. $[\alpha]_D^{20} = +17.4$ (c=0.80 in CHCl₃); ¹H NMR $(400.0 \text{ MHz}, \text{ CDCl}_3): \delta = 7.50-7.44 \text{ (m, 2H)}, 7.34-7.30 \text{ (m, 3H)}, 6.60 \text{ (dt,})$ J=16.9, 10.5 Hz, 1 H), 6.02 (dd, J=11.0, 10.5 Hz, 1 H), 5.58 (s, 1 H), 5.58-5.52 (m, 1 H), 5.53 (dd, J = 12.8, 10.5 Hz, 1 H), 5.38 (dd, J = 11.0, 10.5 Hz, 1H), 5.28-5.18 (m, 3H), 5.14 (d, J=10.5 Hz, 1H), 4.99 (d, J=10.1 Hz, 1 H), 4.82 (dd, J = 10.1, 9.2 Hz, 1 H), 4.74–4.70 (m, 1 H), 4.72 (d, J =6.9 Hz, 1 H), 4.54 (s, 2 H), 4.50 (d, J=6.9 Hz, 1 H), 4.06 (dd, J=10.1, 3.2 Hz, 1 H), 3.81 (dd, J=9.6, 9.2 Hz, 1 H), 3.73 (s, 3 H; CH₃), 3.43 (dd, J = 4.6, 4.1 Hz, 1H), 3.32 (s, 6H; 2CH₃), 3.06 (t, J = 5.5 Hz, 1H), 2.99 (dqd, J=10.5, 6.9, 3.7 Hz, 1 H), 2.83-2.79 (qdm, J=7.3, 5.5 Hz, 1 H), 2.81 (dqd, J=10.1, 6.9, 5.5 Hz, 1 H), 2.52 (dqd, J=10.1, 6.4, 5.5 Hz, 1 H), 2.15-1.79 (m, 5H), 1.74-1.52 (m, 2H), 1.57 (s, 3H; CH₃), 1.25 (d, J=7.3 Hz, 3H; CH₃), 1.01 (d, J=6.9 Hz, 3H; CH₃), 0.94–0.85 (m, 12H; 4CH₃), 0.91 (s, 9H; 3CH₃), 0.73 (d, J=6.9 Hz, 3H; CH₃), 0.07 (s, 3H; CH₃), 0.05 ppm (s, 3H; CH₃); ¹³C NMR (100.5 MHz, CDCl₃): $\delta = 174.9$ (C), 156.9 (C), 138.8 (CH), 137.0 (C), 133.6 (CH), 132.5 (CH), 132.1 (C), 130.1 (CH), 129.8 (CH), 129.0 (CH), 128.2 (CH), 127.9 (2CH), 125.8 (2CH), 117.9 (CH2), 99.7 (CH), 97.8 (CH2), 93.2 (CH2), 86.9 (CH), 82.1 (CH), 78.8 (CH), 77.0 (2CH), 66.4 (CH), 56.0 (CH₃), 55.2 (CH₃), 51.9 (CH₃), 41.2 (CH), 39.2 (CH₂), 37.8 (CH), 36.3 (CH), 35.9 (CH₂), 35.3 (CH), 35.1 (CH), 34.4 (CH), 29.7 (CH), 26.2 (3CH₃), 23.0 (CH₃), 18.5 (C), 18.5 (CH₃), 17.5, 16.5, 11.7, 10.1 (4CH₃), 13.7 (CH₃), 9.4 (CH₃), -3.3 (CH₃), -3.6 ppm (CH₃); IR (film): $\tilde{\nu}$ = 3359, 2981, 2930, 2884, 2857, 1728, 1598, 1455, 1435, 1395, 1363, 1326, 1258, 1214, 1146, 1094, 1035, 836, 773, 757 cm⁻¹; HRMS (ESI): m/z: calcd for C₅₁H₈₅O₁₁NaSi: 938.5784; found: 938.5793.

HCl (4N, 1.5 mL) was added to a solution of the preceding compound (14 mg, 0.015 mmol, 1.0 equiv) in THF (1.5 mL). The resulting mixture was stirred for 72 h at 20 °C and then solid NaHCO3 was added and the mixture extracted with AcOEt $(3 \times)$. The organic layers were washed with water, brine, dried over MgSO4, filtered, and the solvent was removed under reduced pressure. The residue was purified by chromatography on silica gel (CH2Cl2/MeOH 95:5 to 90:10) to give the title compound (+)-discodermolide (1) (6.2 mg, 70%) as a white solid. $[a]_{\rm D}^{20}$ = +16.2 (c=1.0 in MeOH); ¹H NMR (400.0 MHz, CDCl₃): δ =6.61 (ddd, J = 16.9, 11.0, 10.5 Hz, 1 H), 6.03 (t, J = 11.0 Hz, 1 H), 5.53 (dd, J = 11.0, 10.5 Hz) 7.8 Hz, 1 H), 5.42 (dd, J=11.0, 10.5 Hz, 1 H), 5.36 (dd, J=11.0, 10.5 Hz, 1 H), 5.23 (d, J=16.9 Hz, 1 H), 5.14 (d, J=10.5 Hz, 1 H), 5.12 (d, J=11.0 Hz, 1 H), 4.79–4.70 (m, 4 H), 4.63 (ddd, J=10.1, 8.2, 1.8 Hz, 1 H), 3.73 (t, J=3.7 Hz, 1 H), 3.28 (dd, J=5.0, 4.6 Hz, 1 H), 3.21 (dd, J=8.9, 5.0 Hz, 1 H), 3.00 (dqd, J=10.5, 6.9, 3.2 Hz, 1 H), 2.79 (m, 1 H), 2.70 (qd, J=7.3, 3.7 Hz, 1H), 2.61–2.54 (m, 1H; OH), 2.58 (m, 1H), 2.00–1.81 (m; 4H+2OH), 1.72–1.60 (m; 3H+OH), 1.64 (s, 3H; CH_3), 1.32 (d, J=7.3 Hz, 3H; CH₃), 1.08 (d, *J*=6.9 Hz, 3H; CH₃), 1.01 (d, *J*=7.3 Hz, 3H; CH₃), 0.98 (d, J=6.4 Hz, 3H; CH₃), 0.98 (d, J=6.4 Hz, 3H; CH₃), 0.95 (d, J = 6.4 Hz, 3H; CH₃), 0.83 ppm (d, J = 5.9 Hz, 3H; CH₃); ¹³C NMR $(100.5 \text{ MHz}, \text{ CDCl}_3): \delta = 174.3 \text{ (C)}, 157.4 \text{ (C)}, 134.5 \text{ (CH)}, 133.8 \text{ (CH)},$ 133.6 (C), 133.0 (CH), 132.2 (CH), 130.1 (CH), 129.8 (CH), 118.1 (CH₂), 79.1 (CH), 78.8 (CH), 77.7 (CH), 75.7 (CH), 73.4 (CH), 64.5 (CH), 43.4 (CH), 41.1 (CH₂), 37.5 (CH), 36.4 (CH), 35.9 (CH₂), 35.6 (CH), 35.0 (CH), 33.2 (CH), 30.0 (CH), 23.6 (CH₃), 18.6 (CH₃), 17.7 (CH₃), 16.1 (CH₃), 16.0 (CH₃), 13.9 (CH₃), 12.9 (CH₃), 9.3 ppm (CH₃); ¹H NMR (400.0 MHz, CD₃CN): $\delta = 6.69$ (ddd, J = 16.9, 11.0, 10.1 Hz, 1 H), 6.10 (t, J=11.0 Hz, 1 H), 5.56 (dd, J=10.5, 9.6 Hz, 1 H), 5.45 (dd, J=10.5, 9.2 Hz, 1H), 5.41 (dd, J=11.0, 10.5 Hz, 1H), 5.27 (d, J=16.9 Hz, 1H), 5.17 (d, J=10.1 Hz, 1 H), 5.15-5.04 (m, 2 H), 4.98 (d, J=10.1 Hz, 1 H), 4.74 (dd, J=8.2, 4.1 Hz, 1 H,), 4.56–4.41 (m, 2 H), 3.64 (t, J=4.1 Hz, 1 H), 3.16 (dd, J = 7.3, 3.2 Hz, 1H), 3.16–3.14 (m, 1H; OH), 3.09 (dd, J = 6.9, 3.7 Hz, 1H), 2.67-2.56 (m, 3H; OH), 2.35-2.10 (m, 4H; OH), 1.90-1.45 (m, 3H), 1.59 (s, 3H; CH₃), 1.21 (d, J=7.3 Hz, 3H; CH₃), 1.03 (d, J=6.4 Hz, 3H; CH₃), 1.01 (d, J=6.4 Hz, 3H; CH₃), 0.97 (d, J=6.9 Hz, 3H; CH₃), 0.90 (d, J = 6.9 Hz, 3H; CH₃), 0.82 (d, J = 6.9 Hz, 3H; CH₃), 0.75 ppm (d, J =

```
11110 -
```

5.9 Hz, 3H; CH₃); ¹³C NMR (100.5 MHz, CD₃CN): δ = 174.7 (C), 158.3 (C), 134.0 (CH), 133.8 (CH), 133.7 (CH), 133.2 (CH), 131.1 (CH), 130.5 (CH), 118.1 (CH₂), 79.7 (CH), 79.2 (CH), 77.4 (CH), 75.9 (CH), 73.1 (CH), 63.3 (CH), 44.0 (CH), 42.1 (CH₂), 38.4 (CH), 37.1 (CH), 36.5 (CH₂), 36.2 (CH), 36.1 (CH), 34.4 (CH), 34.2 (CH), 23.3 (CH₃), 19.7 (CH₃), 18.1 (CH₃), 17.6 (CH₃), 15.8 (CH₃), 15.6 (CH₃), 13.1 (CH₃), 9.2 ppm (CH₃); HRMS (ESI): *m*/*z*: calcd for C₃₃H₅₅O₈NNa: 616.3820 [*M*+Na]⁺; found: 616.3816.

Acknowledgements

We thank Sanofi-Aventis for a fellowship for E.d.L. Technical support from Paris XI University (Châtenay-Malabry and Orsay centers) is gratefully acknowledged (elemental analysis and HRMS).

- a) S. P. Gunasekera, R. E. Longley, R. A. Isbrucker, J. Nat. Prod. 2002, 65, 1830–1837; b) S. P. Gunasekera, M. Gunasekera, R. E. Longley, G. K. Schulte, J. Org. Chem. 1990, 55, 4912–4915; Correction: J. Org. Chem. 1991, 56, 1346; c) S. P. Gunasekera, G. K. Paul, R. E. Longley, R. A. Isbruker, S. A. Pomponi, J. Nat. Prod. 2002, 65, 1643–1648.
- [2] a) A. Canales, R. Matesanz, N. M. Gardner, J. M. Andreu, I. Paterson, J. F. Diaz, J. Jimenez-Barbero, Chem. Eur. J. 2008, 14, 7557-7569; b) V. M. Sánchez-Pedregal, K. Kubicek, J. Meiler, I. Lyothier, I. Paterson, T. Carlomagno, Angew. Chem. 2006, 118, 7548-7554; Angew. Chem. Int. Ed. 2006, 45, 7388-7394; c) S. Xia, C. S. Kenesky, P. V. Rucker, A. B. Smith, III, G. A. Orr, S. B. Horwitz, Biochemistry 2006, 45, 11762-11775; d) S. Honore, K. Kamath, D. Braguer, L. Wilson, C. Briand, M. A. Jordan, Mol. Cancer Ther. 2003, 2, 1303-1311; e) L. He, G. A. Orr, S. B. Horwitz, Drug Discovery Today 2001, 6, 1153-1164; f) L. He, H. Y. Chia-Ping, S. B. Horwitz, Mol. Cancer Ther. 2001, 1, 3-10; g) L. A. Martello, M. J. LaMarche, L. He, T. J. Beauchamp, A. B. Smith, III, S. B. Horwitz, Chem. Biol. 2001, 8, 843-855; h) M. Kalesse, ChemBioChem 2000, 1, 171-175; i) E. ter Haar, R. J. Kowalski, E. Hamel, C. M. Lin, R. E. Longley, S. P. Gunasekera, H. S. Rosenkranz, B. W. Day, Biochemistry 1996, 35, 243-250; j) R. Balachandran, E. ter Haar, M. J. Welsh, S. G. Grant, B. W. Day, Anti-Cancer Drugs 1998, 9, 67-76.
- [3] a) G. S. Huang, L. Lopez-Barcons, B. S. Freeze, A. B. Smith III, G. L. Goldberg, S. B. Horwitz, H. M. McDaid, *Clin. Cancer Res.* 2006, *12*, 298–304; b) L. A. Martello, H. M. McDaid, D. L. Regl, C.-P. H. Yang, D. Meng, T. R. R. Pettus, M. D. Kaufman, H. Arimoto, S. J. Danishefsky, A. B. Smith, III, S. B. Horwitz, *Clin. Cancer Res.* 2000, *6*, 1978–1987.
- [4] Total synthesis: a) J. B. Nerenberg, D. T. Hung, P. K. Somers, S. L. Schreiber, J. Am. Chem. Soc. 1993, 115, 12621-12622; b) D. T. Hung, J. B. Nerenberg, S. L. Schreiber, J. Am. Chem. Soc. 1996, 118, 11054-11080; c) A. B. Smith, III, Y. Qiu, D. R. Jones, K. Kobayashi, J. Am. Chem. Soc. 1995, 117, 12011-12012; d) A. B. Smith, III, T. J. Beauchamp, M. J. LaMarche, M. D. Kaufman, Y. Qiu, H. Arimoto, D. R. Jones, K. Kobayashi, J. Am. Chem. Soc. 2000, 122, 8654-8664, and references therein; e) S. S. Harried, G. Yang, M. A. Strawn, D. C. Myles, J. Org. Chem. 1997, 62, 6098-6099; f) S. S. Harried, C. P. Lee, G. Yang, T. I. H. Lee, D. C Myles, J. Org. Chem. 2003, 68, 6646-6660; g) J. A. Marshall, Z.-H. Lu, B. A. Johns, J. Org. Chem. 1998, 63, 817-823; h) J. A. Marshall, B. A. Johns, J. Org. Chem. 1998, 63, 7885-7892; i) I. Paterson, G. J. Florence, K. Gerlach, J. P. Scott, Angew. Chem. 2000, 112, 385-388; Angew. Chem. Int. Ed. 2000, 39, 377-380; j) I. Paterson, G. J. Florence, Eur. J. Org. Chem. 2003, 2193-2208; k) I. Paterson, I. Lyothier, J. Org. Chem. 2005, 70, 5494-5507, and references therein; 1) A. Arefolov, J. S. Panek, J. Am. Chem. Soc. 2005, 127, 5596-5603, and references therein. For recent reviews on syntheses and analogue design and activity see: m) S. J. Shaw, Mini-Rev. Med. Chem. 2008, 8, 276-284; n) A. B. Smith, III, B. S. Freeze, Tetrahedron 2008, 64, 261-298;

o) G. J. Florence, N. M. Gardner, I. Paterson, *Nat. Prod. Rep.* 2008, 25, 342–375.

- [5] a) S. J. Mickel, D. Niederer, R. Daeffler, A. Osmani, E. Kuesters, E. Schmid, K. Schaer, R. Gamboni, W. Chen, E. Loeser, F. R. Kinder, Jr., K. Konigsberger, K. Prasad, T. M. Ramsey, O. Repič, R.-M. Wang, G. Florence, I. Lyothier, I. Paterson, *Org. Process Res. Dev.* 2004, *8*, 122–130, and the four precedent papers; b) S. J. Mickel, R. Daeffler, W. Prikoszovich, *Org. Process Res. Dev.* 2005, *9*, 113–120; c) O. Loiseleur, G. Koch, J. Cercus, F. Schürch, *Org. Process Res. Dev.* 2005, *9*, 259–271.
- [6] a) A. Mita, C. Lockhart, T. L. Chen, K. Bocinski, J. Curtright, W. Cooper, L. Hammond, M. Rothenberg, E. Rowinsky, S. Sharma, J. Clin. Oncol. 2004, 22 (14S–ASCO Annual Meeting Proceedings, Post-Meeting Edition), abstract 2025.
- [7] Preliminary communication: E. de Lemos, F. H. Porée, A. Commerçon, J.-F. Betzer, A. Pancrazi, J. Ardisson, Angew. Chem. 2007, 119, 1949–1953; Angew. Chem. Int. Ed. 2007, 46, 1917–1921.
- [8] a) D. Hoppe, Angew. Chem. 1984, 96, 930–946; Angew. Chem. Int. Ed. Engl. 1984, 23, 932–948; b) D. Hoppe, O. Zschage, Angew. Chem. 1989, 101, 67–69; Angew. Chem. Int. Ed. Engl. 1989, 28, 69– 71; c) D. Hoppe, O. Zschage, Tetrahedron 1992, 4R, 8389–8392.
- [9] a) V. Fargeas, P. Le Ménez, I. Berque, J. Ardisson, A. Pancrazi, *Tetrahedron* 1996, 52, 6613–6634; b) I. Berque, P. Le Ménez, P. Razon, C. Anies, A. Pancrazi, J. Ardisson, A. Neuman, T. Prangé, J.-D. Brion, *Synlett* 1998, 1132–1134; c) I. Berque, P. Le Ménez, P. Razon, A. Pancrazi, J. Ardisson, J.-D. Brion, *Synlett* 1998, 1135–1137; d) I. Berque, P. Le Ménez, P. Razon, J. Mahuteau, J.-P. Férézou, A. Pancrazi, J. Ardisson, J.-D. Brion, *J. Org. Chem.* 1999, 64, 373–381.
- [10] D. Hoppe, T. Hense, D. Hoppe, T. Hense, Angew. Chem. 1997, 109, 2376–2410; Angew. Chem. Int. Ed. Engl. 1997, 36, 2282–2316.
- [11] F.-H. Porée, A. Clavel, J.-F. Betzer, A. Pancrazi, J. Ardisson, *Tetrahedron Lett.* 2003, 44, 7553–7556.
- [12] a) P. Fritsch, *Liebigs Ann. Chem.* 1894, 279, 319–323; b) W. P. Buttenberg, *Liebigs Ann. Chem.* 1894, 279, 324–337; c) H. Wiechell, *Liebigs Ann. Chem.* 1894, 279, 337–344; for reviews see: d) M. Braun, *Angew. Chem.* 1998, 110, 444–465; *Angew. Chem. Int. Ed.* 1998, 37, 430–451; e) R. Knorr, *Chem. Rev.* 2004, 104, 3795–3849.
- [13] a) G. Stork, K. Zhao, *Tetrahedron Lett.* **1989**, *30*, 2173–2174; b) J. Chen, T. Wang, K. Zhao, *Tetrahedron Lett.* **1994**, *35*, 2827–2828.
- [14] a) 41% yield with 85:15 Z/E ratio. A. B. Smith, III, Y. Qiu, D. R. Jones, K. Kobayashi, J. Am. Chem. Soc. 1995, 117, 12011–12012; H. Arimoto, M. D. Kaufman, K. Kobayashi, Y. Qiu, A. B. Smith, III, Synlett 1998, 765–767; b) 46% yield with 90:10 (Z)/(E) ratio. A. B. Smith, III, T. J. Beauchamp, M. J. LaMarche, M. D. Kaufman, Y. Qiu, H. Arimoto, D. R. Jones, K. Kobayashi, J. Am. Chem. Soc. 2000, 122, 8654–8664; c) 40% yield with 70:30 to 90:10 Z/E ratio was obtained with small amounts of reduced product arising from the non-iodinated ylide. J. A. Marshall, B. A. Johns, J. Org. Chem. 1998, 63, 7885–7892.
- [15] T. Fujisawa, Y. Kurita, M. Kawashima, T. Sato, Chem. Lett. 1982, 1641–1642.
- [16] a) P. Kocienski, S. Wadman, K. Cooper, J. Am. Chem. Soc. 1989, 111, 2363–2365; b) P. Kocienski, C. Barber, Pure Appl. Chem. 1990, 62, 1933–1940; c) P. Kocienski in Organic Synthesis via Organometallics (OSM 4, Aachen) (Eds.: E. Enders, H.-J. Gais, W. Keim), Vieweg, Wiesbaden (Germany), 1993, pp. 203–223.
- [17] P. Le Ménez, J.-D. Brion, J.-F. Betzer, A. Pancrazi, J. Ardisson, Synlett 2003, 955–958.
- [18] a) T. Iversen, D. R. Bundle, J. Chem. Soc. Chem. Commun. 1981, 1240–1241; b) N. Nakajima, K. Horita, R. Abe, O. Yonemitsu, Tetrahedron Lett. 1988, 29, 4139–4142; c) L. C. Diaz, G. Z. Melgar, L. S. A. Jardim, Tetrahedron Lett. 2005, 46, 4427–4431.
- [19] P. L. Anelli, C. Biffi, F. Montanari, S. Quici, J. Org. Chem. 1987, 52, 2559–2562.
- [20] D. J. Clarke, R. S. Robinson, Tetrahedron 2002, 58, 2831-2837.
- [21] D. Enders, M. Kroll, G. Raabe, J. Runsink, Angew. Chem. 1998, 110, 1770–1773; Angew. Chem. Int. Ed. 1998, 37, 1673–1675.
- [22] J. Wiemann, L. T. Thuan, Bull. Soc. Chim. Fr. 1958, 199–203.

© 2008 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

www.chemeurj.org

CHEMISTRY

- [24] L. Keller, F. Dumas, J. d'Angelo, Eur. J. Org. Chem. 2003, 2488– 2497.
- [25] J.-P. Ducoux, P. Le Menez, N. Kunesch, E. Wenkert, J. Org. Chem. 1993, 58, 1290–1292.
- [26] E. J. Corey, M. A. Letavic, J. Am. Chem. Soc. 1995, 117, 9616–9617.
 [27] P. Le Ménez, J.-D. Brion, N. Lensen, E. Chelain, A. Pancrazi, J. Ar-
- [27] F. De Mollee, S. D. Dich, R. Lensen, E. Chohan, F. Fulletan, S. Fif disson, Synthesis 2003, 2530–2534.
 [28] A. J. Mancuso, S.-L. Huang, D. Swern, J. Org. Chem. 1978, 43,
- 2480–2482.
- [29] J. R. Parikh, W. v. E. Doering, J. Am. Chem. Soc. 1967, 89, 5505– 5507.
- [30] E. J. Corey, J. W. Suggs, Tetrahedron Lett. 1975, 16, 2647-2650.
- [31] S. V. Ley, J. Norman, W. P. Griffith, S. P. Marsden, *Synthesis* **1994**, 639–666.
- [32] a) M. Frigerio, M. Santagostino, *Tetrahedron Lett.* 1994, 35, 8019–8022; b) M. Frigerio, M. Santagostino, S. Sputore, G. Palmisano, *J. Org. Chem.* 1995, 60, 7272–7276.
- [33] K. Narasaka, A. Morikawa, K. Saigo, T. Mukaiyama, Bull. Chem. Soc. Jpn. 1977, 50, 2773–2776.
- [34] J. A. Marshall, W. Y. Gung, Tetrahedron 1989, 45, 1043-1052.
- [35] A. DeMico, R. Margarita, L. Parlanti, A. Vescovi, G. Piancatelli, J. Org. Chem. 1997, 62, 6974–6977.
- [36] J. Chan, T. F. Jamison, J. Am. Chem. Soc. 2004, 126, 10682-10691.
- [37] a) B. M. Trost, J. L. Belletire, S. Godleski, P. G. McDougal, J. M. Balkovec, J. J. Baldwin, M. E. Christy, G. S. Ponticello, S. L. Varga, J. P. Springer, J. Org. Chem. **1986**, *51*, 2370–2374; b) J. M. Seco, E. Quiñoá, R. Riguera, Chem. Rev. **2004**, *104*, 17–117.
- [38] a) G. E. Keck, D. E. Abbott, *Tetrahedron Lett.* 1984, 25, 1883–1886;
 b) Y. Yamamoto, H. Yatagai, Y. Ishihara, N. Maeda, K. Maruyama, *Tetrahedron* 1984, 40, 2239–2246;
 c) G. E. Keck, K. A. Savin, E. N. K. Cressman, D. E. Abbott, *J. Org. Chem.* 1994, 59, 7889–7896.
- [39] a) H. C. Brown, K. S. Bhat, J. Am. Chem. Soc. 1986, 108, 5919– 5923; b) H. C. Brown, K. S. Bhat, R. S. Randad, J. Org. Chem. 1989, 54, 1570–1576.
- [40] J. Mulzer, H. M. Kirstein, J. Buschmann, C. Lehmann, P. Luger, J. Am. Chem. Soc. 1991, 113, 910–923.
- [41] S. D. Rychnovsky, D. J. Skalitzky, Tetrahedron Lett. 1990, 31, 945– 948.
- [42] H. Normant, Bull. Soc. Chim. Fr. 1957, 728-733.

- [43] T. H. Black, T. S. McDermott, G. A. Brown, *Tetrahedron Lett.* 1991, 32, 6501–6502.
- [44] a) H. Neumann, D. Seebach, *Tetrahedron Lett.* 1976, *17*, 4839–4842;
 b) W. F. Bailey, J. J. Patricia, *J. Organomet. Chem.* 1988, *352*, 1–46;
 c) M. E. Jung, J. A. Hagenah, *J. Org. Chem.* 1987, *52*, 1889–1902.
- [45] a) B. Kang, S. Chang, *Tetrahedron* 2004, 60, 7353–7359; b) H. Takahata, Y. Banba, H. Ouchi, H. Nemoto, *Org. Lett.* 2003, 5, 2527– 2529.
- [46] S.-I. Murahashi, J. Organomet. Chem. 2002, 653, 27-33.
- [47] a) P. J. Garegg, B. Samuelsson, J. Chem. Soc. Perkin Trans. 1 1980, 2866–2869; b) E. J. Corey, S. G. Pyne, W.-G. Su, Tetrahedron Lett. 1983, 24, 4883–4886.
- [48] K. B. Sharpless, R. C. Michaelson, J. Am. Chem. Soc. 1973, 95, 6136–6137.
- [49] A. H. Hoveyda, D. A. Evans, G. C. Fu, Chem. Rev. 1993, 93, 1307– 1370.
- [50] B. H. Lipshutz, J. Kozlowski, R. S. Wilhelm, J. Am. Chem. Soc. 1982, 104, 2305–2307.
- [51] J. Cossy, S. BouzBouz, A. H. Hoveyda, J. Organomet. Chem. 2001, 634, 216–221; T. M. Trnka, R. H. Grubbs, Acc. Chem. Res. 2001, 34, 18–29. S.J. Connon, S. Blechert, Angew. Chem. 2003, 115, 1944– 1968; Angew. Chem. Int. Ed. 2003, 42, 1900–1923.
- [52] a) P. Kocienski, N. J. Dixon, Synlett 1989, 52–54; b) R. W. Hoffmann, V. Giesen, M. Fuest, Liebigs Ann. Chem. 1993, 629–639.
- [53] D. A. Evans, J. A. Gauchet-Prunet, J. Org. Chem. 1993, 58, 2446– 2453.
- [54] D. A. Evans, B. W. Trotter, P. J. Coleman, B. Côté, L. C. Dias, H. A. Rajapakse, A. N. Tyler, *Tetrahedron* **1999**, 55, 8671–8726.
- [55] a) A. Mengel, O. Reiser, *Chem. Rev.* **1999**, *99*, 1191–1223; b) M. T. Reetz, K. Kesseler, A. Jung, *Tetrahedron Lett.* **1984**, *25*, 729–732; c) D. A. Evans, J. L. Duffy, M. J. Dart, *Tetrahedron Lett.* **1994**, *35*, 8537–8540.
- [56] R. G. Jones, H. Gilman, Org. React. 1951, 6, 339-366.
- [57] a) E. J. Corey, R. K. Bakshi, S. Shibata, J. Am. Chem. Soc. 1987, 109, 5551–5553; b) E. J. Corey, C. J. Helal, Angew. Chem. 1998, 110, 2092–2118; Angew. Chem. Int. Ed. 1998, 37, 1986–2012; c) B. M. Trost, J. L. Gunzner, O. Dirat, Y. H. Rhee, J. Am. Chem. Soc. 2002, 124, 10396–10415.
- [58] J. A. Marshall, G. M. Schaaf, J. Org. Chem. 2003, 68, 7428-7432.
- [59] P. Kocovský, Tetrahedron Lett. 1986, 27, 5521–5524.

Received: July 21, 2008 Published online: October 30, 2008